





### JEE MAIN SAMPLE PAPER-1

CLASS:12<sup>th</sup> (PCM)

Maximum Marks: 360

#### **Topics Covered:**

Physics : Full Syllabus

Chemistry : Full Syllabus

Mathematics : Full Syllabus

#### **Important Instruction:**

- 1. Attempting all the questions are compulsory.
- 2. Use Blue / Black Ball point pen only.
- 3. There are three sections of equal weightage in the question paper A, B, C (**Physics, Chemistry and Mathematics**) Having 30 questions each.
- 4. For marking scheme, +4 marks for each correct answer and -1 marks for each incorrect answer.
- 5. Use of calculator and other electronic devices is not allowed during the exam.
- 6. No extra sheets will be provided for any kind of work.

Father's Name: Signature:   Branch Name: Contact No:	Name of the Student:	Class:
Branch Name: Contact No:		
	Branch Name:	Contact No:

## <u> PART – A</u> PHYSICS

1. If pressure P, velocity V and time T are taken as fundamental physical quantities, then the dimensional formula for force is (b)  $P^{-1}V^2T^{-2}$ (c)  $PVT^2$ (d)  $P^{-1}VT^2$ 

(a)  $PV^2T^2$ 

(a) $\frac{n}{2\pi}$ 

2. A projectile is given an initial velocity of  $(\hat{i}+2\hat{j})m/s$ , where  $\hat{i}$  is along the ground and  $\hat{j}$  is along the vertical. If  $g = 10m/s^2$ , the equation of its trajectory is: (d)  $4y = 2x - 25x^2$ (a)  $y = x - 5x^2$ (c)  $4y = 2x - 5x^2$ (b)  $y = 2x - 5x^2$ 

3. A boy runs along a straight path for the first half of the distance with a velocity  $v_1$  and the second half of the distance with a velocity  $v_2$ . The average velocity V is given by

(a)  $\frac{2}{V} = \frac{1}{v_1} + \frac{1}{v_2}$ (b)  $V = \frac{v_1 + v_2}{2}$ (c)  $V = \sqrt{v_1 v_2}$ (d)  $\vec{V} = \vec{v}_1 + \vec{v}_2$ 

4. Two blocks of equal mass are connected by a light string and placed on a smooth horizontal surface. If a force F acts on one of the block then the tension in the string is

(b)  $\frac{r}{2}$ (a) F (d) –*F* (c) 2F

5. A cricket player catches a ball of mass 0.1 kg, moving with a speed of  $10ms^{-1}$  in 0.1s. Magnitude of the force exerted by him is (N) (c) 2 (d) 1 (a) 10 (b) 4

6. A spring for spring constant  $240 Nm^{-1}$  is compressed by 10 cm whereas another similar spring is extended by 10 cm. The difference of the stored potential energies of two springs is (a) zero (b) 4 J (c) 1.2 J (d) 12 J

(b)  $\frac{n^2}{2\pi}$ 

7. For an electron in the *nth* Bohr orbit of hydrogen atom, what will be the ratio of radius of orbit to its de-Broglie wavelength

8. If in a plano-convex lens radius of curvature of convex surface is 10 cm and the focal length of the lens is 30 cm, the refractive index of the material of the lens will be: (b) 1.66 (a) 1.5 (c)1.33 (d) 3

**Rough Space** 

(c) $\frac{1}{2\pi n}$  (d) $\frac{1}{2\pi n^2}$ 

9. A body is projected at an angle  $\theta$  to the horizontal with kinetic energy  $E_k$ . The potential energy of the body at the highest point of the trajectory is

(a)  $E_k$  (b)  $E_k \cos^2 \theta$  (c)  $E_k \sin^2 \theta$  (d)  $E_k \tan^2 \theta$ 

10. One gram of ice is mixed with one gram of steam. After thermal equilibrium, the temperature of themixture is(a)  $0^{\circ}C$ (b)  $100^{\circ}C$ (c)  $55^{\circ}C$ (d)  $80^{\circ}C$ 

(d) 75°

11. A ray of light travelling in the direction  $\frac{1}{2}(\hat{i}+\sqrt{3}\hat{j})$  is incident on a plane mirror in the x-z plane . After reflection, it travels along the direction  $\frac{1}{2}(\hat{i}-\sqrt{3}\hat{j})$ . The angle of incidence is

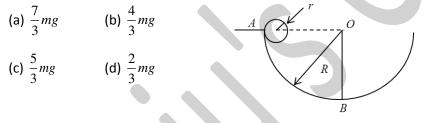
(c) 60°

(a) 30°

12. A radioactive nucleus (initial mass number A and atomic number Z) emits  $3\alpha$  – particles and 2 positrons. The ratio of number of neutrons to that of protons in the final nucleus will be

(a) $\frac{A-Z-8}{2}$	(b) $\frac{A-Z-4}{2}$	(c) $\frac{A - Z - 12}{2}$	(d) $\frac{A-Z-4}{Z-4}$
Z-4	Z-8	Z-4	Z-2

13. A cylinder of mass m and radius r rolls down a circular track from point A as shown in the figure. Assume that the friction is just sufficient to support the rolling. Velocity of the cylinder at point A was zero. Assume  $r \ll R$ . The reaction by the track on the cylinder at point B is



(b) 45°

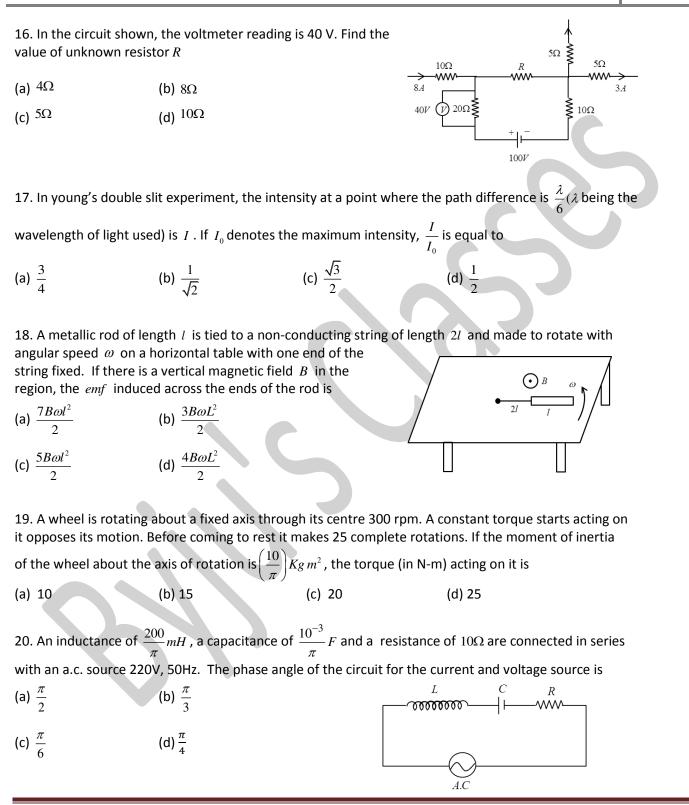
14. A charged particle of a mass m and charge q is released from rest in a uniform electric field E neglecting the effect of gravity, the kinetic energy of the charged particle after t second is

(a)  $\frac{eqm}{t}$  (b)  $\frac{E^2 q^2 t^2}{2m}$  (c)  $\frac{2E^2 t^2}{mg}$  (d)  $\frac{Eq^2 m}{2t^2}$ 

15. A police car moving at 22 m/s, chases a motorcyclist. The police man sounds his horn at 176 Hz, while both of them move towards a stationary siren of frequency 165 Hz. Calculate the speed of the motorcycle, if it is given that motor cyclist does not observes any beats. (velocity of sound in air = 330 m/s) Police Car Motorcycle









- 21. If an electron and a proton having same momenta enter perpendicular to a magnetic field, then
- (a) the length of curved path of electron and proton will be same
- (b) they will move undeflected
- (c) the length of curved path of electron is more curved than that of the proton
- (d) the length of curved path of proton is more curved than that of the electron

22. A heavy uniform chain lies on a horizontal table top. If the coefficient of friction between the chain and the table surface is 0.25, then the maximum fraction of the length of the chain that can hang over one edge of the table is (c) 35% (d)15%

(a) 20% (b) 25%

23. A cylindrical tube, open at both ends, has a fundamental frequency  $f_0$ , in air. The tube is dipped vertically into water such that half of its length is inside water. The fundamental frequency of the air column now is

(a) 
$$\frac{3f_0}{4}$$
 (b)  $f_0$  (c)  $\frac{f_0}{2}$  (d)  $3f_0$ 

24. An infinite long straight wire is bent into a semicircle of radius R, as shown in the figure. A current I is sent through the conductor. The magnetic field at the centre of the semicircle is: (a) infinite (b) zero

(c)  $\frac{\mu_0 \pi I}{4\pi R}$ (d)  $\frac{\mu_0}{4\pi} \frac{I}{R} (\pi + 1)$ 

25. Choose the correct relation regarding potential of an electric dipole. Here A, B, C and D all are at equal distance from point O which is much larger than the dimensions of •C the dipole. Then

(a) $ V_A  =  V_B  >  V_C  =  V_D $	(b) $ V_C  =  V_D  >  V_A  =  V_B $	A
(c) $ V_A  >  V_C  =  V_D  >  V_B $	(d) $ V_C  >  V_B  =  V_D  >  V_A $	Aq O +q B
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26. In an A.C. circuit, V and I are given by  $V = 100\sin(100t)volt$ ,  $I = 100\sin\left(100t + \frac{\pi}{3}\right)A$  Then the power dissipated in the circuit is

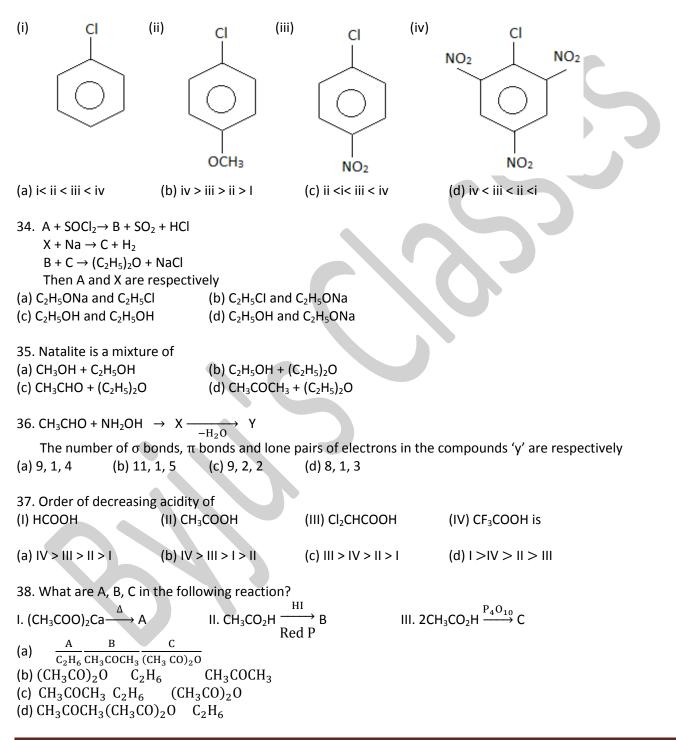
(a)  $10^4 W$ (c) 2500W (d) 5W (b) 10W

27. The magnetic flux  $\phi$  (in weber) linked with a coil of resistance 10 $\Omega$  varies with time t (in second) as  $\phi = 8t^2 - 4t + 1$ . The current induced in the coil at t = 0.1 sec is (a) 10A (b) 0.24A (c) 0.12A (d) 4.8A

28. If R, C and L denote resistance, capacitance and inductance. Which of the following will NOT have the dimensions of frequency? (a)  $RL^{-1}$  (b)  $R^{1}C^{1}$  (c) $L^{-1/2}C^{-1/2}$  (d) RCL

29. The following figure shows a logic gate circuit with two inputs A and B and the output C. The voltage wavefront of A, B and C are as shown below A٠ Logic gate circuit The logic circuit gate is (a) NAND gate (b) NOR gate (c) OR gate (d) AND gate 30. If  $N_0$  is the original mass of the substance of half-life period  $t_{1/2} = 5$  years, then the amount of substance left after 15 years is (b)  $\frac{N_0}{16}$ (a)  $\frac{N_0}{8}$ PART-B CHEMISTRY 31. The IUPAC name of the following compound is: Cl  $CH_3 - C - CH_2 - CH = CH - CH_3$ Н (a) 2-chlorohex-5-ene (b) 5-chlorohex-2-ene (c) 1-chloro-1-methylpent-3-ene (d) 5-chloro-5-methylpent-2-ene 32. Formation of polyethylene from calcium carbide takes place as follows:  $CaC_2 + 2H_2O \rightarrow Ca(OH)_2 + C_2H_2$  $C_2H_2 + H_2 \rightarrow C_2H_4$  $nC_2H_4 \rightarrow (-CH_2 - CH_2 -)_n$ the amount of polyethylene obtained from 64 kg of  $CaC_2$  is (a) 7 kg (b) 14 kg (c) 21 kg (d) 28 kg

33. Arrange the following compounds in order of increasing reactivity towards nucleophilic substitution.



39. The positive carbylamines test is given by (i) N, N-dimethyl aniline (ii) 2, 4-dimethyl aniline (iii) N-methyl-O-methyl aniline (iv) p-methyl benzylamine (a) Only (i) (b) (ii) and (iv) (c) (iii) and (iv) (d) (i) and (iv) 40. The rate constant for the reaction,  $2N_2O_5 \rightarrow 4NO_2 + O_2$  is  $3.0 \times 10^{-5} s^{-1}$ . If the rate is  $2.40 \times 10^{-5} mol/L. sec^{-1}$ , then the initial concentration of  $N_2O_5$  (in mol/L) is (a) 1.4 (b) 1.2 (c) 0.04 (d) 0.8 41. Sodium has a bcc structure with nearest neighbour distance of 365.9 pm. Calculate its density (atomic mass of sodium = 23) (a) 1.51 g/ $cm^3$ (b) 2.65 g/ $cm^3$ (c) 3.3 g/ $cm^3$ (d) 6.2 g/ $cm^3$ 42. The latent heat of Vaporisation of a liquid at 500k and 1atm pressure is 30 Kcal/mole. What will be the change in internal energy of 3 mol of liquid at same temperature? (d) -27 Kcal (a)13 Kcal (b) -13 Kcal (c) 27Kcal 43. For the reaction  $3Br_2 + 60H^{\ominus} \rightarrow 5Br^{\ominus} + BrO_3^{\ominus} + 3H_2O$ Equivalent weight of Br<sub>2</sub>(Molecular weight M)is (d)  $\frac{M}{6}$ (a)  $\frac{M}{2}$ (b)  $\frac{M}{10}$  $(C)\left(\frac{M}{2}+\frac{M}{10}\right)$ 44. The specific conductance of saturated solution of Agcl is found to be 1.86×10<sup>-6</sup> ohm<sup>-1</sup> cm<sup>-1</sup> and that of water is 6×10<sup>-8</sup> ohm<sup>-1</sup> cm<sup>-1</sup>. The Solubility of Agcl is Given  $n_{Agcl}^{0} = 137.2 \text{ ohm}^{-1} \text{cm}^{2} \text{eq}^{-1}$ (a)1.7×10<sup>-3</sup>M (b)1.3×10⁵M  $(c)1.3 \times 10^{-4} M$ (d)1.3×10<sup>-6</sup>M 45. The vapour density of the equilibrium mixture of the reaction  $SO_2Cl_{2(g)} \Rightarrow SO_{2(g)} + Cl_{2(g)}$  is 50. The percentage dissociation of SO<sub>2</sub>Cl<sub>2</sub> is (c) 30 (a) 33 (b) 35 (d) 66 46. The ionisation energy of hydrogen atom is 13.6 ev. What will be the ionisation energy of Li<sup>+2</sup> ions? (d)48.6 ev (a)13.6 ev (b) 54.4 ev (c) 122.4 ev 47. The first and second dissociation constants of an acid,  $H_2A$  are  $1.0 \times 10^{-5}$  and  $5.0 \times 10^{-10}$  respectively. The Overall dissociation constant of the acid will (a)  $0.2 \times 10^5$ (b) 5.0×10<sup>-5</sup> (c)  $5.0 \times 10^{-15}$ (d)  $5.0 \times 10^{15}$ 48. The density of Oxygen is 1.43 g/L at STP. The density of oxygen at  $17^{\circ}$  c and 800 torr is (a) 1.4 (b) 1.69 (c) 1.8 (d) 2.5

49. Match the List-I with List-II by using the postulates of VBT of complexes

List-IList-II $(P)[Ni(CN)_4]^{2-}$  $(1) sp^3$  hybridization $(Q)[Ni(CO)_4]$  $(2) dsp^2$  hybridization $(R) [Cu(NH_3)_4]^{2+}$  $(3) \mu = 0 BM$  $(S) [Pd(CI)_4]^{2-}$  $(4) \mu = 1.732 BM$ (a) P-3,1 Q-1,4 R-2,3 S-1,4(b) P-2,3 Q-1,3 R-1,4 S-2,3(c)p-2,4 Q-1,4 R-2,4 S-2(d) P-2,3 Q-1,3 R-2,4 S-2,3

50. The incorrect order regarding 15 th group hydrides is

(a) Reducing nature  $: NH_3 < PH_3 < AsH_3 < SbH_3 < BiH_3$ 

(b) Bond angle  $:NH_3 > PH_3 > AsH_3 > SbH_3 > BiH_3$ 

(c) Basic nature  $: NH_3 > PH_3 > AsH_3 > SbH_3 > BiH_3$ 

(d) Boiling point  $: NH_3 < PH_3 < AsH_3 < SbH_3 < BiH_3$ 

51. Which of the among the following is more acidic ? (a)  $H_2O$  (b)  $H_2S$  (c) $H_2Se$  (d)  $H_2Te$ 

52. The true statement for the acids of phosphorus :  $H_3PO_2$ ,  $H_3PO_3$  and  $H_3PO_4$  is

(a) The order of acidity is  $H_3PO_2 < H_3PO_3 < H_3PO_4$ 

(b) All of these are reducing in nature

(c) All are tribasic acids

(d) The geometry of phosphorus is tetrahedral in all the three

53. The salt used for performing 'bead' test in qualitative inorganic analysis is : (a)  $K_2SO_4$ .  $AI_2(SO_4)_3$ .  $24H_2O$ (b)  $FeSO_4$ .  $(NH_4)_2SO_4$ .  $6H_2O$ (c)  $Na (NH_4)HPO_4$ .  $4H_2O$ (d)  $Na_2SO_4$ .  $Cr_2(SO_4)_3$ .  $24H_2O$ 

54. Which are true statements among the following ? (A)  $PH_5$  and  $BiCl_5$  do not exist (B) Bond order of  $CO_3^{-2}$  is 1.33 (C)  $SeF_4$  and  $CH_4$  have same shape (D)  $I_3^+$  has bent geometry (E) The number of  $P_{\pi} - d_{\pi}$  bonds in  $SO_2 \& SO_3$  are same (a) A&C (b) A,B&E (c) A,B&D (d) all the above

55. The increasing order of acidic nature of the following oxidies is(a)  $SiO_2 < P_2O_5 < CI_2O_7 < SO_3$ (b)  $SiO_2 < P_2O_5 < SO_3 < CI_2O_7$ (c)  $CI_2O_7 < SO_3 < P_2O_5 < SiO_2$ (d)  $SO_3 < CI_2O_7 < SiO_2 < P_2O_5$ 

56. An orange coloured solution acidified with  $H_2SO_4$  and treated with a substance'X' gives a blue coloured

solutio (a) H <sub>2</sub> C	n of CrO <sub>5</sub> .The subs )    (b) dil HCl		H <sub>2</sub> O <sub>2</sub>	(d) Conc. HCl	
57. In E	. Coli DNA, AT/GC ra	atio of is 0.93. If	the number of mo	oles of adenine in the D	NA sample is 465,000,
then th (a) 465	e moles of guanine ,000 (b)	present is ) 535,000	(c) 50,000	(d) 93,000	
58. An (a) Ethy	example of a conde /lene (b)	nsation homopo ) Melamine-form	•	(c) Alkyd resin	(d) Perlon or Nylon-6
59.					
	Colum	nn I	Colu	mn II	
	(A) Antacid (B) Antibiotic (C) Traquillize (D) Antioxidan	r	(p) Equanil (q) BHT (r) Bromphenira (s) Aminoglycosi		
	; B-q; C-s; D-r	(b) A-q; B-p			
(c) A-r;	B-s; C-p; D-q	(d) A-r; B-s	; C-q; D-p		
60. The (a) O <sub>2</sub> a	e smog is essentially and $O_3$ (b)	caused by prese ) $O_3$ and $N_2$		sulphur and nitrogen	(d) $O_2$ and $N_2$
		Ň	<u>PART</u> MATHEN		
	the areas of the sq			he sides PQ and QR res	pectively, then the
	ne tangent at the po n the y-axis, then ler		$e x^2 + y^2 + 6x + 6y$	= 2 meets the line 5x –	2y + 6 = 0 at a point
(a) 4	(b) 5√5	(c) 5	(d) 4√5		
63. The (a) $\frac{\pi}{2}$	e angle between the (b) $\frac{\pi}{3}$	tangents drawn (c) $\frac{\pi}{4}$	from the point (1 (d) $\frac{\pi}{6}$	, 4) to the parabola y <sup>2</sup> =	= 4x is

64. Mean of 100 items is 49. It was discovered that three items which should have been 60, 70, 80 were

wrongly read as 40, 20, 50 respectively. Correct mean will be (b)  $82\frac{1}{2}$ (c) 50 (a) 48 (d) 80 65. Angle between asymptotes of the hyperbola  $3x^2 - y^2 = 3$  is (b)  $\frac{2\pi}{3}$ (a)  $\frac{\pi}{3}$  $(c)\frac{\pi}{c}$ (d)  $\frac{3\pi}{4}$ 66. P  $\rightarrow$  (q V r) is false, then the true valves of p, q , r respectively are (a) T, T, F (b) T, F, T (c) F, T, T (d) F, F, T 67. Lt  $\frac{1-\cos^3 x}{\sin 3x \sin 5x} =$ (a)  $\frac{1}{15}$  (b)  $\frac{2}{15}$  $(c)\frac{1}{30}$ (d)  $\frac{1}{10}$ 68.  $\sin^{-1}x + \sin^{-1}y = \pi/2$  then  $\frac{dy}{dx} =$ (a) - x/y(b) - y/x(c) x/y (d) y/x69. The tangent at A (2, 4) on  $y = x^3 - 2x^2 + 4$  cuts the x axis at T then AT = (b)  $\sqrt{17}/4$ (a)  $4\sqrt{17}$ (c)  $\sqrt{17}$ (d) 17 70. The value of 'a' for which the function  $f(x) = a \sin x + \frac{1}{3} \sin 3x$  has an extremum at  $x = \pi/3$  is (a) 2 (b) -2 (c) 2/3 (d) - 2/371. The quadratic equation  $3ax^2 + 2bx + c = 0$  has at least one root between 0 and 1 if (b) a + b + c = 0(a) a + b + c = 1(c) 3a + 2ab + c = 0(d) 6a + 2b = 072.  $\int e^{x} \left(\frac{x^{2}+5x+7}{(x+3)^{2}}\right) dx = e^{x} f(x) + c$  then f(x) = $(c) - \left(\frac{x+2}{x+3}\right)$ (d)  $\frac{x+2}{x+3}$ (a)  $\frac{1}{x+3}$  $\pi/2$  $\left(\frac{2-\sin x}{2+\sin x}\right) dx =$ 73. ∫ In  $-\pi/2$ (b)  $-\frac{\pi}{2}\log 2$ (c)  $\frac{\pi}{2} \log 2$ (a) – πlog 2 (d) 0

74. The area enclosed by the curves $y =  \sin x $ , x axis and $ x  = \pi$ is (in sq units)				
(a) 4	(b) 2	(c) 8	(d) 3/2	
75. xdy – ydx = $\sqrt{x^2}$ –	$y^2$ dx and y(1) = 0 then y	$y(e^{\pi/2}) =$		
(a) π/2	(b) $e^{\pi/2}$	(c) 1	(d) e	
76. If the angle betwee its coordinates, then $\lambda$		$\lambda \hat{\imath} + \lambda \hat{\jmath} - k$ , is acute and $\bar{k}$	$\overline{\mathfrak{o}}$ makes obtuse angle with the axes of	
(a) (−∞, 0)		$-\infty, \frac{1}{2} \cup (1, \infty)$	(d) (1, ∞)	
77. Let $\bar{V} = 2i + j - k$ , $\bar{V}$ (a) $\sqrt{33}$ (b) 7	$\overline{V}$ = i + 3k, if $\overline{U}$ is a unit v (c) $\sqrt{5}$	ector then the maximum $\overline{59}$ (d) not defined		
78. The distance of the	e plane passing through	(1, 1, 1) and perpendicula	ar to the line $\frac{x-1}{3} = \frac{y-1}{0} = \frac{z-1}{4}$	
from the origin is (a) $\frac{3}{4}$	(b) $\frac{4}{5}$	$(c)\frac{7}{5}$	(d) 0	
79. The point in which the join of A(-9, 4, 5) and B(11, 0, -1) is met by the perpendicular from the origin				
is (a) (2, 2, 1)	(b) (2, 1, 2)	(c) (1, 2, 2)	(d) (2, 2, 2)	
80. The value of 'm' for which the straight line $3x - 2y + z + 3 = 0 = 4x - 3y + 4z + 1$ is parallel to the plane $2x - y + mz - 2 = 0$ is				
(a) – 2	(b) 8	(c) 4	(d) 2	
81. If 5 different things are placed at random in 3 different boxes, then the probability of placing them such that no box remains empty is				
(a) $\frac{30}{81}$	(b) $\frac{20}{81}$	$(c)\frac{50}{81}$	(d) $\frac{40}{81}$	
82. Two person A and then the probability of		nother. The person who t	hrows 6 wins. If A starts the game,	
(a) $\frac{4}{11}$	(b) $\frac{3}{11}$	(c) $\frac{5}{11}$	(d) $\frac{6}{11}$	
83. The Expansion [ $x^2$ (a) 8 (b) 10	+ $(x^6 - 1)^{1/2}$ ] <sup>5</sup> + $[x^2$ (c) 13	$-(x^6-1)^{1/2}$ is a point (d) 14	olynomial of degree.	

84. If $ z - \frac{4}{2}  = 2$ ,	then the maxim	tumvalue of $ z $ is c) $\sqrt{5} - 1$		
a) $\sqrt{3} + 1$	b) $\sqrt{5} + 1$	c) $\sqrt{5} - 1$	d) <del>√3</del> -1	
85. For positive	numbers x, y an	d z the numerica	al value of the determin	ant $\begin{vmatrix} 1 & \log_x y & \log_x z \\ \log_y x & 1 & \log_y z \\ \log_z x & \log_z y & 1 \end{vmatrix}$ is
(a) 0	(b) 1	(c) $\log_e xyz$	(d) -log xyz	
-		by 30 is to be fo it can be done is		1,2,3,4,5 with our repetition of (d)60
87. The Equatio (a) no root	on x - $\frac{2}{x-1} = 1 - \frac{1}{x-1}$ (b) one	$\frac{2}{x-1}$ has root	(c) two roots	(d) infinitely many
88. tan9° – tan2 (a) 1	27° – tan63° + ta (b) 2	n81º = (c) 3	(d) 4	
89. Fractional p (a) $\frac{2}{31}$		$(c)\frac{8}{31}$	(d) $\frac{10}{31}$	
90. If $ z_1 + z_2  = a$ ) 0	$ z_1 - z_2 $ then a b) $\pm \frac{\pi}{4}$	$rg z_1-arg z_2= c)\pm \frac{\pi}{2}$	d)π	