# Answers & Solutions For JEE MAIN 2017

Time Duration: 3 hrs. **(Code-B)** Maximun Marks : 360

## (Physics, Mathematics and Chemistry)

**Important Instructions :**

1. The test is of 3 hours duration.
2. The Test Booklet consists of 90 questions. The maximum marks are 360.
3. There are three parts in the question paper A, B, C consisting of Physics, Mathematics and Chemistry having 30 questions in each part of equal weightage. Each question is allotted 4 (four) marks for each correct response.
4. Candidates will be awarded marks as stated above in Instructions No. 3 for correct response of each question. ¼ (one-fourth) marks of the total marks allotted to the question (i.e. 1 mark) will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
5. There is only one correct response for each question. Filling up more than one response in any question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 4 above.
6. For writing particulars/marking responses on Side-1 and Side-2 of the Answer Sheet use only Black BallPoint Pen provided in the examination hall.
7. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. except the Admit Card inside the examination hall/room.

# PART–A : PHYSICS

1. A particle is executing simple harmonic motion with a time period $T$. At time $t$ = 0, it is at its position of equilibrium. The kinetic energy-time graph of the particle will look like :

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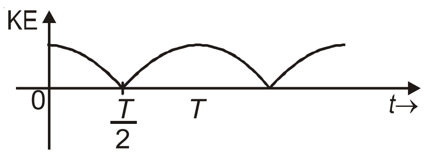


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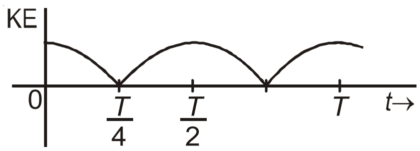


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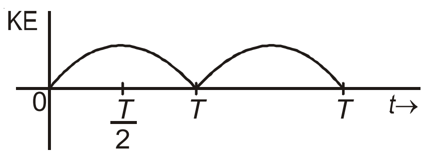


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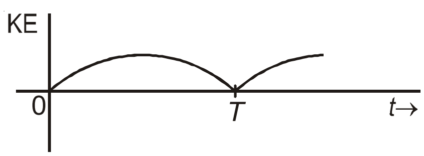
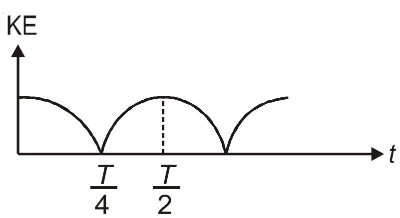



**Answer (2)**

**Sol.** $K.E = \frac{1}{2} m\omega^2 A^2 \cos^2 \omega t$




2. The temperature of an open room of volume 30 m$^3$ increases from 17°C to 27°C due to the sunshine. The atmospheric pressure in the room remains 1 × 10$^5$ Pa. If $n_i$ and $n_f$ are the number of molecules in the room before and after heating, then $n_f - n_i$ will be

(1) 2.5 × 10$^{25}$

(2) –2.5 × 10$^{25}$

(3) –1.61 × 10$^{23}$

(4) 1.38 × 10$^{23}$

**Answer (2)**

**Sol.** $n_1$ = initial number of moles;

$$n_1 = \frac{P_1V_1}{RT_1} = \frac{10^5 \times 30}{8.3 \times 290} \approx 1.24 \times 10^3$$

$n_2$ = final number of moles

$$= \frac{P_2V_2}{RT_2} = \frac{10^5 \times 30}{8.3 \times 300} \approx 1.20 \times 10^3$$

Change of number of molecules

$n_f - n_i = (n_2 - n_1) \times 6.023 \times 10^{23}$

$\approx -2.5 \times 10^{25}$

3. Which of the following statements is false?

(1) A rheostat can be used as a potential divider

(2) Kirchhoff's second law represents energy conservation

(3) Wheatstone bridge is the most sensitive when all the four resistances are of the same order of magnitude

(4) In a balanced Wheatstone bridge if the cell and the galvanometer are exchanged, the null point is disturbed

**Answer (4)**

**Sol.** In a balanced Wheatstone bridge, the null point remains unchanged even if cell and galvanometer are interchanged.

4. The following observations were taken for determining surface tension $T$ of water by capillary method:

diameter of capillary, $D$ = 1.25 × 10$^{-2}$ m

rise of water, $h$ = 1.45 × 10$^{-2}$ m.

Using $g$ = 9.80 m/s$^2$ and the simplified relation $T = \frac{rhg}{2} \times 10^3 \text{N/m}$, the possible error in surface tension is closest to

(1) 2.4%

(2) 10%

(3) 0.15%

(4) 1.5%

**Answer (4)**

**Sol.** $$\frac{\Delta T}{T} \times 100 = \frac{\Delta D}{D} \times 100 + \frac{\Delta h}{h} \times 100$$

$$= \frac{0.01}{1.25} \times 100 + \frac{0.01}{1.45} \times 100$$

$= \dfrac{100}{125} + \dfrac{100}{145}$

$= 0.8 + 0.689$

$= 1.489$

$\simeq 1.5\%$

5. In amplitude modulation, sinusoidal carrier frequency used is denoted by $\omega_c$ and the signal frequency is denoted by $\omega_m$. The bandwidth $(\Delta\omega_m)$ of the signal is such that $\Delta\omega_m << \omega_c$. Which of the following frequencies is not contained in the modulated wave?

(1) $\omega_m + \omega_c$ (2) $\omega_c - \omega_m$

(3) $\omega_m$ (4) $\omega_c$

**Answer (3)**

**Sol.** Modulated wave has frequency range.

$\omega_c \pm \omega_m$

$\therefore$ Since $\omega_c >> \omega_m$

$\therefore$ $\omega_m$ is excluded.

6. A diverging lens with magnitude of focal length 25 cm is placed at a distance of 15 cm from a converging lens of magnitude of focal length 20 cm. A beam of parallel light falls on the diverging lens. The final image formed is

(1) Real and at a distance of 40 cm from the divergent lens

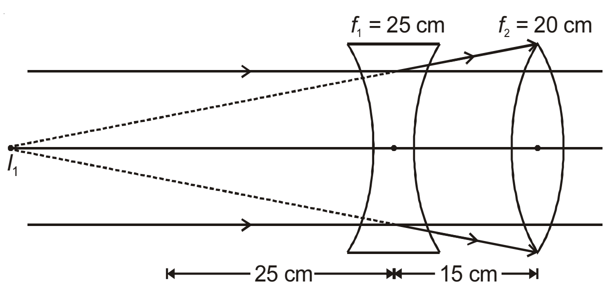
(2) Real and at a distance of 6 cm from the convergent lens

(3) Real and at a distance of 40 cm from convergent lens

(4) Virtual and at a distance of 40 cm from convergent lens

**Answer (3)**

**Sol.**




For converging lens

$u = -40$ cm which is equal to $2f$

$\therefore$ Image will be real and at a distance of 40 cm from convergent lens.

7. The moment of inertia of a uniform cylinder of length $\ell$ and radius $R$ about its perpendicular bisector is $I$. What is the ratio $\dfrac{\ell}{R}$ such that the moment of inertia is minimum?

(1) 1

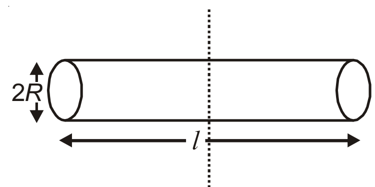
(2) $\dfrac{3}{\sqrt{2}}$

(3) $\sqrt{\dfrac{3}{2}}$

(4) $\dfrac{\sqrt{3}}{2}$

**Answer (3)**

**Sol.**




$$I = \frac{mR^2}{4} + \frac{m\ell^2}{12}$$

$$I = \frac{m}{4}\left[R^2 + \frac{\ell^2}{3}\right]$$

$$= \frac{m}{4}\left[\frac{V}{\pi\ell} + \frac{\ell^2}{3}\right]$$

$$\frac{dI}{d\ell} = \frac{m}{4}\left[\frac{-V}{\pi\ell^2} + \frac{2\ell}{3}\right] = 0$$

$$\frac{V}{\pi\ell^2} = \frac{2\ell}{3}$$

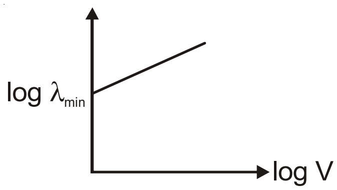
$$V = \frac{2\pi\ell^3}{3}$$

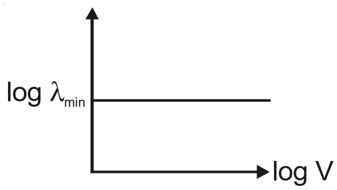
$$\pi R^2\ell = \frac{2\pi\ell^3}{3}$$

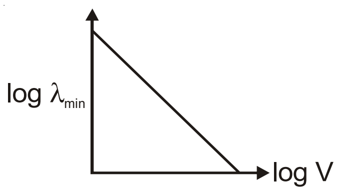
$$\frac{\ell^2}{R^2} = \frac{3}{2}$$

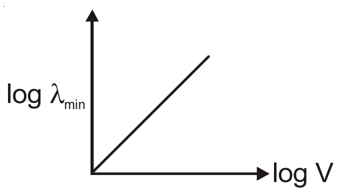
$$\frac{\ell}{R} = \sqrt{\frac{3}{2}}$$

8. An electron beam is accelerated by a potential difference $V$ to hit a metallic target to produce X-rays. It produces continuous as well as characteristic X-rays. If $\lambda_{min}$ is the smallest possible wavelength of X-ray in the spectrum, the variation of $\log \lambda_{min}$ with $\log V$ is correctly represented in

(1) 


(2) 


(3) 


(4) 


**Answer (3)**

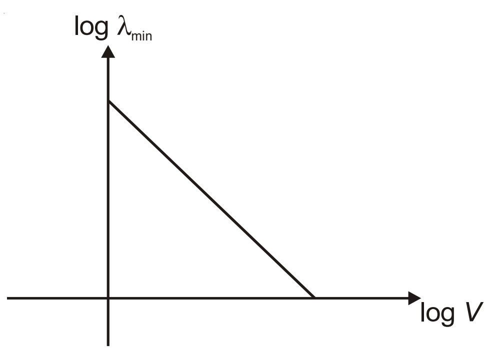
**Sol.** In X-ray tube

$$\lambda_{min} = \frac{hc}{eV}$$

$$\ln \lambda_{min} = \ln\left(\frac{hc}{e}\right) - \ln V$$

Slope is negative

Intercept on $y$-axis is positive




9. A radioactive nucleus $A$ with a half life $T$, decays into a nucleus $B$. At $t$ = 0, there is no nucleus $B$. At sometime $t$, the ratio of the number of $B$ to that of $A$ is 0.3. Then, $t$ is given by

(1) $t = T\log(1.3)$

(2) $t = \dfrac{T}{\log(1.3)}$

(3) $t = \dfrac{T}{2}\dfrac{\log 2}{\log 1.3}$

(4) $t = T\dfrac{\log 1.3}{\log 2}$

**Answer (4)**

**Sol.** $\dfrac{N_0 - N_0 e^{-\lambda t}}{N_0 e^{-\lambda t}} = 0.3$

$\Rightarrow \ e^{\lambda t} = 1.3$

$\therefore \ \lambda t = \ln 1.3$

$$\left(\frac{\ln 2}{T}\right)t = \ln 1.3$$

$$t = T.\frac{\ln(1.3)}{\ln 2}$$

$$t = T\frac{\log(1.3)}{\log 2}$$

10. An electric dipole has a fixed dipole moment $\vec{p}$, which makes angle $\theta$ with respect to $x$-axis. When subjected to an electric field $\vec{E}_1 = E\hat{i}$, it experiences a torque $\vec{T}_1 = \tau\hat{k}$. When subjected to another electric field $\vec{E}_2 = \sqrt{3}E_1\hat{j}$ it experiences a torque $\vec{T}_2 = -\vec{T}_1$. The angle $\theta$ is

(1) 60°

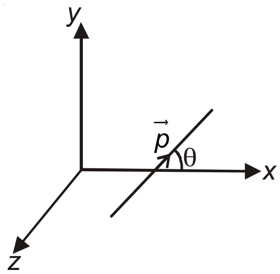
(2) 90°

(3) 30°

(4) 45°

**Answer (1)**

**Sol.**




$$\vec{p} = p\cos\theta\hat{i} + p\sin\theta\hat{j}$$

$$\vec{E}_1 = E\hat{i}$$

$$\vec{T}_1 = \vec{p}\times\vec{E}_1$$

$$= (p\cos\theta\hat{i} + p\sin\theta\hat{j})\times E\left(\hat{i}\right)$$

$$\tau\hat{k} = pE\sin\theta\left(-\hat{k}\right) \quad \ldots\text{(i)}$$

$$\vec{E}_2 = \sqrt{3}E_1\hat{j}$$

$$\vec{T_2} = (p\cos\theta\hat{i} + p\sin\theta\hat{j})\times\sqrt{3}E_1\hat{j}$$

$$-\tau\hat{k} = \sqrt{3}pE_1\cos\theta\hat{k} \quad \ldots\text{(ii)}$$

From (i) and (ii)

$$pE\sin\theta = \sqrt{3}pE\cos\theta$$

$$\tan\theta = \sqrt{3}$$

$\theta = 60°$

11. In a common emitter amplifier circuit using an n-p-n transistor, the phase difference between the input and the output voltages will be

(1) 135° (2) 180°

(3) 45° (4) 90°

**Answer (2)**

**Sol.** In common emitter configuration for n-p-n transistor, phase difference between output and input voltage is 180°.

12. $C_p$ and $C_v$ are specific heats at constant pressure and constant volume respectively. It is observed that

$C_p - C_v = a$ for hydrogen gas

$C_p - C_v = b$ for nitrogen gas

The correct relation between $a$ and $b$ is :

(1) $a = 14b$ (2) $a = 28b$

(3) $a = \dfrac{1}{14}b$ (4) $a = b$

**Answer (1)**

**Sol.** Let molar heat capacity at constant pressure = $X_p$

and molar heat capacity at constant volume = $X_v$

$$X_p - X_v = R$$

$$MC_p - MC_v = R$$

$$C_p - C_v = \frac{R}{M}$$

For hydrogen; $a = \dfrac{R}{2}$

For $N_2$; $b = \dfrac{R}{28}$

$$\frac{a}{b} = 14$$

$$a = 14b$$

13. A copper ball of mass 100 gm is at a temperature $T$. It is dropped in a copper calorimeter of mass 100 gm, filled with 170 gm of water at room temperature. Subsequently, the temperature of the system is found to be 75°C. $T$ is given by :

(Given : room temperature = 30°C, specific heat of copper = 0.1 cal/gm°C)

(1) 1250°C (2) 825°C

(3) 800°C (4) 885°C

**Answer (4)**

**Sol.** $100 \times 0.1 \times (t - 75) = 100 \times 0.1 \times 45 + 170 \times 1 \times 45$

$10t - 750 = 450 + 7650$

$10t = 1200 + 7650$

$10t = 8850$

$t = 885°C$

14. A body of mass $m = 10^{-2}$ kg is moving in a medium and experiences a frictional force $F = -kv^2$. Its initial speed is $v_0 = 10\ ms^{-1}$. If, after 10 s, its energy is $\dfrac{1}{8}mv_0^2$, the value of $k$ will be

(1) $10^{-4}$ kg $m^{-1}$ (2) $10^{-1}$ kg $m^{-1}$ $s^{-1}$

(3) $10^{-3}$ kg $m^{-1}$ (4) $10^{-3}$ kg $s^{-1}$

**Answer (1)**

**Sol.** $\dfrac{k_f}{k_i} = \dfrac{\frac{1}{8}mv_0^2}{\frac{1}{2}mv_0^2} = \dfrac{1}{4}$

$$\frac{v_f}{v_i} = \frac{1}{2}$$

$$v_f = \frac{v_0}{2}$$

$$-kv^2 = \frac{mdv}{dt}$$

$$\int_{v_0}^{\frac{v_0}{2}} \frac{dv}{v^2} = \int_0^{t_0} \frac{-kdt}{m}$$

$$\left[-\frac{1}{v}\right]_{v_0}^{\frac{v_0}{2}} = \frac{-k}{m}t_0$$

$$\frac{1}{v_0} - \frac{2}{v_0} = -\frac{k}{m}t_0$$

$$-\frac{1}{v_0} = -\frac{k}{m}t_0$$

$$k = \frac{m}{v_0 t_0}$$

$$= \frac{10^{-2}}{10 \times 10}$$

$= 10^{-4}$ kg m$^{-1}$

15. When a current of 5 mA is passed through a galvanometer having a coil of resistance 15 Ω, it shows full scale deflection. The value of the resistance to be put in series with the galvanometer to convert it into a voltmeter of range 0-10 V is

(1) $2.535 \times 10^3$ Ω

(2) $4.005 \times 10^3$ Ω

(3) $1.985 \times 10^3$ Ω

(4) $2.045 \times 10^3$ Ω

**Answer (3)**

**Sol.** $i_g = 5 \times 10^{-3}$ A

$G = 15$ Ω

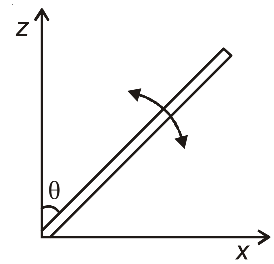
Let series resistance be $R$.

$V = i_g(R + G)$

$10 = 5 \times 10^{-3}(R + 15)$

$R = 2000 - 15 = 1985 = 1.985 \times 10^3$ Ω

16. A slender uniform rod of mass $M$ and length $l$ is pivoted at one end so that it can rotate in a vertical plane (see figure). There is negligible friction at the pivot. The free end is held vertically above the pivot and then released. The angular acceleration of the rod when it makes an angle θ with the vertical is




(1) $\frac{3g}{2\ell}\cos\theta$

(2) $\frac{2g}{3\ell}\cos\theta$

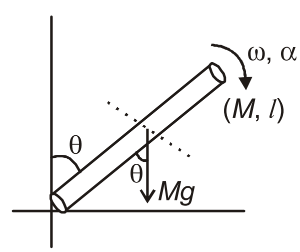
(3) $\frac{3g}{2\ell}\sin\theta$

(4) $\frac{2g}{3\ell}\sin\theta$

**Answer (3)**

**Sol.** Torque at angle θ

$$\tau = Mg\sin\theta \cdot \frac{\ell}{2}$$




$$\tau = I\alpha$$

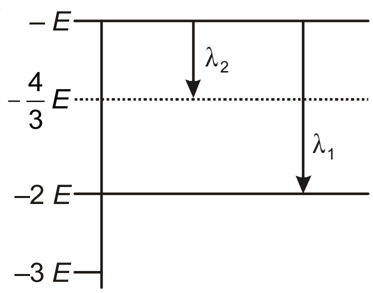
$$I\alpha = Mg\sin\theta\frac{\ell}{2} \qquad \therefore\ I = \frac{M\ell^2}{3}$$

$$\frac{M\ell^2}{3}\cdot\alpha = Mg\sin\theta\frac{\ell}{2}$$

$$\frac{\ell\alpha}{3} = g\frac{\sin\theta}{2}$$

$$\alpha = \frac{3g\sin\theta}{2\ell}$$

17. Some energy levels of a molecule are shown in the figure. The ratio of the wavelengths $r = \lambda_1/\lambda_2$, is given by




(1) $r = \frac{3}{4}$ (2) $r = \frac{1}{3}$

(3) $r = \frac{4}{3}$ (4) $r = \frac{2}{3}$

**Answer (2)**

**Sol.** From energy level diagram

$$\lambda_1 = \frac{hc}{E}$$

$$\lambda_2 = \frac{hc}{\left(\frac{E}{3}\right)}$$

$$\therefore \quad \frac{\lambda_1}{\lambda_2} = \frac{1}{3}$$

18. A man grows into a giant such that his linear dimensions increase by a factor of 9. Assuming that his density remains same, the stress in the leg will change by a factor of

(1) 81 (2) $\frac{1}{81}$

(3) 9 (4) $\frac{1}{9}$

**Answer (3)**

**Sol.** $\frac{v_f}{v_i} = 9^3$

$\because$ Density remains same

So, mass $\propto$ Volume

$$\frac{m_f}{m_i} = 9^3$$

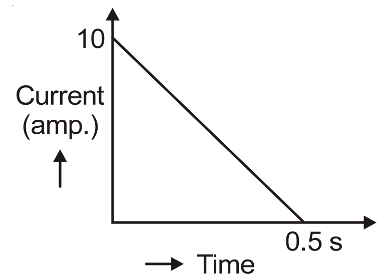
$$\frac{(\text{Area})_f}{(\text{Area})_i} = 9^2$$

$$\text{Stress} = \frac{(\text{Mass}) \times g}{\text{Area}}$$

$$\frac{\sigma_2}{\sigma_1} = \left(\frac{m_f}{m_i}\right)\left(\frac{A_i}{A_f}\right)$$

$$= \frac{9^3}{9^2} = 9$$

19. In a coil of resistance 100 Ω, a current is induced by changing the magnetic flux through it as shown in the figure. The magnitude of change in flux through the coil is




(1) 250 Wb (2) 275 Wb

(3) 200 Wb (4) 225 Wb

**Answer (1)**

**Sol.** $\varepsilon = \frac{d\phi}{dt}$

$$iR = \frac{d\phi}{dt}$$

$$\int d\phi = R\int i\,dt$$

Magnitude of change in flux = $R$ × area under current vs time graph

$$= 100 \times \frac{1}{2} \times \frac{1}{2} \times 10$$

$$= 250 \text{ Wb}$$

20. In a Young's double slit experiment, slits are separated by 0.5 mm, and the screen is placed 150 cm away. A beam of light consisting of two wavelengths, 650 nm and 520 nm, is used to obtain interference fringes on the screen. The least distance from the common central maximum to the point where the bright fringes due to both the wavelengths coincide is

(1) 9.75 mm (2) 15.6 mm

(3) 1.56 mm (4) 7.8 mm

**Answer (4)**

**Sol.** For $\lambda_1$

$$y = \frac{m\lambda_1 D}{d}$$

For $\lambda_2$

$$y = \frac{n\lambda_2 D}{d}$$

$$\Rightarrow \quad \frac{m}{n} = \frac{\lambda_2}{\lambda_1} = \frac{4}{5}$$

For $\lambda_1$

$$y = \frac{m\lambda_1 D}{d}, \ \lambda_1 = 650 \text{ nm}$$

$$= 7.8 \text{ mm}$$

21. A magnetic needle of magnetic moment $6.7 \times 10^{-2}$ Am$^2$ and moment of inertia $7.5 \times 10^{-6}$ kg m$^2$ is performing simple harmonic oscillations in a magnetic field of 0.01 T. Time taken for 10 complete oscillations is

(1) 6.98 s (2) 8.76 s

(3) 6.65 s (4) 8.89 s

**Answer (3)**

**Sol.** $T = 2\pi\sqrt{\frac{I}{MB}}$

$$= 2\pi\sqrt{\frac{7.5\times10^{-6}}{6.7\times10^{-2}\times0.01}} = \frac{2\pi}{10}\times1.06$$

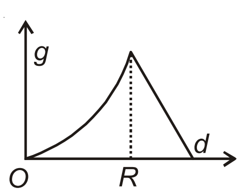
For 10 oscillations,

$t = 10T = 2\pi \times 1.06$

$= 6.6568 \approx 6.65$ s

22. The variation of acceleration due to gravity $g$ with distance $d$ from centre of the earth is best represented by ($R$ = Earth's radius) :

(1)

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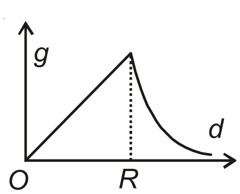


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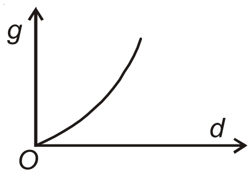


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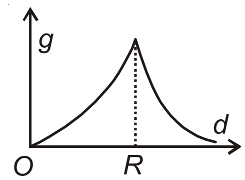
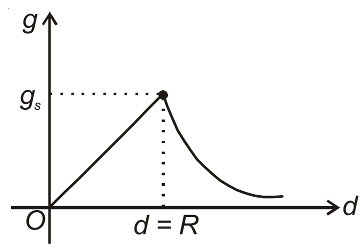



**Answer (2)**

**Sol.**



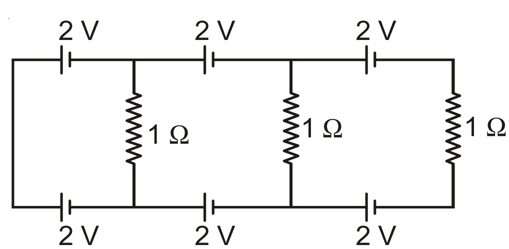

Variation of $g$ inside earth surface

$$d < R = g = \frac{Gm}{R^2}\cdot d$$

$$d = R = g_s = \frac{Gm}{R^2}$$

$$d > R = g = \frac{Gm}{d^2}$$

23.




In the above circuit the current in each resistance is

(1) 0.5 A (2) 0 A

(3) 1 A (4) 0.25 A

**Answer (2)**

**Sol.** The potential difference in each loop is zero.

$\therefore$ No current will flow.

with a particle $B$ of mass $\frac{m}{2}$ which is at rest. The collision is head on, and elastic. The ratio of the de-Broglie wavelengths $\lambda_A$ to $\lambda_B$ after the collision is

(1) $\frac{\lambda_A}{\lambda_B} = \frac{2}{3}$ (2) $\frac{\lambda_A}{\lambda_B} = \frac{1}{2}$

(3) $\frac{\lambda_A}{\lambda_B} = \frac{1}{3}$ (4) $\frac{\lambda_A}{\lambda_B} = 2$

**Answer (4)**

**Sol.** $v_1 = \frac{(m_1 - m_2)v}{m_1 + m_2} + 0$

$m_1 = m$

$m_2 = \frac{m}{2}$

$= \frac{v}{3}$

$\therefore \quad p_1 = m.\left[\frac{v}{3}\right]$

$v_2 = \frac{2m_1 v}{m_1 + m_2} + 0$

$= \frac{4v}{3}$

$p_2 = \frac{m}{2}\left[\frac{4v}{3}\right] = \frac{2mv}{3}$

$\therefore$ de-Broglie wavelength $\frac{\lambda_A}{\lambda_B} = \frac{p_2}{p_1} = 2:1$

25. An external pressure $P$ is applied on a cube at 0°C so that it is equally compressed from all sides. $K$ is the bulk modulus of the material of the cube and $\alpha$ is its coefficient of linear expansion. Suppose we want to bring the cube to its original size by heating. The temperature should be raised by :

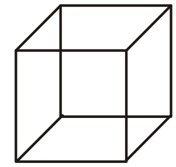
(1) $\frac{3\alpha}{PK}$ (2) $3PK\alpha$

(3) $\frac{P}{3\alpha K}$ (4) $\frac{P}{\alpha K}$

**Answer (3)**

**Sol.** $K = \frac{\Delta P}{\left(-\frac{\Delta V}{V}\right)}$

$\frac{\Delta V}{V} = \frac{P}{K}$



$\gamma = 3\alpha$

$\therefore\ V = V_0\,(1 + \gamma\Delta t)$

$\frac{\Delta V}{V_0} = \gamma\Delta t$

$\therefore\ \frac{P}{K} = \gamma\Delta t \Rightarrow \Delta t = \frac{P}{\gamma K} = \frac{P}{3\alpha K}$

26. A time dependent force $F = 6t$ acts on a particle of mass 1 kg. If the particle starts from rest, the work done by the force during the first 1 second will be

(1) 9 J (2) 18 J

(3) 4.5 J (4) 22 J

**Answer (3)**

**Sol.** $6t = 1\cdot\frac{dv}{dt}$

$\int_0^v dv = \int 6t\ dt$

$v = 6\left[\frac{t^2}{2}\right]_0^1$

$= 3\ \text{ms}^{-1}$

$W = \Delta\text{KE} = \frac{1}{2}\times 1\times 9 = 4.5\ \text{J}$

27. An observer is moving with half the speed of light towards a stationary microwave source emitting waves at frequency 10 GHz. What is the frequency of the microwave measured by the observer? (speed of light = $3 \times 10^8\ \text{ms}^{-1}$)

(1) 17.3 GHz (2) 15.3 GHz

(3) 10.1 GHz (4) 12.1 GHz

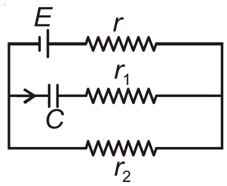
**Answer (1)**

**Sol.** For relativistic motion

$f = f_0\sqrt{\frac{c+v}{c-v}}$ ; $v$ = relative speed of approach

$f = 10\sqrt{\frac{c+\frac{c}{2}}{c-\frac{c}{2}}} = 10\sqrt{3} = 17.3\ \text{GHz}$

28. In the given circuit diagram when the current reaches steady state in the circuit, the charge on the capacitor of capacitance $C$ will be :

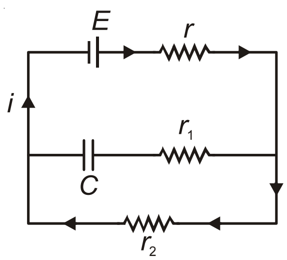



(1) $CE\dfrac{r_2}{(r+r_2)}$ (2) $CE\dfrac{r_1}{(r_1+r)}$

(3) $CE$ (4) $CE\dfrac{r_1}{(r_2+r)}$

**Answer (1)**

**Sol.** In steady state, flow of current through capacitor will be zero.




$$i = \frac{E}{r+r_2}$$

$$V_C = i\, r_2 C = \frac{Er_2C}{r+r_2}$$

$$V_C = CE\,\frac{r_2}{r+r_2}$$

29. A capacitance of 2 μF is required in an electrical circuit across a potential difference of 1.0 kV. A large number of 1 μF capacitors are available which can withstand a potential difference of not more than 300 V.

The minimum number of capacitors required to achieve this is

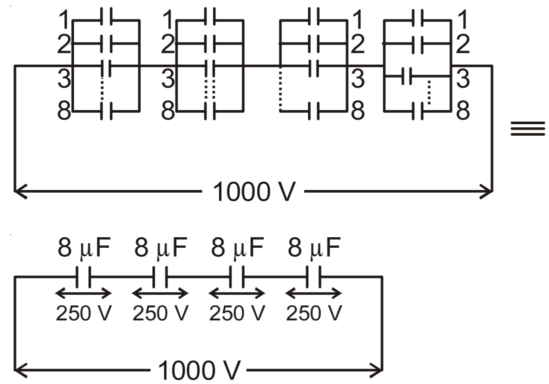
(1) 24 (2) 32

(3) 2 (4) 16

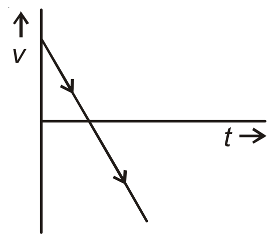
**Answer (2)**

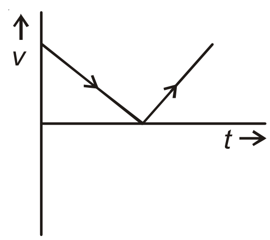
**Sol.** Following arrangement will do the needful :

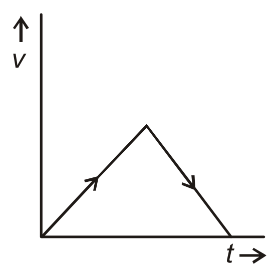
8 capacitors of 1μF in parallel with four such branches in series.

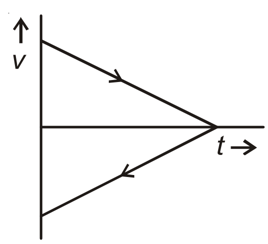



30. A body is thrown vertically upwards. Which one of the following graphs correctly represent the velocity vs time?

(1)


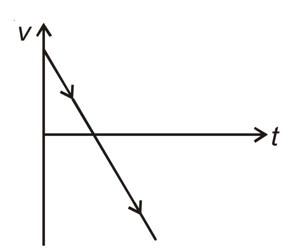

(2)



(3)



(4)



**Answer (1)**

**Sol.** Acceleration is constant and negative



31. Let $k$ be an integer such that the triangle with vertices $(k, -3k)$, $(5, k)$ and $(-k, 2)$ has area 28 sq. units. Then the orthocentre of this triangle is at the point

(1) $\left(2, \frac{1}{2}\right)$ (2) $\left(2, -\frac{1}{2}\right)$

(3) $\left(1, \frac{3}{4}\right)$ (4) $\left(1, -\frac{3}{4}\right)$

**Answer (1)**

**Sol.** Area = $\left|\frac{1}{2}\begin{vmatrix} k & -3k & 1 \\ 5 & k & 1 \\ -k & 2 & 1 \end{vmatrix}\right| = 28$

$$\begin{vmatrix} k-5 & -4k & 0 \\ 5+k & k-2 & 0 \\ -k & 2 & 1 \end{vmatrix} = \pm 56$$

$$(k^2 - 7k + 10) + 4k^2 + 20k = \pm 56$$

$$5k^2 + 13k + 10 = \pm 56$$

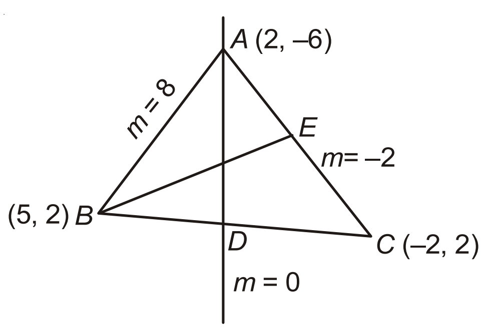
$$5k^2 + 13k - 46 = 0 \quad \Big| \quad 5K^2 + 13K + 66 = 0$$

$$5k^2 + 13k - 46 = 0$$

$$k = \frac{-13 \pm \sqrt{169 + 920}}{10}$$

= 2, –4.6
reject

For $k = 2$




Equation of $AD$,

$x = 2$ ...(i)

Also equation of $BE$,

$$y - 2 = \frac{1}{2}(x - 5)$$

$$2y - 4 = x - 5$$

$$x - 2y - 1 = 0 \quad \text{...(ii)}$$

Solving (i) & (ii), $2y = 1$

$$y = \frac{1}{2}$$

Orthocentre is $\left(2, \frac{1}{2}\right)$

32. If, for a positive integer $n$, the quadratic equation, $x(x+1) + (x+1)(x+2) + \ldots + (x+\overline{n-1})(x+n) = 10n$ has two consecutive integral solutions, then $n$ is equal to

(1) 11 (2) 12

(3) 9 (4) 10

**Answer (1)**

**Sol.** Rearranging equation, we get

$$nx^2 + \{1 + 3 + 5 + \ldots. + (2n-1)\}x$$

$$+\{1\cdot 2 + 2\cdot 3 + \ldots + (n-1)n\} = 10n$$

$$\Rightarrow nx^2 + n^2x + \frac{(n-1)n(n+1)}{3} = 10n$$

$$\Rightarrow x^2 + nx + \left(\frac{n^2 - 31}{3}\right) = 0$$

Given difference of roots = 1

$\Rightarrow$ $|\alpha - \beta| = 1$

$\Rightarrow$ $D = 1$

$$\Rightarrow n^2 - \frac{4}{3}(n^2 - 31) = 1$$

So, $n = 11$

33. The function $f : R \to \left[-\frac{1}{2}, \frac{1}{2}\right]$ defined as $f(x) = \frac{x}{1+x^2}$, is

(1) Neither injective nor surjective

(2) Invertible

(3) Injective but not surjective

(4) Surjective but not injective

**Answer (4)**

**Sol.** $f(x) = \dfrac{x}{1+x^2}$

$$f'(x) = \frac{(1+x^2)\cdot 1 - x\cdot 2x}{(1+x^2)^2} = \frac{1-x^2}{(1+x^2)^2}$$

$f'(x)$ changes sign in different intervals.

$\therefore$ Not injective.

$$y = \frac{x}{1+x^2}$$

$$yx^2 - x + y = 0$$

For $y \neq 0$

$$D = 1 - 4y^2 \geq 0 \;\Rightarrow\; y \in \left[-\frac{1}{2}, \frac{1}{2}\right] - \{0\}$$

For, $y = 0 \;\Rightarrow x = 0$

$\therefore$ Part of range

$\therefore$ Range : $\left[-\dfrac{1}{2}, \dfrac{1}{2}\right]$

$\therefore$ Surjective but not injective.

34. The following statement $(p \to q) \to [(\sim p \to q) \to q]$ is

(1) A fallacy

(2) A tautology

(3) Equivalent to $\sim p \to q$

(4) Equivalent to $p \to \sim q$

**Answer (2)**

**Sol.**

| $p$ | $q$ | $p \to q$ | $(\sim p \to q)$ | $(\sim p \to q) \to q$ | $(p \to q) \to [(\sim p \to q) \to q]$ |
|---|---|---|---|---|---|
| T | T | T | T | T | T |
| T | F | F | T | F | T |
| F | T | T | T | T | T |
| F | F | T | F | T | T |

(a tautology)

35. If $S$ is the set of distinct values of $b$ for which the following system of linear equations

$$x + y + z = 1$$

$$x + ay + z = 1$$

$$ax + by + z = 0$$

has no solution, then $S$ is

(1) A singleton

(2) An empty set

(3) An infinite set

(4) A finite set containing two or more elements

**Answer (1)**

**Sol.**

$$\begin{vmatrix} 1 & 1 & 1 \\ 1 & a & 1 \\ a & b & 1 \end{vmatrix} = 0$$

$\Rightarrow -(1-a)^2 = 0$

$\Rightarrow a = 1$

For $a = 1$

Eq. (1) & (2) are identical *i.e.*, $x + y + z = 1$

To have no solution with $x + by + z = 0$

$b = 1$

36. The area (in sq. units) of the region

$\{(x, y) : x \geq 0,\ x + y \leq 3,\ x^2 \leq 4y \text{ and } y \leq 1 + \sqrt{x}\}$

is

(1) $\dfrac{5}{2}$

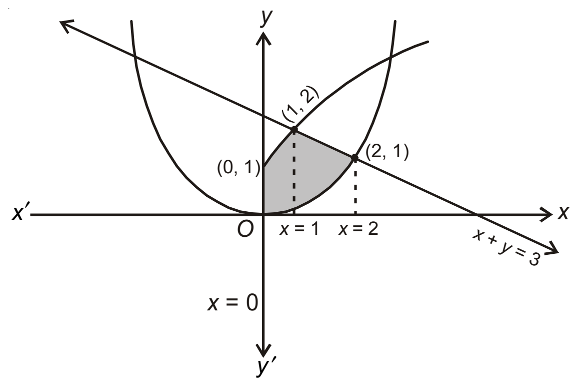
(2) $\dfrac{59}{12}$

(3) $\dfrac{3}{2}$

(4) $\dfrac{7}{3}$

**Answer (1)**

**Sol.**




Area of shaded region

$$= \int_0^1 \left(\sqrt{x} + 1 - \frac{x^2}{4}\right)dx + \int_1^2 \left((3-x) - \frac{x^2}{4}\right)dx$$

$$= \frac{5}{2} \text{ sq. unit}$$

37. For any three positive real numbers $a$, $b$ and $c$,

$$9(25a^2 + b^2) + 25(c^2 - 3ac) = 15b(3a + c).$$

Then

(1) $a$, $b$ and $c$ are in G.P.

(2) $b$, $c$ and $a$ are in G.P.

(3) $b$, $c$ and $a$ are in A.P.

(4) $a$, $b$ and $c$ are in A.P.

**Answer (3)**

**Sol.** $9(25a^2+b^2)+25(c^2-3ac)=15b(3a+c)$

$\Rightarrow \quad (15a)^2+(3b)^2+(5c)^2-45ab-15bc-75ac=0$

$\Rightarrow \quad (15a-3b)^2+(3b-5c)^2+(15a-5c)^2=0$

It is possible when

$15a-3b=0$ and $3b-5c=0$ and $15a-5c=0$

$15a=3b=5c$

$\frac{a}{1}=\frac{b}{5}=\frac{c}{3}$

$\therefore \quad b, c, a$ are in A.P.

38. A man $X$ has 7 friends, 4 of them are ladies and 3 are men. His wife $Y$ also has 7 friends, 3 of them are ladies and 4 are men. Assume $X$ and $Y$ have no common friends. Then the total number of ways in which $X$ and $Y$ together can throw a party inviting 3 ladies and 3 men, so that 3 friends of each of $X$ and $Y$ are in this party, is

(1) 484 (2) 485

(3) 468 (4) 469

**Answer (2)**

**Sol.**

| | |
|---|---|
| $X$(4 L 3 G) | $Y$(3 L 4 G) |
| 3 L 0 G | 0 L 3 G |
| 2 L 1 G | 1 L 2 G |
| 1 L 2 G | 2 L 1 G |
| 0 L 3 G | 3 L 0 G |

Required number of ways

$= {}^4C_3\cdot{}^4C_3+\left({}^4C_2\cdot{}^3C_1\right)^2+\left({}^4C_1\cdot{}^3C_2\right)^2+\left({}^3C_3\right)^2$

= 16 + 324 + 144 + 1

= 485

39. The normal to the curve $y(x-2)(x-3)=x+6$ at the point where the curve intersects the $y$-axis passes through the point

(1) $\left(\frac{1}{2},\frac{1}{3}\right)$ (2) $\left(-\frac{1}{2},-\frac{1}{2}\right)$

(3) $\left(\frac{1}{2},\frac{1}{2}\right)$ (4) $\left(\frac{1}{2},-\frac{1}{3}\right)$

**Answer (3)**

**Sol.** $y(x-2)(x-3)=x+6$

At $y$-axis, $x = 0$, $y = 1$

Now, on differentiation.

$\frac{dy}{dx}(x-2)(x-3)+y(2x-5)=1$

$\frac{dy}{dx}(6)+1(-5)=1$

$\frac{dy}{dx}=\frac{6}{6}=1$

Now slope of normal = –1

Equation of normal $y - 1 = -1(x - 0)$

$y + x - 1 = 0$ ... (i)

Line (i) passes through $\left(\frac{1}{2}, \frac{1}{2}\right)$

40. A hyperbola passes through the point $P(\sqrt{2},\sqrt{3})$ and has foci at (±2, 0). Then the tangent to this hyperbola at $P$ also passes through the point

(1) $(-\sqrt{2},-\sqrt{3})$ (2) $(3\sqrt{2},2\sqrt{3})$

(3) $(2\sqrt{2},3\sqrt{3})$ (4) $(\sqrt{3},\sqrt{2})$

**Answer (3)**

**Sol.** $\frac{x^2}{a^2}-\frac{y^2}{b^2}=1$

$a^2+b^2=4$

and $\frac{2}{a^2}-\frac{3}{b^2}=1$

$\frac{2}{4-b^2}-\frac{3}{b^2}=1$

$\Rightarrow \quad b^2=3$

$\therefore \quad a^2=1$

$\therefore \quad x^2-\frac{y^2}{3}=1$

$\therefore \quad$ Tangent at $P(\sqrt{2},\sqrt{3})$ is $\sqrt{2}x-\frac{y}{\sqrt{3}}=1$

Clearly it passes through $(2\sqrt{2},3\sqrt{3})$

41. Let $a, b, c \in R$. If $f(x) = ax^2 + bx + c$ is such that $a + b + c = 3$ and

$f(x+y)=f(x)+f(y)+xy,\ \forall\ x, y\in R,$

then $\sum_{n=1}^{10} f(n)$ is equal to

(1) 255 (2) 330

(3) 165 (4) 190

**Answer (2)**

**Sol.** As, $f(x+y)=f(x)+f(y)+xy$

Given, $f(1)=3$

Putting, $x=y=1 \Rightarrow f(2)=2f(1)+1=7$

Similarly, $x=1, y=2 \Rightarrow f(3)=f(1)+f(2)+2=12$

Now, $\sum_{n=1}^{10} f(n) = f(1)+f(2)+f(3)+...+f(10)$

$= 3 + 7 + 12 + 18 + ... = S$ (let)

Now, $S=3+7+12+18+...+t_n$

Again, $S=3+7+12+...+t_{n-1}+t_n$

We get, $t_n=3+4+5+...n$ terms

$$=\frac{n(n+5)}{2}$$

*i.e.*, $S_n = \sum_{n=1}^{n} t_n$

$$=\frac{1}{2}\left\{\sum n^2+5\sum n\right\}$$

$$=\frac{n(n+1)(n+8)}{6}$$

So, $S_{10} = \frac{10\times 11\times 18}{6}=330$

42. Let $\vec{a}=2\hat{i}+\hat{j}-2\hat{k}$ and $\vec{b}=\hat{i}+\hat{j}$. Let $\vec{c}$ be a vector such that $|\vec{c}-\vec{a}|=3, \left|(\vec{a}\times\vec{b})\times\vec{c}\right|=3$ and the angle between $\vec{c}$ and $\vec{a}\times\vec{b}$ be 30°. Then $\vec{a}\cdot\vec{c}$ is equal to

(1) $\frac{1}{8}$

(2) $\frac{25}{8}$

(3) 2

(4) 5

**Answer (3)**

**Sol.** $|(\vec{a}\times\vec{b})\times\vec{c}|=3$ $\qquad$ $\vec{a}\times\vec{b}=2\hat{i}-2\hat{j}+\hat{k}$

$\Rightarrow |\vec{a}\times\vec{b}||\vec{c}|\sin 30^\circ=3$ $\qquad$ $|\vec{a}|=3=|\vec{a}\times\vec{b}|$

$\Rightarrow |\vec{c}|=2$

$|\vec{c}-\vec{a}|=3$

$\Rightarrow |\vec{c}|^2+|\vec{a}|^2-2(\vec{a}\cdot\vec{c})=9$

$$\vec{a}\cdot\vec{c}=\frac{9-3-2}{2}=2$$

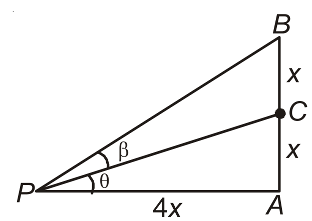
43. Let a vertical tower $AB$ have its end $A$ on the level ground. Let $C$ be the mid-point of $AB$ and $P$ be a point on the ground such that $AP = 2AB$. If $\angle BPC = \beta$ then $\tan\beta$ is

(1) $\frac{4}{9}$ $\qquad$ (2) $\frac{6}{7}$

(3) $\frac{1}{4}$ $\qquad$ (4) $\frac{2}{9}$

**Answer (4)**

**Sol.** $\tan\theta=\frac{1}{4}$




$\tan(\theta+\beta)=\frac{1}{2}$

$$\therefore \frac{\frac{1}{4}+\tan\beta}{1-\frac{1}{4}\tan\beta}=\frac{1}{2}$$

Solving $\tan\beta=\frac{2}{9}$

44. Twenty meters of wire is available for fencing off a flower-bed in the form of a circular sector. Then the maximum area (in sq. m) of the flower-bed, is

(1) 30

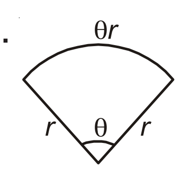
(2) 12.5

(3) 10

(4) 25

**Answer (4)**

**Sol.**




$2r+\theta r=20$ $\qquad$ ... (i)

$\boldsymbol{A}$ = area = $\frac{\theta}{2\pi}\times\pi r^2=\frac{\theta r^2}{2}$ $\qquad$ ... (ii)

$$A=\frac{r^2}{2}\left(\frac{20-2r}{r}\right)$$

$$A = \left(\frac{20r - 2r^2}{2}\right) = 10r - r^2$$

A to be maximum

$$\frac{dA}{dr} = 10 - 2r = 0 \Rightarrow r = 5$$

$$\frac{d^2A}{dr^2} = -2 < 0$$

Hence for $r$ = 5, $A$ is maximum

Now, 10 + $\theta$·5 = 20 $\Rightarrow$ $\theta$ = 2 (radian)

Area = $\frac{2}{2\pi} \times \pi(5)^2 = 25$ sq m

45. The integral $\int_{\frac{\pi}{4}}^{\frac{3\pi}{4}} \frac{dx}{1+\cos x}$ is equal to

(1) –1 (2) –2

(3) 2 (4) 4

**Answer (3)**

**Sol.** $\int_{\frac{\pi}{4}}^{\frac{3\pi}{4}} \frac{dx}{2\cos^2 \frac{x}{2}} dx = \frac{1}{2} \int_{\frac{\pi}{4}}^{\frac{3\pi}{4}} \sec^2 \frac{x}{2} dx$

$$= \frac{1}{2}\left[\frac{\tan \frac{x}{2}}{\frac{1}{2}}\right]_{\frac{\pi}{4}}^{\frac{3\pi}{4}}$$

$$= \tan\frac{3\pi}{8} - \tan\frac{\pi}{8}$$

$$\left[\tan\frac{\pi}{8} = \sqrt{\frac{1-\cos\frac{\pi}{4}}{1+\cos\frac{\pi}{4}}} = \sqrt{\frac{\sqrt{2}-1}{\sqrt{2}+1}} = \frac{\sqrt{2}-1}{1}\right.$$

$$\left.\tan\frac{3\pi}{8} = \sqrt{\frac{1-\cos\frac{3\pi}{4}}{1+\cos\frac{3\pi}{4}}} = \sqrt{\frac{\sqrt{2}+1}{\sqrt{2}-1}} = \sqrt{2}+1\right]$$

$$= (\sqrt{2}+1) - (\sqrt{2}-1)$$

$$= 2$$

46. If $(2+\sin x)\frac{dy}{dx} + (y+1)\cos x = 0$ and $y(0)$ = 1, then $y\left(\frac{\pi}{2}\right)$ is equal to

(1) $\frac{4}{3}$ (2) $\frac{1}{3}$

(3) $-\frac{2}{3}$ (4) $-\frac{1}{3}$

**Answer (2)**

**Sol.** $(2+\sin x)\frac{dy}{dx} + (y+1)\cos x = 0$

$$y(0) = 1,\ y\left(\frac{\pi}{2}\right) = ?$$

$$\frac{1}{y+1}dy + \frac{\cos x}{2+\sin x}dx = 0$$

$$\ln|y+1| + \ln(2+\sin x) = \ln C$$

$$(y+1)(2+\sin x) = C$$

Put $x$ = 0, $y$ = 1

$(1+1)\cdot 2 = C \Rightarrow C = 4$

Now, $(y+1)(2+\sin x) = 4$

For, $x = \frac{\pi}{2}$

$$(y+1)(2+1) = 4$$

$$y + 1 = \frac{4}{3}$$

$$y = \frac{4}{3} - 1 = \frac{1}{3}$$

47. Let $I_n = \int \tan^n x dx, (n > 1)$. If $I_4 + I_6 = a \tan^5 x + bx^5 + C$, where $C$ is a constant of integration, then the ordered pair ($a$, $b$) is equal to

(1) $\left(-\frac{1}{5}, 0\right)$ (2) $\left(-\frac{1}{5}, 1\right)$

(3) $\left(\frac{1}{5}, 0\right)$ (4) $\left(\frac{1}{5}, -1\right)$

**Answer (3)**

**Sol.** $I_n = \int \tan^n x dx,\ n > 1$

$$I_4 + I_6 = \int (\tan^4 x + \tan^6 x) dx$$

$$= \int \tan^4 x \sec^2 x dx$$

Let $\tan x = t$

$\sec^2 x\ dx = dt$

$$= \int t^4 dt$$

$$= \frac{t^5}{5} + C$$

$$= \frac{1}{5}\tan^5 x + C$$

$$a = \frac{1}{5},\ b = 0$$

48. Let $\omega$ be a complex number such that $2\omega + 1 = z$ where $z = \sqrt{-3}$. If

$$\begin{vmatrix} 1 & 1 & 1 \\ 1 & -\omega^2 - 1 & \omega^2 \\ 1 & \omega^2 & \omega^7 \end{vmatrix} = 3k, \text{ then } k \text{ is equal to}$$

(1) 1 (2) $-z$

(3) $z$ (4) $-1$

**Answer (2)**

**Sol.** $2\omega + 1 = z$, $z = \sqrt{3}\,i$

$$\omega = \frac{-1 + \sqrt{3}i}{2} \rightarrow \text{Cube root of unity.}$$

$$C_1 \rightarrow C_1 + C_2 + C_3$$

$$\begin{vmatrix} 1 & 1 & 1 \\ 1 & -1-\omega^2 & \omega^2 \\ 1 & \omega^2 & \omega^7 \end{vmatrix} = \begin{vmatrix} 1 & 1 & 1 \\ 1 & \omega & \omega^2 \\ 1 & \omega^2 & \omega \end{vmatrix} = \begin{vmatrix} 3 & 1 & 1 \\ 0 & \omega & \omega^2 \\ 0 & \omega^2 & \omega \end{vmatrix}$$

$= 3(\omega^2 - \omega^4)$

$$= 3\left[\left(\frac{-1-\sqrt{3}i}{2}\right) - \left(\frac{-1+\sqrt{3}i}{2}\right)\right]$$

$= -3\sqrt{3}i$

$= -3z$

$\therefore\ k = -z$

49. The value of

$$({}^{21}C_1 - {}^{10}C_1) + ({}^{21}C_2 - {}^{10}C_2) + ({}^{21}C_3 - {}^{10}C_3) + ({}^{21}C_4 - {}^{10}C_4) + \ldots + ({}^{21}C_{10} - {}^{10}C_{10}) \text{ is}$$

(1) $2^{20} - 2^{10}$

(2) $2^{21} - 2^{11}$

(3) $2^{21} - 2^{10}$

(4) $2^{20} - 2^9$

**Answer (1)**

**Sol.** ${}^{21}C_1 + {}^{21}C_2 + \ldots + {}^{21}C_{10} = \frac{1}{2}\left\{{}^{21}C_0 + {}^{21}C_1 + \ldots + {}^{21}C_{21}\right\} - 1$

$= 2^{20} - 1$

$\left({}^{10}C_1 + {}^{10}C_2 + \ldots + {}^{10}C_{10}\right) = 2^{10} - 1$

$\therefore$ Required sum $= (2^{20} - 1) - (2^{10} - 1)$

$= 2^{20} - 2^{10}$

50. $\lim\limits_{x \to \frac{\pi}{2}} \dfrac{\cot x - \cos x}{(\pi - 2\pi)^3}$ equals

(1) $\frac{1}{4}$

(2) $\frac{1}{24}$

(3) $\frac{1}{16}$

(4) $\frac{1}{8}$

**Answer (3)**

**Sol.** $\lim\limits_{x \to \frac{\pi}{2}} \dfrac{\cot x - \cos x}{(\pi - 2x)^3}$

Put, $\frac{\pi}{2} - x = t$

$$\lim_{t \to 0} \frac{\tan t - \sin t}{8t^3}$$

$$= \lim_{t \to 0} \frac{\sin t \cdot 2\sin^2 \frac{t}{2}}{8t^3}$$

$$= \frac{1}{16}.$$

51. If 5 ($\tan^2 x - \cos^2 x$) = 2cos 2$x$ + 9, then the value of cos 4$x$ is

(1) $-\frac{7}{9}$ (2) $-\frac{3}{5}$

(3) $\frac{1}{3}$ (4) $\frac{2}{9}$

**Answer (1)**

**Sol.** $5\tan^2 x = 9\cos^2 x + 7$

$5\sec^2 x - 5 = 9\cos^2 x + 7$

Let $\cos^2 x = t$

$\frac{5}{t} = 9t + 12$

$9t^2 + 12t - 5 = 0$

$t = \frac{1}{3}$ as $t \neq -\frac{5}{3}$

$\cos^2 x = \frac{1}{3}$, $\cos 2x = 2\cos^2 x - 1$

$= -\frac{1}{3}$

$\cos 4x = 2\cos^2 2x - 1$

$= \frac{2}{9} - 1$

$= -\frac{7}{9}$

52. If the image of the point $P(1, -2, 3)$ in the plane, $2x + 3y - 4z + 22 = 0$ measured parallel to the line, $\frac{x}{1} = \frac{y}{4} = \frac{z}{5}$ is $Q$, then $PQ$ is equal to

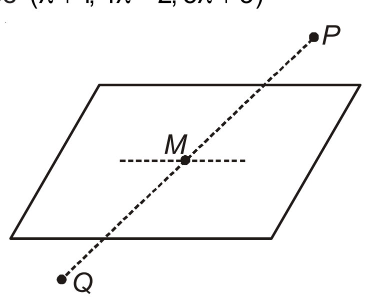
(1) $6\sqrt{5}$ (2) $3\sqrt{5}$

(3) $2\sqrt{42}$ (4) $\sqrt{42}$

**Answer (3)**

**Sol.** Equation of $PQ$, $\frac{x-1}{1} = \frac{y+2}{4} = \frac{z-3}{5}$

Let $M$ be $(\lambda+1, 4\lambda-2, 5\lambda+3)$




As it lies on $2x + 3y - 4z + 22 = 0$

$\lambda = 1$

For $Q$, $\lambda = 2$

Distance $PQ = 2\sqrt{1^2 + 4^2 + 5^2} = 2\sqrt{42}$

53. The distance of the point (1, 3, –7) from the plane passing through the point (1, –1, –1), having normal perpendicular to both the lines $\frac{x-1}{1} = \frac{y+2}{-2} = \frac{z-4}{3}$ and $\frac{x-2}{2} = \frac{y+1}{-1} = \frac{z+7}{-1}$, is

(1) $\frac{10}{\sqrt{74}}$ (2) $\frac{20}{\sqrt{74}}$

(3) $\frac{10}{\sqrt{83}}$ (4) $\frac{5}{\sqrt{83}}$

**Answer (3)**

**Sol.** Let the plane be

$a(x-1) + b(y+1) + c(z+1) = 0$

It is perpendicular to the given lines

$a - 2b + 3c = 0$

$2a - b - c = 0$

Solving, $a : b : c = 5 : 7 : 3$

$\therefore$ The plane is $5x + 7y + 3z + 5 = 0$

Distance of (1, 3, –7) from this plane = $\frac{10}{\sqrt{83}}$

54. If for $x \in \left(0, \frac{1}{4}\right)$, the derivative of $\tan^{-1}\left(\frac{6x\sqrt{x}}{1-9x^3}\right)$ is $\sqrt{x} \cdot g(x)$, then $g(x)$ equals

(1) $\frac{3}{1+9x^3}$ (2) $\frac{9}{1+9x^3}$

(3) $\frac{3x\sqrt{x}}{1-9x^3}$ (4) $\frac{3x}{1-9x^3}$

**Answer (2)**

**Sol.** $f(x) = 2\tan^{-1}(3x\sqrt{x})$ For $x \in \left(0, \frac{1}{4}\right)$

$f'(x) = \frac{9\sqrt{x}}{1+9x^3}$

$g(x) = \frac{9}{1+9x^3}$

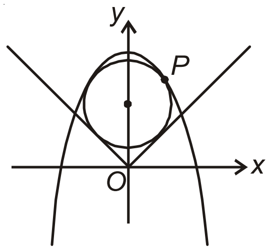
55. The radius of a circle, having minimum area, which touches the curve $y = 4 - x^2$ and the lines, $y = |x|$ is

(1) $4(\sqrt{2}+1)$ (2) $2(\sqrt{2}+1)$

(3) $2(\sqrt{2}-1)$ (4) $4(\sqrt{2}-1)$

**Answer (4)**

**Sol.**




$x^2 = -(y-4)$

Let a point on the parabola $P\left(\frac{t}{2}, 4-\frac{t^2}{4}\right)$

Equation of normal at $P$ is

$$y+\frac{t^2}{4}-4=\frac{1}{t}\left(x-\frac{t}{2}\right)$$

$$\Rightarrow \quad x-ty-\frac{t^3}{4}+\frac{7}{2}t=0$$

It passes through centre of circle, say (0, $k$)

$$-tk-\frac{t^3}{4}+\frac{7}{2}t=0 \qquad \text{...(i)}$$

$t = 0, \; t^2 = 14-4k$

Radius $= r=\left|\frac{0-k}{\sqrt{2}}\right|$ (Length of perpendicular from (0, $k$) to $y = x$)

$$\Rightarrow \quad r=\frac{k}{\sqrt{2}}$$

Equation of circle is $x^2+(y-k)^2=\frac{k^2}{2}$

It passes through point $P$

$$\frac{t^2}{4}+\left(4-\frac{t^2}{4}-k\right)^2=\frac{k^2}{2}$$

$$t^4+t^2(8k-28)+8k^2-128k+256=0 \qquad \text{...(ii)}$$

For $t = 0 \;\Rightarrow\; k^2-16k+32=0$

$k=8\pm4\sqrt{2}$

$\therefore \quad r=\frac{k}{\sqrt{2}}=4(\sqrt{2}-1)$ (discarding $4(\sqrt{2}+1)$) ...(iii)

For $t=\pm\sqrt{14-4k}$

$$(14-4k)^2+(14-4k)(8k-28)+8k^2-128k+256=0$$

$$2k^2+4k-15=0$$

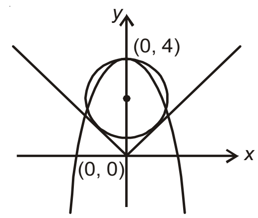
$$k=\frac{-2\pm\sqrt{34}}{2}$$

$\therefore \quad r=\frac{k}{\sqrt{2}}=\frac{\sqrt{17}-\sqrt{2}}{2}$ (Ignoring negative value of $r$) ...(iv)

From (iii) & (iv),

$$r_{\min}=\frac{\sqrt{17}-\sqrt{2}}{2}$$

But from options, $r=4(\sqrt{2}-1)$




56. A box contains 15 green and 10 yellow balls. If 10 balls are randomly drawn, one-by-one, with replacement, then the variance of the number of green balls drawn is

(1) $\frac{6}{25}$ (2) $\frac{12}{5}$

(3) 6 (4) 4

**Answer (2)**

**Sol.** $n$ = 10

$p$(Probability of drawing a green ball) = $\frac{15}{25}$

$\therefore \quad p=\frac{3}{5},\; q=\frac{2}{5}$

var(X) = $n.p.q$

$= 10\cdot\frac{6}{25}=\frac{12}{5}$

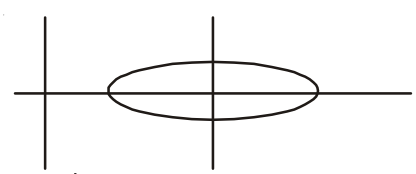
57. The eccentricity of an ellipse whose centre is at the origin is $\frac{1}{2}$. If one of its directrices is $x$ = – 4, then the equation of the normal to it at $\left(1, \frac{3}{2}\right)$ is

(1) $x + 2y = 4$ (2) $2y - x = 2$

(3) $4x - 2y = 1$ (4) $4x + 2y = 7$

**Answer (3)**

**Sol.**



$x = -4$

$e=\frac{1}{2}$

$\frac{-a}{e} = -4$

$-a = -4 \times e$

$\boxed{a = 2}$

Now, $b^2 = a^2(1-e^2) = 3$

Equation to ellipse

$$\frac{x^2}{4} + \frac{y^2}{3} = 1$$

Equation of normal is

$$\frac{x-1}{\frac{1}{4}} = \frac{y-\frac{3}{2}}{\frac{3}{2\times 3}}$$

$\Rightarrow 4x - 2y - 1 = 0$

58. If two different numbers are taken from the set {0, 1, 2, 3, ......, 10}; then the probability that their sum as well as absolute difference are both multiple of 4, is

(1) $\frac{7}{55}$ (2) $\frac{6}{55}$

(3) $\frac{12}{55}$ (4) $\frac{14}{45}$

**Answer (2)**

**Sol.** Total number of ways = $^{11}C_2$
= 55

Favourable ways are

(0, 4), (0, 8), (4, 8), (2, 6), (2, 10), (6, 10)

Probability = $\frac{6}{55}$

59. For three events $A$, $B$ and $C$, $P$ (Exactly one of $A$ or $B$ occurs) = $P$(Exactly one of $B$ or $C$ occurs)

= $P$ (Exactly one of $C$ or $A$ occurs) = $\frac{1}{4}$ and

$P$(All the three events occur simultaneously) = $\frac{1}{16}$.

Then the probability that at least one of the events occurs, is

(1) $\frac{3}{16}$ (2) $\frac{7}{32}$

(3) $\frac{7}{16}$ (4) $\frac{7}{64}$

**Answer (3)**

**Sol.** $P(A) + P(B) - P(A \cap B) = \frac{1}{4}$

$P(B) + P(C) - P(B \cap C) = \frac{1}{4}$

$P(C) + P(A) - P(A \cap C) = \frac{1}{4}$

$$P(A) + P(B) + P(C) - P(A \cap B) - P(B \cap C) - P(A \cap C) = \frac{3}{8}$$

$\because \ P(A \cap B \cap C) = \frac{1}{16}$

$\therefore \ P(A \cup B \cup C) = \frac{3}{8} + \frac{1}{16} = \frac{7}{16}$

60. If $A = \begin{bmatrix} 2 & -3 \\ -4 & 1 \end{bmatrix}$, then adj $(3A^2 + 12A)$ is equal to

(1) $\begin{bmatrix} 72 & -63 \\ -84 & 51 \end{bmatrix}$ (2) $\begin{bmatrix} 72 & -84 \\ -63 & 51 \end{bmatrix}$

(3) $\begin{bmatrix} 51 & 63 \\ 84 & 72 \end{bmatrix}$ (4) $\begin{bmatrix} 51 & 84 \\ 63 & 72 \end{bmatrix}$

**Answer (3)**

**Sol.** $A = \begin{bmatrix} 2 & -3 \\ -4 & 1 \end{bmatrix}$

$$|A - \lambda I| = \begin{vmatrix} 2-\lambda & -3 \\ -4 & 1-\lambda \end{vmatrix}$$

$= (2 - 2\lambda - \lambda + \lambda^2) - 12$

$f(\lambda) = \lambda^2 - 3\lambda - 10$

$\because A$ satisfies $f(\lambda)$

$\therefore \ A^2 - 3A - 10I = 0$

$A^2 - 3A = 10I$

$3A^2 - 9A = 30I$

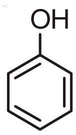
$3A^2 + 12A = 30I + 21A$

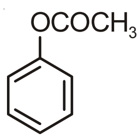
$$= \begin{bmatrix} 30 & 0 \\ 0 & 30 \end{bmatrix} + \begin{bmatrix} 42 & -63 \\ -84 & 21 \end{bmatrix}$$

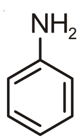
$$= \begin{bmatrix} 72 & -63 \\ -84 & 51 \end{bmatrix}$$

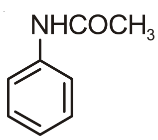
$$\text{adj}(3A^2 + 12A) = \begin{bmatrix} 51 & 63 \\ 84 & 72 \end{bmatrix}$$

61. Which of the following compounds will form significant amount of *meta* product during mono-nitration reaction?

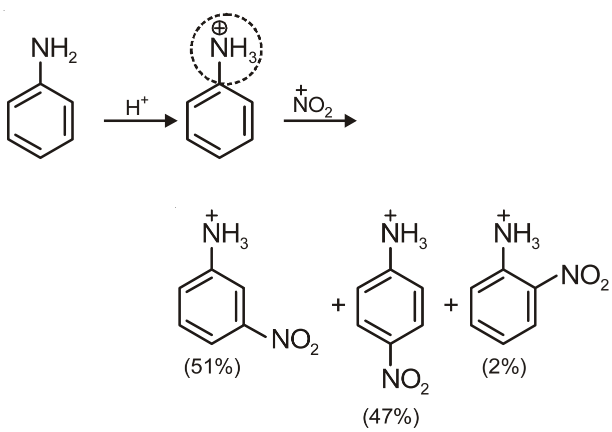
(1) 


(2) 


(3) 


(4) 


**Answer (3)**

**Sol.** 


62. $\Delta U$ is equal to

(1) Isochoric work

(2) Isobaric work

(3) Adiabatic work

(4) Isothermal work

**Answer (3)**

**Sol.** For adiabatic process, q = 0

$\therefore$ As per 1st law of thermodynamics,

$\Delta U = W$

63. The increasing order of the reactivity of the following halides for the $S_N1$ reaction is

I. $CH_3\underset{\underset{Cl}{|}}{C}HCH_2CH_3$

II. $CH_3CH_2CH_2Cl$

III. $p$–$H_3CO - C_6H_4 - CH_2Cl$

(1) (III) < (II) < (I)  (2) (II) < (I) < (III)

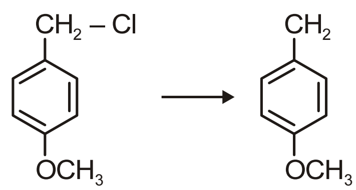
(3) (I) < (III) < (II)  (4) (II) < (III) < (I)

**Answer (2)**

**Sol.** Rate of $S_N1$ reaction $\propto$ stability of carbocation

I. $CH_3 - \underset{\underset{Cl}{|}}{C}H - CH_2 - CH_3 \longrightarrow CH_3 - \overset{+}{C}H - CH_2 - CH_3$

II. $CH_3 - CH_2 - CH_2 - Cl \longrightarrow CH_3 - CH_2 - \overset{+}{C}H_2$

III. 


So, II < I < III

Increase stability of carbocation and hence increase reactivity of halides.

64. The radius of the second Bohr orbit for hydrogen atom is

(Planck's Const. $h = 6.6262 \times 10^{-34}$ Js;

mass of electron $= 9.1091 \times 10^{-31}$ kg;

charge of electron $e = 1.60210 \times 10^{-19}$ C;

permittivity of vacuum

$\varepsilon_0 = 8.854185 \times 10^{-12}\ kg^{-1}\ m^{-3}\ A^2$)

(1) 1.65 Å

(2) 4.76 Å

(3) 0.529 Å

(4) 2.12 Å

**Answer (4)**

**Sol.** $r = a_0 \dfrac{n^2}{Z} = 0.529 \times 4$

$= 2.12$ Å

65. $pK_a$ of a weak acid (HA) and $pK_b$ of a weak base (BOH) are 3.2 and 3.4, respectively, The pH of their salt (AB) solution is

(1) 7.2

(2) 6.9

(3) 7.0

(4) 1.0

**Answer (2)**

**Sol.** $pH = 7 + \frac{1}{2}(pK_a - pK_b)$

$= 7 + \frac{1}{2}(3.2 - 3.4)$

$= 6.9$

66. The formation of which of the following polymers involves hydrolysis reaction?

(1) Nylon 6

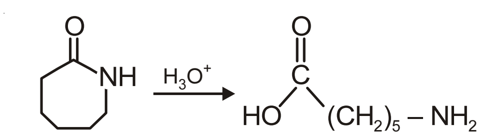
(2) Bakelite

(3) Nylon 6, 6

(4) Terylene

**Answer (1)**

**Sol.** Caprolactam is hydrolysed to produce caproic acid which undergoes condensation to produce Nylon-6.




(Caprolactam) (Caproic acid)

67. The most abundant elements by mass in the body of a healthy human adult are :

Oxygen (61.4%); Carbon (22.9%); Hydrogen (10.0%) and Nitrogen (2.6%).

The weight which a 75 kg person would gain if all $^1H$ atoms are replaced by $^2H$ atoms is

(1) 15 kg

(2) 37.5 kg

(3) 7.5 kg

(4) 10 kg

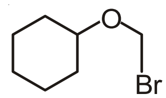
**Answer (3)**

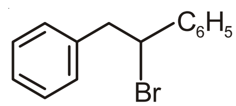
**Sol.** Mass of hydrogen = $\frac{10}{100} \times 75$ = 7.5 kg

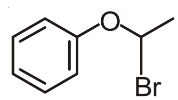
Replacing $^1H$ by $^2H$ would replace 7.5 kg with 15 kg

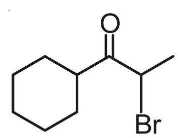
$\therefore$ Net gain = 7.5 kg

68. Which of the following, upon treatment with *tert*-BuONa followed by addition of bromine water, fails to decolourize the colour of bromine?

(1) 

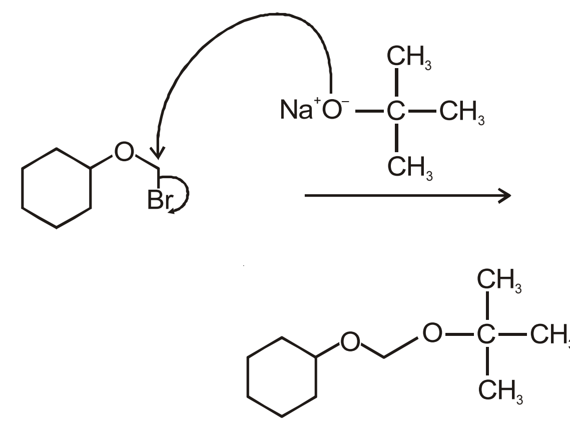

(2) 


(3) 


(4) 


**Answer (1)**

**Sol.**




(Product)

The above product does not have any C = C or C ≡ C bond, so, it will not give $Br_2$-water test.

69. In the following reactions, ZnO is respectively acting as a/an

(a) $ZnO + Na_2O \rightarrow Na_2ZnO_2$

(b) $ZnO + CO_2 \rightarrow ZnCO_3$

(1) Base and acid

(2) Base and base

(3) Acid and acid

(4) Acid and base

**Answer (4)**

**Sol.** In (a), ZnO acts as acidic oxide as $Na_2O$ is basic oxide.

In (b), ZnO acts as basic oxide as $CO_2$ is acidic oxide.

70. Both lithium and magnesium display several similar properties due to the diagonal relationship, however, the one which is incorrect, is

(1) Both form basic carbonates

(2) Both form soluble bicarbonates

(3) Both form nitrides

(4) Nitrates of both Li and Mg yield $NO_2$ and $O_2$ on heating

**Answer (1)**

**Sol.** Mg forms basic carbonate

$3MgCO_3 \cdot Mg(OH)_2 \cdot 3H_2O$ but no such basic carbonate is formed by Li.

71. 3-Methyl-pent-2-ene on reaction with HBr in presence of peroxide forms an addition product. The number of possible stereoisomers for the product is

(1) Six

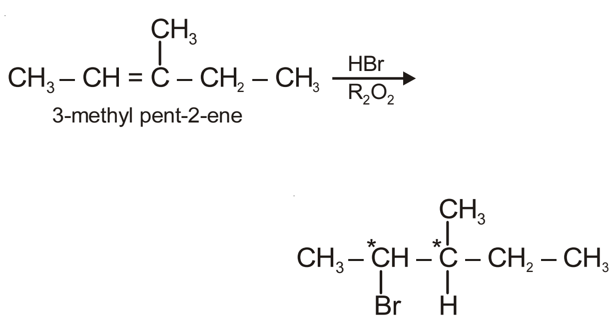
(2) Zero

(3) Two

(4) Four

**Answer (4)**

**Sol.**




Product (X)

Since product (X) contains two chiral centres and it is unsymmetrical.

So, its total stereoisomers = $2^2 = 4$.

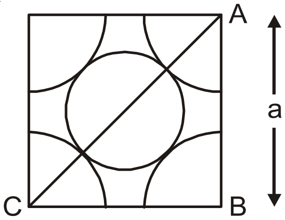
72. A metal crystallises in a face centred cubic structure. If the edge length of its unit cell is 'a', the closest approach between two atoms in metallic crystal will be

(1) 2a (2) $2\sqrt{2}a$

(3) $\sqrt{2}a$ (4) $\frac{a}{\sqrt{2}}$

**Answer (4)**

**Sol.** In FCC, one of the face is like




By $\Delta ABC$,

$$2a^2 = 16r^2$$

$$\Rightarrow r^2 = \frac{1}{8}a^2$$

$$\Rightarrow r = \frac{1}{2\sqrt{2}}a$$

Distance of closest approach = $2r = \frac{a}{\sqrt{2}}$

73. Two reactions $R_1$ and $R_2$ have identical pre-exponential factors. Activation energy of $R_1$ exceeds that of $R_2$ by 10 kJ $mol^{-1}$. If $k_1$ and $k_2$ are rate constants for reactions $R_1$ and $R_2$ respectively at 300 K, then $\ln(k_2/k_1)$ is equal to

($R = 8.314$ J $mole^{-1}$ $K^{-1}$)

(1) 8 (2) 12

(3) 6 (4) 4

**Answer (4)**

**Sol.** $k_1 = Ae^{-E_{a_1}/RT}$

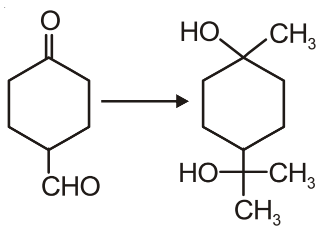
$$k_2 = Ae^{-E_{a_2}/RT}$$

$$\frac{k_2}{k_1} = e^{\frac{1}{RT}(E_{a_1} - E_{a_2})}$$

$$\ln\frac{k_2}{k_1} = \frac{E_{a_1} - E_{a_2}}{RT}$$

$$= \frac{10 \times 10^3}{8.314 \times 300} \approx 4$$

74. The correct sequence of reagents for the following conversion will be



(1) $[Ag(NH_3)_2]^+OH^-$, $H^+/CH_3OH$, $CH_3MgBr$

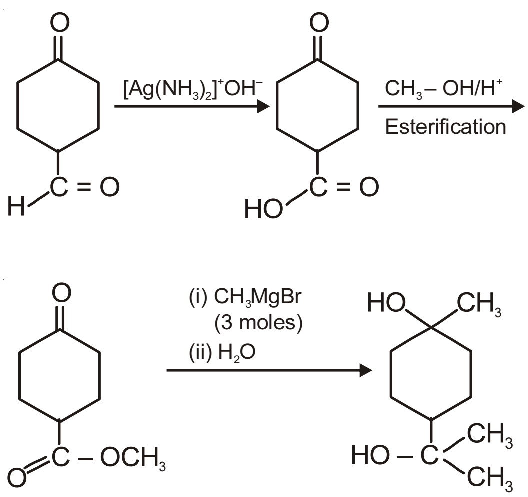
(2) $CH_3MgBr$, $H^+/CH_3OH$, $[Ag(NH_3)_2]^+OH^-$

(3) $CH_3MgBr$, $[Ag(NH_3)_2]^+OH^-$, $H^+/CH_3OH$

(4) $[Ag(NH_3)_2]^+OH^-$, $CH_3MgBr$, $H^+/CH_3OH$

**Answer (1)**

**Sol.**




75. The Tyndall effect is observed only when following conditions are satisfied

(a) The diameter of the dispersed particles is much smaller than the wavelength of the light used.

(b) The diameter of the dispersed particle is not much smaller than the wavelength of the light used

(c) The refractive indices of the dispersed phase and dispersion medium are almost similar in magnitude

(d) The refractive indices of the dispersed phase and dispersion medium differ greatly in magnitude

(1) (a) and (d)

(2) (b) and (d)

(3) (a) and (c)

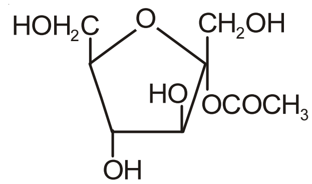
(4) (b) and (c)

**Answer (2)**

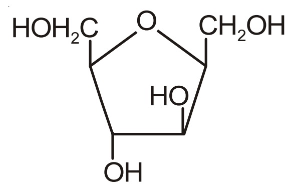
**Sol.** For Tyndall effect refractive index of dispersion phase and dispersion medium must differ significantly. Secondly, size of dispersed phase should not differ much from wavelength used.

76. Which of the following compounds will behave as a reducing sugar in an aqueous KOH solution?

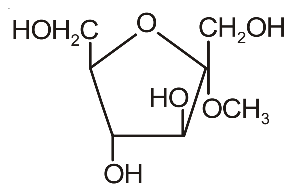
(1)



(2)



(3)



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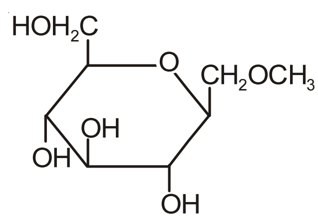
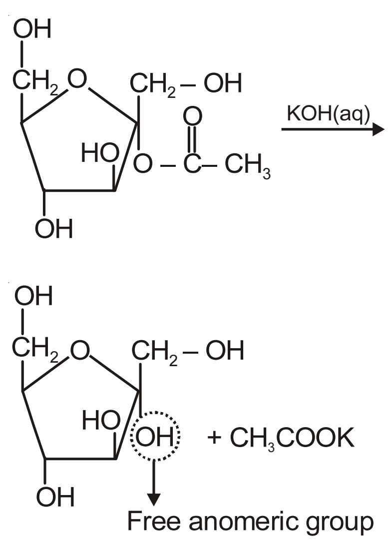


**Answer (1)**

**Sol.** Sugars in which there is free anomeric –OH group are reducing sugars



77. Given

$C_{(graphite)} + O_2(g) \rightarrow CO_2(g);$

$\Delta_r H^o = -393.5 \text{ kJ mol}^{-1}$

$H_2(g) + \frac{1}{2}O_2(g) \rightarrow H_2O(l);$

$\Delta_r H^o = -285.8 \text{ kJ mol}^{-1}$

$CO_2(g) + 2H_2O(l) \rightarrow CH_4(g) + 2O_2(g);$

$\Delta_r H^o = +890.3 \text{ kJ mol}^{-1}$

Based on the above thermochemical equations, the value of $\Delta_r H^o$ at 298 K for the reaction

$C_{(graphite)} + 2H_2(g) \rightarrow CH_4(g)$ will be

(1) $+74.8 \text{ kJ mol}^{-1}$

(2) $+144.0 \text{ kJ mol}^{-1}$

(3) $-74.8 \text{ kJ mol}^{-1}$

(4) $-144.0 \text{ kJ mol}^{-1}$

**Answer (3)**

**Sol.** $C_{(graphite)} + O_2(g) \rightarrow CO_2(g);$

$\Delta_r H^\circ = -393.5 \text{ kJ mol}^{-1}$ ...(i)

$H_2(g) + \frac{1}{2}O_2(g) \rightarrow H_2O(l);$

$\Delta_r H^\circ = -285.8 \text{ kJ mol}^{-1}$ ...(ii)

$CO_2(g) + 2H_2O(l) \rightarrow CH_4(g) + 2O_2(g);$

$\Delta_r H^\circ = 890.3 \text{ kJ mol}^{-1}$ ...(iii)

By applying the operation

(i) + 2 × (ii) + (iii), we get

$C_{(graphite)} + 2H_2(g) \rightarrow CH_4(g);$

$\Delta_r H^\circ = -393.5 - 285.8 \times 2 + 890.3$

$= -74.8 \text{ kJ mol}^{-1}$

78. Which of the following reactions is an example of a redox reaction?

(1) $XeF_4 + O_2F_2 \rightarrow XeF_6 + O_2$

(2) $XeF_2 + PF_5 \rightarrow [XeF]^+ PF_6^-$

(3) $XeF_6 + H_2O \rightarrow XeOF_4 + 2HF$

(4) $XeF_6 + 2H_2O \rightarrow XeO_2F_2 + 4HF$

**Answer (1)**

**Sol.** Xe is oxidised from +4(in $XeF_4$) to +6(in $XeF_6$)

Oxygen is reduced from +1 (in $O_2F_2$) to zero (in $O_2$)

79. The products obtained when chlorine gas reacts with cold and dilute aqueous NaOH are

(1) $ClO^-$ and $ClO_3^-$

(2) $ClO_2^-$ and $ClO_3^-$

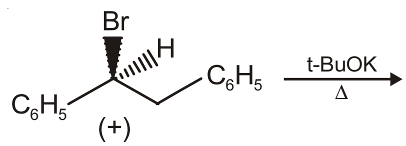
(3) $Cl^-$ and $ClO^-$

(4) $Cl^-$ and $ClO_2^-$

**Answer (3)**

**Sol.** $Cl_2 + \underset{\text{Cold \& dilute}}{2NaOH} \longrightarrow NaCl + \underset{\text{Sodium hypochlorite}}{NaOCl} + H_2O$

80. The major product obtained in the following reaction is




(1) $(\pm)C_6H_5CH(O^tBu)CH_2C_6H_5$

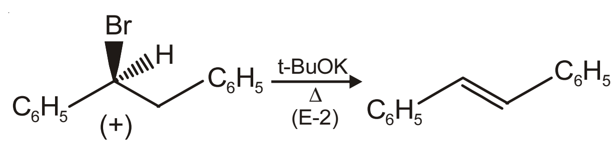
(2) $C_6H_5CH = CHC_6H_5$

(3) $(+)C_6H_5CH(O^tBu)CH_2C_6H_5$

(4) $(-)C_6H_5CH(O^tBu)CH_2C_6H_5$

**Answer (2)**

**Sol.**




81. Sodium salt of an organic acid 'X' produces effervescence with conc. $H_2SO_4$. 'X' reacts with the acidified aqueous $CaCl_2$ solution to give a white precipitate which decolourises acidic solution of $KMnO_4$. 'X' is

(1) $C_6H_5COONa$

(2) $HCOONa$

(3) $CH_3COONa$

(4) $Na_2C_2O_4$

**Answer (4)**

**Sol.** $\underset{(X)}{Na_2C_2O_4} + \underset{\text{Conc.}}{H_2SO_4} \longrightarrow Na_2SO_4 + \underset{\text{oxalic acid}}{H_2C_2O_4}$

$H_2C_2O_4 \xrightarrow[\Delta, \ -H_2O]{\text{Conc. } H_2SO_4} \underbrace{CO\uparrow + CO_2\uparrow}_{\text{effervescence}}$

$\underset{(X)}{Na_2C_2O_4} + CaCl_2 \longrightarrow \underset{\text{white ppt.}}{CaC_2O_4\downarrow} + 2NaCl$

$2MnO_4^- + 5C_2O_4^{2-} + 16H^+ \rightarrow 2Mn^{2+} + 10CO_2 + 8H_2O$

82. Which of the following species is not paramagnetic?

(1) NO

(2) CO

(3) $O_2$

(4) $B_2$

**Answer (2)**

**Sol.** CO has 14 electrons (even) $\therefore$ it is diamagnetic

NO has $15e^-$(odd) $\therefore$ it is paramagnetic and has 1 unpaired electron in $\pi^*2p$ molecular orbital.

$B_2$ has $10e^-$ (even) but still paramagnetic and has two unpaired electrons in $\pi 2p_x$ and $\pi 2p_y$ (s-p mixing).

$O_2$ has 16 $e^-$ (even) but still paramagnetic and has two unpaired electrons in $\pi^*2p_x$ and $\pi^*2p_y$ molecular orbitals.

83. The freezing point of benzene decreases by 0.45ºC when 0.2 g of acetic acid is added to 20 g of benzene. If acetic acid associates to form a dimer in benzene, percentage association of acetic acid in benzene will be

($K_f$ for benzene = 5.12 K kg $mol^{-1}$)

(1) 64.6% (2) 80.4%

(3) 74.6% (4) 94.6%

**Answer (4)**

**Sol.** $0.45 = i(5.12)\dfrac{0.2/60}{20}\times 1000$

$\Rightarrow$ i = 0.527

$$\underset{1-\alpha}{2CH_3COOH} \rightleftharpoons \underset{\frac{\alpha}{2}}{(CH_3COOH)_2}$$

$\Rightarrow \ i = 1-\dfrac{\alpha}{2}$

$\Rightarrow \ 0.527 = 1-\dfrac{\alpha}{2}$

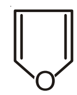
$\Rightarrow \ \dfrac{\alpha}{2} = 0.473$

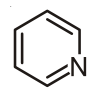
$\Rightarrow \ \alpha = 0.946$

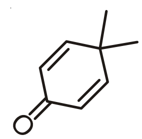
$\therefore$ % association = 94.6%

84. Which of the following molecules is least resonance stabilized?

(1) 

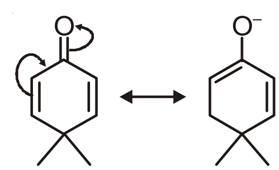
(2) 

(3) 

(4) 

**Answer (4)**

**Sol.** However, all molecules given in options are stabilised by resonance but compound given in option (4) is least resonance stabilised (other three are aromatic)



85. On treatment of 100 mL of 0.1 M solution of $CoCl_3 \cdot 6H_2O$ with excess $AgNO_3$; $1.2 \times 10^{22}$ ions are precipitated. The complex is

(1) $[Co(H_2O)_4Cl_2]Cl \cdot 2H_2O$

(2) $[Co(H_2O)_3Cl_3] \cdot 3H_2O$

(3) $[Co(H_2O)_6]Cl_3$

(4) $[Co(H_2O)_5Cl]Cl_2 \cdot H_2O$

**Answer (4)**

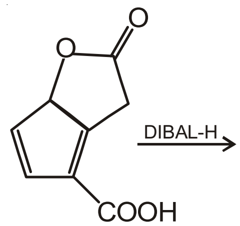
**Sol.** Millimoles of $AgNO_3$ = $\dfrac{1.2\times 10^{22}}{6\times 10^{23}}\times 1000$ = 20

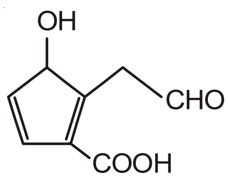
Millimoles of $CoCl_3{\cdot}6H_2O$ = 0.1 × 100 = 10

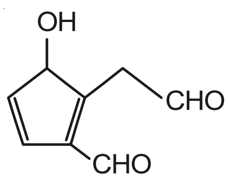
$\therefore$ Each mole of $CoCl_3{\cdot}6H_2O$ gives two chloride ions.

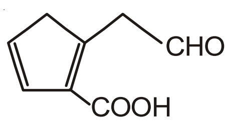
$\therefore$ $[Co(H_2O)_5Cl]Cl_2{\cdot}H_2O$

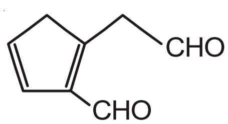
86. The major product obtained in the following reaction is




(1) 

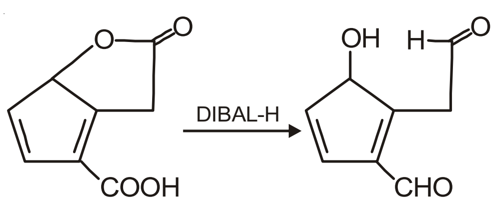

(2) 


(3) 


(4) 


**Answer (2)**

**Sol.** DIBAL — H reduces esters and carboxylic acids into aldehydes




87. A water sample has ppm level concentration of following anions

$F^- = 10$; $SO_4^{2-} = 100$; $NO_3^- = 50$

The anion/anions that make/makes the water sample unsuitable for drinking is/are

(1) Only $NO_3^-$

(2) Both $SO_4^{2-}$ and $NO_3^-$

(3) Only $F^-$

(4) Only $SO_4^{2-}$

**Answer (3)**

**Sol.** Permissible limit of $F^-$ in drinking water is upto 1 ppm. Excess concentration of $F^-$ > 10 ppm causes decay of bones.

88. 1 gram of a carbonate ($M_2CO_3$) on treatment with excess HCl produces 0.01186 mole of $CO_2$. The molar mass of $M_2CO_3$ in g mol$^{-1}$ is

(1) 1186 (2) 84.3

(3) 118.6 (4) 11.86

**Answer (2)**

**Sol.** $M_2CO_3 + 2HCl \rightarrow 2MCl + H_2O + CO_2$

$$n_{M_2CO_3} = n_{CO_2}$$

$$\frac{1}{M_{M_2CO_3}} = 0.01186$$

$$M_{M_2CO_3} = \frac{1}{0.01186}$$

$$= 84.3 \text{ g/mol}$$

89. Given

$$E^\circ_{Cl_2/Cl^-} = 1.36 \text{ V}, E^\circ_{Cr^{3+}/Cr} = -0.74 \text{ V}$$

$$E^\circ_{Cr_2O_7^{2-}/Cr^{3+}} = 1.33 \text{ V}, E^\circ_{MnO_4^-/Mn^{2+}} = 1.51 \text{ V}$$

Among the following, the strongest reducing agent is

(1) Cr (2) $Mn^{2+}$

(3) $Cr^{3+}$ (4) $Cl^-$

**Answer (1)**

**Sol.** For $Cr^{3+}$, $E^\circ_{Cr^{3+}/Cr_2O_7^{2-}} = -1.33$ V

For $Cl^-$, $E^\circ_{Cl^-/Cl_2} = -1.36$ V

For Cr, $E^\circ_{Cr/Cr^{3+}} = 0.74$ V

For $Mn^{2+}$, $E^\circ_{Mn^{2+}/MnO_4^-} = -1.51$ V

Positive E° is for Cr, hence it is strongest reducing agent.

90. The group having isoelectronic species is

(1) $O^{2-}$, $F^-$, $Na^+$, $Mg^{2+}$

(2) $O^-$, $F^-$, Na, $Mg^+$

(3) $O^{2-}$, $F^-$, Na, $Mg^{2+}$

(4) $O^-$, $F^-$, $Na^+$, $Mg^{2+}$

**Answer (1)**

**Sol.** $Mg^{2+}$, $Na^+$, $O^{2-}$ and $F^-$ all have 10 electrons each.

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