Time: 3Hours





JEE MAIN SAMPLE PAPER

CLASS:12th (PCM)

Maximum Marks: 360

Topics Covered:

Physics : Full Syllabus

Chemistry: Full Syllabus

Mathematics : Full Syllabus

Important Instruction:

- 1. Attempting all the questions are compulsory.
- 2. Use Blue / Black Ball point pen only.
- 3. There are three sections of equal weightage in the question paper A, B, C (**Physics, Chemistry and Mathematics**) Having 30 questions each.
- 4. For marking scheme, +4 marks for each correct answer and -1 marks for each incorrect answer.
- 5. Use of calculator and other electronic devices is not allowed during the exam.
- 6. No extra sheets will be provided for any kind of work.

Name of the Student:	. Class:
Father's Name:	Signature:
Branch Name:	. Contact No:
Didirent Name.	Contact No.

			_	
1. If pressure P, v	velocity V and time T are t	aken as fundamental p	ohysical quantities, then	the dimensional
(a) PV^2T^2	(b) $P^{-1}V^2T^{-2}$	(c) PVT^2	(d) $P^{-1}VT^2$	
2. A projectile is	given an initial velocity of	$(\hat{i}+2\hat{j})m/s$, where \hat{i}	is along the ground and	\hat{j} is along the
vertical. If $g = 10$	m/s^2 , the equation of its	trajectory is:		

(c) $4y = 2x - 5x^2$ (d) $4y = 2x - 25x^2$ (a) $y = x - 5x^2$ (b) $y = 2x - 5x^2$ 3. A boy runs along a straight path for the first half of the distance with a velocity v_1 and the second half of the

(a) $\frac{2}{V} = \frac{1}{v_1} + \frac{1}{v_2}$

(b) $V = \frac{v_1 + v_2}{2}$

distance with a velocity v_2 . The average velocity V is given by

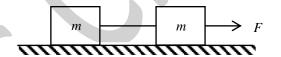
(c) $V = \sqrt{v_1 v_2}$

(d) $\vec{V} = \vec{v}_1 + \vec{v}_2$

4. Two blocks of equal mass are connected by a light string and placed on a smooth horizontal surface. If a force F acts on one of the block then the tension in the string is

(a) F

(c) 2F



5. A cricket player catches a ball of mass 0.1 kg, moving with a speed of $10ms^{-1}$ in 0.1s. Magnitude of the force exerted by him is (N)

(a) 10

(b) 4

(c) 2

(d) 1

6. A spring for spring constant $240 \, Nm^{-1}$ is compressed by $10 \, cm$ whereas another similar spring is extended by 10~cm. The difference of the stored potential energies of two springs is

(a) zero

(b) 4 J

(c) 1.2 J

(d) 12 J

7. For an electron in the nth Bohr orbit of hydrogen atom, what will be the ratio of radius of orbit to its de-Broglie wavelength

 $(a)\frac{n}{2\pi}$

(c) $\frac{1}{2\pi n}$ (d) $\frac{1}{2\pi n^2}$

8. If in a plano-convex lens radius of curvature of convex surface is 10 cm and the focal length of the lens is 30 cm, the refractive index of the material of the lens will be:

(a) 1.5

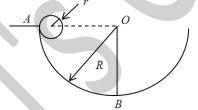
(b) 1.66

(c)1.33

(d) 3

- 9. A body is projected at an angle θ to the horizontal with kinetic energy E_k . The potential energy of the body at the highest point of the trajectory is
- (a) E_k
- (b) $E_{\nu} \cos^2 \theta$
- (c) $E_{\nu} \sin^2 \theta$
- (d) $E_{\nu} \tan^2 \theta$
- 10. One gram of ice is mixed with one gram of steam. After thermal equilibrium, the temperature of the
- (a) $0^{\circ}C$
- (b) 100°C
- (c) $55^{\circ}C$
- (d) 80°C
- 11. A ray of light travelling in the direction $\frac{1}{2}(\hat{i}+\sqrt{3}\hat{j})$ is incident on a plane mirror in the x-z plane.
- After reflection, it travels along the direction $\frac{1}{2}(\hat{i}-\sqrt{3}\hat{j})$. The angle of incidence is
- (a) 30°
- (b) 45°
- (c) 60°
- (d) 75°
- 12. A radioactive nucleus (initial mass number A and atomic number Z) emits 3α particles and 2 positrons. The ratio of number of neutrons to that of protons in the final nucleus will be
- (a) $\frac{A-Z-8}{Z-4}$

- (b) $\frac{A-Z-4}{Z-8}$ (c) $\frac{A-Z-12}{Z-4}$ (d) $\frac{A-Z-4}{Z-2}$
- 13. A cylinder of mass m and radius r rolls down a circular track from point A as shown in the figure. Assume that the friction is just sufficient to support the rolling. Velocity of the cylinder at point A was zero. Assume $r \ll R$. The reaction by the track on the cylinder at point B is
- (a) $\frac{7}{3} mg$
- (c) $\frac{5}{3}mg$ (d) $\frac{2}{3}mg$



- 14. A charged particle of a mass m and charge q is released from rest in a uniform electric field E neglecting the effect of gravity, the kinetic energy of the charged particle after t second is
- (a)
- $(c)\frac{2E^2t^2}{mg}$
- (d) $\frac{Eq^2m}{2t^2}$
- 15. A police car moving at 22 m/s, chases a motorcyclist. The police man sounds his horn at 176 Hz, while both of them move towards a stationary siren of frequency 165 Hz. Calculate the speed of the motorcycle, if it is given that motor cyclist does not observes any beats. (velocity of sound in air = 330m/s)
- (a) 33 m/s
- (b) 22 m/s
- (c) 55 m/s
- (d) 11 m/s







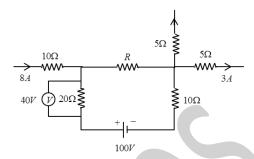
16. In the circuit shown, the voltmeter reading is 40 V. Find the value of unknown resistor $\it R$



(b) 8Ω



(d) 10Ω



17. In young's double slit experiment, the intensity at a point where the path difference is $\frac{\lambda}{6}(\lambda)$ being the wavelength of light used) is I. If I_0 denotes the maximum intensity, $\frac{I}{I_0}$ is equal to

(a) $\frac{3}{4}$

- (b) $\frac{1}{\sqrt{2}}$
- (c) $\frac{\sqrt{3}}{2}$
- (d) $\frac{1}{2}$

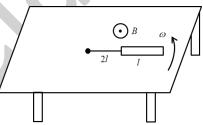
18. A metallic rod of length l is tied to a non-conducting string of length 2l and made to rotate with angular speed ω on a horizontal table with one end of the string fixed. If there is a vertical magnetic field B in the region, the *emf* induced across the ends of the rod is



(b)
$$\frac{3B\omega L^2}{2}$$

(c)
$$\frac{5B\omega l^2}{2}$$

(d)
$$\frac{4B\omega L^2}{2}$$



19. A wheel is rotating about a fixed axis through its centre 300 rpm. A constant torque starts acting on it opposes its motion. Before coming to rest it makes 25 complete rotations. If the moment of inertia of the wheel about the axis of rotation is $\left(\frac{10}{\pi}\right) Kg \, m^2$, the torque (in N-m) acting on it is

- (a) 10
- (b) 15

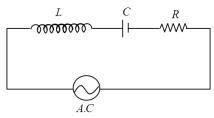
- (c) 20
- (d) 25

20. An inductance of $\frac{200}{\pi}mH$, a capacitance of $\frac{10^{-3}}{\pi}F$ and a resistance of 10Ω are connected in series with an a.c. source 220V, 50Hz. The phase angle of the circuit for the current and voltage source is

- (a) $\frac{\pi}{2}$
- (b) $\frac{\pi}{3}$

(c) $\frac{\pi}{6}$

(d) $\frac{\pi}{4}$

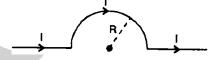


- 21. If an electron and a proton having same momenta enter perpendicular to a magnetic field, then
- (a) the length of curved path of electron and proton will be same
- (b) they will move undeflected
- (c) the length of curved path of electron is more curved than that of the proton
- (d) the length of curved path of proton is more curved than that of the electron
- 22. A heavy uniform chain lies on a horizontal table top. If the coefficient of friction between the chain and the table surface is 0.25, then the maximum fraction of the length of the chain that can hang over one edge of the table is
- (a) 20%

- (b) 25%
- (c) 35%
- (d) 15%
- 23. A cylindrical tube, open at both ends, has a fundamental frequency f_0 , in air. The tube is dipped vertically into water such that half of its length is inside water. The fundamental frequency of the air column now is
- (a) $\frac{3f_0}{4}$

- (b) f_0
- $(c)\frac{f_0}{2}$

- (d) $3f_0$
- 24. An infinite long straight wire is bent into a semicircle of radius R, as shown in the figure. A current I is sent through the conductor. The magnetic field at the centre of the semicircle is:
- (a) infinite
- (b) zero
- (c) $\frac{\mu_0 \pi I}{4\pi R}$
- (d) $\frac{\mu_0}{4\pi} \frac{I}{R} (\pi + 1)$



- 25. Choose the correct relation regarding potential of an electric dipole. Here A, B, C and D all are at equal distance from point O which is much larger than the dimensions of the dipole. Then
- (a) $|V_A| = |V_B| > |V_C| = |V_D|$
- (b) $|V_C| = |V_D| > |V_A| = |V_B|$

A -q O +q B

- (c) $|V_A| > |V_C| = |V_D| > |V_B|$
- (d) $|V_C| > |V_B| = |V_D| > |V_A|$
- •D $\sin\left(100t + \frac{\pi}{2}\right)A$ Then the

26. In an A.C. circuit, V and I are given by $V = 100\sin(100t)volt$, $I = 100\sin(100t + \frac{\pi}{3})A$ Then the power dissipated in the circuit is

- (a) $10^4 W$
- (b) 10W
- (c) 2500W
- (d) 5W
- 27. The magnetic flux ϕ (in weber) linked with a coil of resistance 10Ω varies with time t (in second) as $\phi = 8t^2 4t + 1$. The current induced in the coil at t = 0.1 sec is
- (a) 10A
- (b) 0.24A
- (c) 0.12A
- (d) 4.8A

28. If R,C and L denote resistance, capacitance and inductance. Which of the following will NOT have the dimensions of frequency?

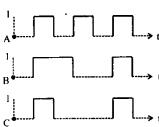
(a) RL^{-1}

- (b) $R^{1}C^{1}$
- (c) $L^{-1/2}C^{-1/2}$
- (d) RCL

29. The following figure shows a logic gate circuit with two inputs A and B and the output C. The voltage wavefront of A, B and C are as shown below

The logic circuit gate is

- (a) NAND gate
- (b) NOR gate
- (c) OR gate
- (d) AND gate



circuit

30. If N_0 is the original mass of the substance of half-life period $t_{1/2} = 5$ years, then the amount of substance left after 15 years is

- (a) $\frac{N_0}{8}$
- (b) $\frac{N_0}{16}$
- (c) $\frac{N_0}{2}$
- (d) $\frac{N_0}{4}$

PART-B CHEMISTRY

31. The IUPAC name of the following compound is:

$$\begin{array}{c} Cl\\ |\\ CH_3-C-CH_2-CH=CH-CH_3\\ |\\ H \end{array}$$

(a) 2-chlorohex-5-ene

- (b) 5-chlorohex-2-ene
- (c) 1-chloro-1-methylpent-3-ene
- (d) 5-chloro-5-methylpent-2-ene

32. Formation of polyethylene from calcium carbide takes place as follows:

$$CaC_2 + 2H_2O \rightarrow Ca(OH)_2 + C_2H_2$$

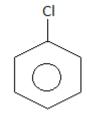
 $C_2H_2 + H_2 \rightarrow C_2H_4$
 $nC_2H_4 \rightarrow (-CH_2 - CH_2 -)_n$

the amount of polyethylene obtained from 64 kg of CaC₂ is

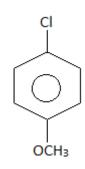
- (a) 7 kg
- (b) 14 kg
- (c) 21 kg
- (d) 28 kg

33. Arrange the following compounds in order of increasing reactivity towards nucleophilic substitution.

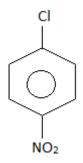
(i)



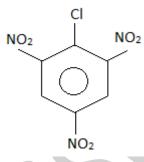
(ii)



(iii)



(iv)



34. A +
$$SOCl_2 \rightarrow B + SO_2 + HCl$$

$$X + Na \rightarrow C + H_2$$

$$B + C \rightarrow (C_2H_5)_2O + NaCl$$

Then A and X are respectively

- (a) C₂H₅ONa and C₂H₅Cl
- (b) C₂H₅Cl and C₂H₅ONa
- (c) C₂H₅OH and C₂H₅OH
- (d) C₂H₅OH and C₂H₅ONa

35. Natalite is a mixture of

- (a) $CH_3OH + C_2H_5OH$
- (b) $C_2H_5OH + (C_2H_5)_2O$
- (c) $CH_3CHO + (C_2H_5)_2O$
- (d) $CH_3COCH_3 + (C_2H_5)_2O$

36.
$$CH_3CHO + NH_2OH \rightarrow X \xrightarrow{-H_2O} Y$$

The number of σ bonds, π bonds and lone pairs of electrons in the compounds 'y' are respectively

- (a) 9, 1, 4
- (b) 11, 1, 5
- (c) 9, 2, 2
- (d) 8, 1, 3

37. Order of decreasing acidity of

- (I) HCOOH
- (II) CH₃COOH
- (III) Cl₂CHCOOH
- (IV) CF₃COOH is

- (a) IV > III > II > I
- (b) IV > III > I > II
- (c) |I| > |V| > |I| > |I|
- (d) I > IV > II > III

38. What are A, B, C in the following reaction?

I.
$$(CH_3COO)_2Ca$$

II.
$$CH_3CO_2H \xrightarrow{HI} B$$
 Red P

III.
$$2CH_3CO_2H \stackrel{P_4O_{10}}{-} C$$

(a)
$$\frac{A}{C_2H_6} \frac{B}{CH_3COCH_3} \frac{C}{(CH_3CO)_2O}$$

- (b) $(CH_3CO)_2O$
- CH₃COCH₃
- (c) CH₃COCH₃ C₂H₆
- $(CH_3CO)_2O$
- (d) $CH_3COCH_3(CH_3CO)_2O$ C_2H_6

(i) N, N-dimethyl anilir	lamines test is given by ne (ii) 2 yl aniline (iv) p (b) (ii) and (iv)	, 4-dimethyl aniline	(d) (i) and (iv)
	sec^{-1} ,then the initial c	$_5 \rightarrow 4NO_2 + O_2 \text{ is } 3.0 \times 0$ concentration of N_2O_5 (in (d) 0.8	
41. Sodium has a bcc s mass of sodium = 23)	structure with nearest i	neighbour distance of 36	55.9 pm. Calculate its density (atomic
· · · · · · · · · · · · · · · · · · ·	(b) $2.65 \text{ g/}cm^3$	(c) $3.3 \text{ g/}cm^3$	(d) 6.2 g/ cm^3
		·	sure is 30 Kcal/mole. What will be the
(a)13 Kcal	ergy of 3 mol of liquid a (b) -13 Kcal	(c) 27Kcal	(d) -27 Kcal
43. For the reaction $3Br_2$ -	$+60H^{\Theta} \rightarrow 5Br^{\Theta} +$	$BrO_3^{\ominus} + 3H_2O$	
Equivalent we (a) $\frac{M}{2}$	ight of $Br_2(Molecular$ (b) $\frac{M}{10}$	weight M) is $(c)\left(\frac{M}{2} + \frac{M}{10}\right)$	(d) $\frac{M}{6}$
44. The specific condu is 6×10^{-8} ohm ⁻¹ cm ⁻¹ . T Given $\Lambda^0_{Agcl} = 137.2$ oh (a)1.7×10 ⁻³ M	he Solubility of Agcl is	lution of Agcl is found to (c)1.3×10 ⁻⁴ M	be 1.86×10^{-6} ohm ⁻¹ cm ⁻¹ and that of water (d) 1.3×10^{-6} M
dissociation of SO_2Cl_2i (a) 33 (b) 35	S	(d) 66	$_{2}\mathrm{Cl}_{2(\mathrm{g})} ightleftharpoons \mathrm{SO}_{2(\mathrm{g})} + \mathrm{Cl}_{2(\mathrm{g})}$ is 50. The percentage
46. The ionisation ene (a)13.6 ev	rgy of hydrogen atom i (b) 54.4 ev	s 13.6 ev. What will be t (c) 122.4 ev	he ionisation energy of Li ⁺² ions? (d)48.6 ev
		· —	0×10^{-5} and 5.0×10^{-10} respectively. The
Overall dissociation (a) 0.2×10^5	constant of the acid with (b) 5.0×10 ⁻⁵	(c) 5.0×10 ⁻¹⁵	(d) 5.0×10 ¹⁵
48. The density of Oxy (a) 1.4	gen is 1.43 g/L at STP. (b) 1.69	The density of oxygen at (c) 1.8	: 17 ^o c and 800 torr is (d) 2.5

49. Match the List-I with List-II by using the postulates of VBT of complexes

List-I	List-II
$(P)[Ni(\mathit{CN})_4]^{2-}$	(1) sp^3 hybridization
$(Q)[Ni(CO)_4]$	(2) dsp^2 hybridization
(R) $[Cu(NH_3)_4]^{2+}$	$(3) \mu = 0 BM$
(S) $[Pd(CI)_4]^{2-}$	(4) $\mu = 1.732$ BM
(a) P-3,1 Q-1,4 R-2,3 S-1,4	(b) P-2,3 Q-1,3 R-1,4 S-2,3
(c)p-2,4 Q-1,4 R-2,4 S-2	(d) P-2,3 Q-1,3 R-2,4 S-2,3

50. The incorrect order regarding 15 th group hydrides is

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\begin{array}{ll} \text{(a) Reducing nature} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(b) Bond angle} & : \text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3 \\ \text{(c) Basic nature} & : \text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{BiH}_3 \\ \text{(d) Boiling point} & : \text{NH}_3 < \text{PH}_3 < \text
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51. Which of the among the following is more acidic?

- (a) H_2O (b) H_2S (c) H_2Se (d) H_2Te
- 52. The true statement for the acids of phosphorus : H_3PO_2 , H_3PO_3 and H_3PO_4 is
- (a) The order of acidity is $H_3PO_2 < H_3PO_3 < H_3PO_4$
- (b) All of these are reducing in nature
- (c) All are tribasic acids
- (d) The geometry of phosphorus is tetrahedral in all the three

53. The salt used for performing 'bead' test in qualitative inorganic analysis is:

- (a) K_2SO_4 . $AI_2(SO_4)_3$. $24H_2O$
- (b) FeSO₄. (NH₄)₂SO₄. 6H₂O
- (c) Na (NH₄)HPO₄. 4H₂O
- (d) Na_2SO_4 . $Cr_2(SO_4)_3$. $24H_2O$

54. Which are true statements among the following?

- (A) PH₅ and BiCl₅ do not exist
- (B) Bond order of CO_3^{-2} is 1.33
- (C) SeF₄ and CH₄have same shape
- (D) I₃⁺has bent geometry
- (E) The number of $P_{\pi}-d_{\pi}$ bonds in $SO_2\&SO_3$ are same
- (a) A&C
- (b) A, B&E
- (c) A,B&D
- (d) all the above

55. The increasing order of acidic nature of the following oxidies is

- (a) $SiO_2 < P_2O_5 < CI_2O_7 < SO_3$
- (b) $SiO_2 < P_2O_5 < SO_3 < CI_2O_7$
- (c) $CI_2O_7 < SO_3 < P_2O_5 < SiO_2$
- (d) $SO_3 < CI_2O_7 < SiO_2 < P_2O_5$

56. An orange coloured solution acidified with H₂SO₄ and treated with a substance'X' gives a blue coloured

solutio (a) H ₂ (ne substance 'X'is : dil HCl		$H_{2}O_{2}$	(d) Conc. HC	ïl	
		T/GC ratio of is 0.9 Janine present is	3. If t	the number of mo	oles of adenine in	n the DNA	sample is 465,000,
(a) 465	,000	(b) 535,000		(c) 50,000	(d) 9	93,000	
58. An (a) Eth		condensation hon (b) Melamine-		ymer is aldehyde resin	(c) Alkyd res	in ((d) Perlon or Nylon-
59.	Г						V ,-
		Column I		Colu	mn II		
	(A) Anta (B) Anti (C) Traq (D) Anti	biotic Juillizer		(p) Equanil(q) BHT(r) Bromphenira(s) Aminoglycos		3	
	; B-q; C-s; D-r B-s; C-p; D-q			r; C-r; D-s C-q; D-p			
60. The	_	ntially caused by μ (b) O_3 and N_2	orese			rogen	(d) O ₂ and N ₂
	f the areas of	uare and M, N are the square and th 3:3 (c) 2:	e tria		the sides PQ and	QR respe	ctively, then the
	_	the point P on the nen length of PQ =		$e^{x^2 + y^2 + 6x + 6y}$	v = 2 meets the li	ne 5x – 2y	v + 6 = 0 at a point
(a) 4	(b) 5			(d) $4\sqrt{5}$			
63. The (a) $\frac{\pi}{2}$	e angle betwe	en the tangents do $\frac{\pi}{3}$ (c) $\frac{\pi}{4}$	rawn	from the point (1 (d) $\frac{\pi}{6}$., 4) to the parab	ola y ² = 4	x is
64. Me	an of 100 iter	ns is 49. It was dis	cover	ed that three iter	ms which should	have bee	n 60, 70, 80 were

wrongly read as 40, 20, 50 respectively. Correct mean will be

(b)
$$82\frac{1}{2}$$
(c) 50

65. Angle between asymptotes of the hyperbola $3x^2 - y^2 = 3$ is

(a)
$$\frac{\pi}{3}$$

(b)
$$\frac{2\pi}{3}$$

$$(c)\frac{\pi}{6}$$

(d)
$$\frac{3\pi}{4}$$

66. P \rightarrow (q V r) is false, then the true valves of p, q, r respectively are

$$67. Lt_{x\to 0} \frac{1-\cos^3 x}{\sin 3x \sin 5x} =$$

(a)
$$\frac{1}{15}$$

(b)
$$\frac{2}{15}$$

$$(c)\frac{1}{30}$$

(d)
$$\frac{1}{10}$$

68.
$$\sin^{-1} x + \sin^{-1} y = \pi/2$$
 then $\frac{dy}{dx} =$

$$(a) - x/y$$

$$(b) - y/x$$

(d)
$$y/x$$

69. The tangent at A (2, 4) on $y = x^3 - 2x^2 + 4$ cuts the x axis at T then AT =

(a)
$$4\sqrt{17}$$

(b)
$$\sqrt{17}/4$$

(c)
$$\sqrt{17}$$

70. The value of 'a' for which the function $f(x) = a \sin x + \frac{1}{3} \sin 3x$ has an extremum at $x = \pi/3$ is

(b)
$$-2$$

$$(d) - 2/3$$

71. The quadratic equation $3ax^2 + 2bx + c = 0$ has at least one root between 0 and 1 if

(a)
$$a + b + c = 1$$

(b)
$$a + b + c = 0$$

(c)
$$3a + 2ab + c = 0$$

(d)
$$6a + 2b = 0$$

72.
$$\int e^{x} \left(\frac{x^2 + 5x + 7}{(x+3)^2} \right) dx = e^{x} f(x) + c then f(x) =$$

(a)
$$\frac{1}{x+3}$$

(b)
$$-\frac{1}{x+3}$$

(c)
$$-\left(\frac{x+2}{x+3}\right)$$

(d)
$$\frac{x+2}{x+3}$$

73.
$$\int_{-\pi/2}^{\pi/2} \ln\left(\frac{2-\sin x}{2+\sin x}\right) dx = \underline{\hspace{1cm}}$$

(a) –
$$\pi$$
log 2

(b)
$$-\frac{\pi}{2} \log 2$$

(c)
$$\frac{\pi}{2} \log 2$$

74. The area enclosed b	by the curves y = sin x , (b) 2	x axis and $ x = \pi$ is (in s (c) 8	q units) (d) 3/2		
75. $xdy - ydx = \sqrt{x^2 - y^2} dx$ and $y(1) = 0$ then $y(e^{\pi/2}) = 0$					
(a) π/2	(b) $e^{\pi/2}$	(c) 1	(d) e		
76. If the angle between its coordinates, then $\lambda \epsilon$			makes obtuse angle with the axes of		
(a) (-∞, 0)	(b) $\left(-\infty, \frac{1}{2}\right)$ (c) $\left(-\frac{1}{2}\right)$	∞ , $\frac{1}{2}$) \cup $(1, \infty)$	(d) (1, ∞)		
77. Let \overline{V} = 2i + j - k, \overline{W} (a) $\sqrt{33}$ (b) 7	$V=i+3k$, if \overline{U} is a unit veloc) $\sqrt{59}$	ctor then the maximum of the defined	value of $[\overline{U}\overline{V}\overline{W}]$ is		
	plane passing through (1	1, 1, 1) and perpendicula	r to the line $\frac{x-1}{3} = \frac{y-1}{0} = \frac{z-1}{4}$		
from the origin is 3	4	7			
(a) $\frac{5}{4}$	(b) $\frac{4}{5}$	$(c)\frac{7}{5}$	(d) 0		
79. The point in which is	the join of A(– 9, 4, 5) a	nd B(11, 0, -1) is met by	the perpendicular from the origin		
(a) (2, 2, 1)	(b) (2, 1, 2)	(c) (1, 2, 2)	(d) (2, 2, 2)		
80. The value of 'm' for $2x - y + mz - 2 = 0$ is	which the straight line 3	2x - 2y + z + 3 = 0 = 4x - 3	By + 4z + 1 is parallel to the plane		
(a) – 2	(b) 8	(c) 4	(d) 2		
81. If 5 different things are placed at random in 3 different boxes, then the probability of placing them such that no box remains empty is					
(a) $\frac{30}{81}$	(b) $\frac{20}{81}$	$(c)\frac{50}{81}$	(d) $\frac{40}{81}$		
82. Two person A and B toss a die one after another. The person who throws 6 wins. If A starts the game, then the probability of his winning is					
(a) $\frac{4}{11}$	(b) $\frac{3}{11}$	$(c)\frac{5}{11}$	(d) $\frac{6}{11}$		
83. The Expansion $[x^2 - (a)]$ (b) 10	+ $(x^6 - 1)^{1/2}$] ⁵ + $[x^2 - (c)]$ 13	$-(x^6-1)^{1/2}]^5$ is a polarity (d) 14	lynomial of degree.		

- 84. If $\left|z-\frac{4}{2}\right|=2$, then the maximum value of |z| is a) $\sqrt{3}+1$ b) $\sqrt{5}+1$ c) $\sqrt{5}-1$

- d) $\sqrt{3}$ -1
- 85. For positive numbers x, y and z the numerical value of the determinant $\log_y x$ $\log_z x - \log_z y$
- (a) 0
- (b) 1
- (c) $\log_e xyz$
- (d) -log xyz
- 86. A five digit number divisible by 30 is to be formed using the digits 0,1,2,3,4,5 with our repetition of the digits. The number of ways it can be done is
- (a) 36

- (b)24
- (c)48
- (d)60

- 87. The Equation $x \frac{2}{x-1} = 1 \frac{2}{x-1}$ has (a) no root (b) one root

- (c) two roots
- (d) infinitely many

- $88. \tan 9^{\circ} \tan 27^{\circ} \tan 63^{\circ} + \tan 81^{\circ} =$
- (a) 1
- (b) 2
- (c) 3
- (d) 4

- 89. Fractional part $\frac{2^{78}}{31}$ is (a) $\frac{2}{31}$ (b) $\frac{4}{31}$

- (d) $\frac{10}{31}$
- 90. If $|z_1+z_2|=|z_1-z_2|$ then arg z_1 -arg z_2 = a) 0 b) $\pm \frac{\pi}{4}$ c) $\pm \frac{\pi}{2}$

- d)π