NCERT Class 9 Science Syllabus



Theme/ Sub-theme	Questions	Key concepts	Resources	Activities/ Processes
1. Food Higher yields	What do we do to get higher yields in our farms?	Plant and animal breeding and selection for quality improvement, use of fertilizers, manures;	Visit to any fish/ bee/dairy/pig etc farms; data showing harmful effects of insecticides; process	Collection of weeds found in fields of different crops; collection of diseased crops;
2 Materials		protection from pests and diseases; organic farming.	for the preparation of compost, vermi- compost.	discussion and studying composting/vermi- composting (Periods 8)
2. Material in our	What kinds of	Cooling by	Work done in	Experiments to
clothing	clothes help us keep	cooming by	Class VII: classware	show cooling by
ciotining	cool?	Absorption of heat	heat source black	evaporation
	Why do wet clothes	ribsorption of neat.	naner.	Experiments to
	feel cool?	3	thermometers.	show that the white objects get less hot. (Periods 5)
Different	In what way are	All things occupy	Everyday substances	To feel the texture,
kinds of	materials different	space, possess mass.	like wood, salt,	observe the colour
materials	from each other?	Definition of matter.	paper, ice, steel,	and lustre, effect of
	Is there some similarity in materials?		water, etc.	air, water and heat, etc. on each of the materials (Periods 4)
	In how many ways	Solid, liquid and gas;	Wax, water, ice, oil,	Sorting out a
	can you group the	characteristics –	sugar, camphor/	medley of materials,
	different materials	shape, volume,	ammonium	in various ways.
	you see around?	density; change of	chloride/	Observe shape and
	How do solids,	state – melting,	naphthalene.	physical state of
	liquids and gases	freezing,		different materials.

Syllabus for Secondary and Higher Secondary Levels 4



Theme/	Questions	Key concepts	Resources	Activities/
Sub-theme				Processes
	differ from each other? Can materials exist in all the three states?	evaporation, condensation, sublimation.		Observe effect of heat on each of the resources. (Teacher to perform the experiment for
				camphor, ammonium chloride and naphthalene.) (Periods 4)
What are	What are things	Elements,	Samples of	Discussion on claims
things made	around you made	compounds and	commonly available	'Air is a mixture'
of?	of?	mixtures.	elements,	(Mixture of what?
	what are the	homogeneous	mixtures Samples	How call these be
	chemical	mixtures Colloids	of solution	compound' and
	substances?	and suspensions.	suspension and colloid.	'Oxygen is an element'.
	Do substances	Equivalence – that x	Historical accounts	Titration using
	combine in a	grams of A is	Glassware, chemicals	droppers or
	definite manner?	chemically not equal	(oxalic acid, sodium	syringes, quantitative
		to x grams of B.	hydroxide, magnesium ribbon).	experiments.
	How do things combine with each other?	Particle nature, basic units: atoms and molecules.	Kits for making molecular models.	
	Are there any	Law of constant	Historical account	Discussion on the
	patterns which can	proportions. Atomic	including	fact that elements
	help us guess how	and molecular	experiments of	combine in a fixed
	things will combine	masses.	Lavoisier and	proportion through
	with each other?		Priestley.	discussion on chemical formulae of familiar
				compounds.

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Sub-theme	How do chemists weigh and count particles of matter?	Mole concept. Relationship of mole to mass of the particles and numbers. Valency. Chemical formulae of common compounds.		Processes Simple numericals to be done by the students. A game for writing formulae. e.g. criss crossing of valencies to be taught through dividing students into pairs. Each student to hold two placards: one with the symbol and the other with the valency. Keeping symbols in place, teacher to move only valencies to form the formula
What is there inside an atom?	Can we see an atom or a molecule under a microscope or by some other means? What is there inside an atom?	Atoms are made up of smaller particles: electrons, protons, and neutrons. These smaller particles are present in all the atoms but their numbers vary in different atoms. Isotopes and isobars.	Charts, films etc.	orm the formula of a compound. Brief historical account of Rutherford's experiment. (Periods 18)
3. The World of the Living Biological Diversity	How do the various plants around us	Diversity of plants and animals – basic	Specimens of some animals, and plants	Discussion on diversity and the

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	differ from each other? How are they similar? What about animals? How are they similar to and different from each other?	issues in scientific naming, Basis of classification, Hierarchy of categories/groups, Major groups of plants (salient features) (Bacteria, Thallophyta, Bryophyta, Pteridophyta, Gymnosperms and Angiosperms)	not easily observable around you.	characteristics associated with any group. (Periods 14)
		Angiosperms). Major groups of animals (salient features) (Non- chordates up to phyla and Chordates up to classes).		19 A PP
What is the living being made up of?	What are we made up of? What are the different parts of our body? What is the smallest living unit?	Cell as a basic unit of life; Prokaryotic and eukaryotic cells, multicellular organisms; cell membrane and cell wall, cell organelles: chloroplast, mitochondria, vacuoles, ER, Golgi Apparatus; nucleus, chromosomes – basic structure, number. Tissues, organs, organ	Permanent slides, model of the human body.	Observation of model of human body to learn about levels of organization – tissue, organ, system, and organism, observe blood smears (frog and human), cheek cells, onion peel cell, Spirogyra, Hydrilla leaves (cyclosis). (Periods 12)

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Syllabus for Secondary and Higher Secondary Levels 8	How do we fall sick? How do substances move from cell to cell?	What are the various causes of diseases? How can diseases be prevented? How can we remain healthy? How do food and water move from cell to cell? How do gases get into the cells? What are the substances that living organisms exchange with the external world? How do they obtain these substances?	Structure and functions of animal and plant tissues (four types in animals; meristematic and permanent tissues in plants). Health and its failure. Disease and its causes. Diseases caused by microbes and their prevention – Typhoid, diarrhoea, malaria, hepatitis, rabies, AIDS, TB, polio; pulse polio programme. Diffusion/exchange of substances between cells and their environment, and between the cells themselves in the living system; role in nutrition, water and food transport, excretion, gaseous exchange.	Newspaper articles, information from health centres, photographs of various causal organisms. Photographs, permanent slides of bacteria. Egg membrane, <i>Rhoeo</i> leaves, sugar, microscope, slides.	Surveying neighbourhood to collect information on disease occurrence pattern. Studying the life cycle of the mosquito and malarial parasite. Discussion on how malaria is spread, how to prevent mosquito breeding. (Periods 10) Looking at closed and open stomata, plasmolysis in <i>Rhoeo</i> leaf peels. (Periods 15)

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4. Moving Things, People and Ideas				
Motion	How do we describe motion?	Motion – displacement, velocity; uniform and non-uniform motion along a straight line, acceleration, distance-time and velocity time graphs for uniform and uniformly accelerated motion, equations of motion by graphical method; elementary idea of uniform circular motion.		Analysis of motion of different common objects. Drawing distance- time and velocity- time graphs for uniform motion and for uniformly accelerated motion. (Periods 12)
Force and Newton's laws	What makes things change their state of motion?	Force and motion, Newton's laws of motion: inertia of a body, inertia and mass, momentum, force and acceleration. Elementary idea of conservation of momentum, action and reaction forces.	Historical accounts; Experiences from daily life; wooden and glass boards, sand, balls; wooden support, some coins (say of Rs. 2 or Rs. 5); tumbler; balloons etc.	Demonstrating the effect of force on the state of motion of objects in a variety of daily-life situations. Demonstrate the change in direction of motion of an object by applying force. (Periods 10)
Gravitation	What makes things fall?	Gravitation; universal law of gravitation,	Spring balance	Analysis of motion of ball falling down

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		Do all things fall in the same way?	force of gravitation of the earth (gravity), acceleration due to gravity; mass and weight; free fall.		and of ball thrown up. Measuring mass and weight by a spring balance. (Periods 7)
Syllabus for Secondary and	Work, energy and power	How do we measure work done in moving anything? How does falling water make a mill run?	Work done by a force, energy, power; kinetic and potential energy; law of conservation of energy.	Rope (or string), board or plank, wooden block, ball, arrow, bamboo stick, spring, etc.	Experiments on body rolling down inclined plane pushing another body. Experiments with pendulum. Experiments with spring. Discussion. (Periods 6)
Higher Secondary Levels 10	Floating bodies	How does a boat float on water?	Thrust and pressure. Archimedes' principle, buoyancy, elementary idea of relative density.	Cycle pump; board pins, bulletin board, mug, bucket, water etc.	Experiments with floating and sinking objects. (Periods 4)
0000	How do we hear from a distance?	How does sound travel? What kind of sounds can we hear? What is an echo? How do we hear?	Nature of sound and its propagation in various media, speed of sound, range of hearing in humans; ultrasound; reflection of sound; echo and sonar. Structure of the human ear (auditory aspect only).	String, ball or stone as bob, water tank, stick, slinky, rope, echo tube, rubber pipe etc. Model or chart showing structure of the ear.	Experiment on reflection of sound. (Periods 10)

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5. How Things Work 6. Natural Phenomena 7. Natural Resources Balance in Nature	Why do air, water and soil seem not to be consumed? How does the presence of air support life on earth? How have human activities created disturbances in the atmosphere? How does nature work to maintain	Physical resources: air, water, soil. Air for respiration, for combustion, for moderating temp- eratures, movements of air and its role in bringing rains across India. Air, water and soil pollution (brief introduction). Holes in ozone layer	Daily newspapers, magazines and other reading materials. Weather reports over a few months and air quality reports over the same time period. Case study material.	Case studies of actual situation in India with more generalised overview of inter relationship of air, water, soils, forests. Debates on these issues using resources mentioned alongside, visit to/ from an environmental
	balance of its components?	and the probable damages. Bio-geo chemical cycles in nature: water, oxygen, carbon, nitrogen.		NGO; discussion. (Periods 15)