CBSE Class 10 Science Sample Paper Solution SA 1 Set 2

- 1. Ωm
- 2. Potasium

SCORE KEY

- 3. Acid facilitates the action of pepsin on proteins
- 4. Salivary amylase
- 5. Definition . V I
- 6. Antioxidants. Fill the packet with nitrogen.
- 7. In parallel connection failure of one device does not affect others.(1) Different devices need different amount of electric current. This is possible only in parallel connection.
- 8. Pancreatic lipase converts fats into fatty acids and glycerol(1 mark). It takes place in the small intestine(1 mark).
- 9. (a)Ca(OH)₂ + CO₂ ----- \rightarrow CaCO₃ + H₂ O (b) 2Mg + O₂ ----- \rightarrow 2MgO
- 10. Alloys are homogeneous mixture of a metal with another metal or non metal.
- 11. Sodium is highly reactive. To prevent oxide formation it is kept in kerosene.
- **12.** Hydrogen Gas. Fe + H_2SO_4 ------ FeSO₄ + H_2
- **13.** Oxygen is transported in the form of oxy-heamoglobin.
- 14. Glucose \rightarrow Pyruvate \rightarrow absence of oxygen \rightarrow Carbon dioxide & alchohol + Energy
 - → Lack of oxygen → Lactic acid+Energy
 - → Presence of oxygen→ Carbon dioxide +Water+Energy
- 15. Resistance of a conductor depends on length, area of cross section and nature of the material.R L
- **16.** Metals are malleable, ductile, lustrous and sonorous. They are good conductors of heat and electricity.
- **17.** Nephrons are the structural and functional unit of kidneys. Bowmans Cup glomurlus, pressure filtration of blood, tubular reabsorption etc.,
- **18.** Magnetic lines originate from north pole. They will never intersect. The degree of closeness of the magnetic field lines shows the strength of the magnet.
- 19. Fossil fuels are non-renewable.they pollute air. They are not available everywhere.
- **20.** Definition. Alloys have higher resistivity than that of pure metals.
- **21.** Renewable sources are available at all times.thay will not get exhausted.Non-renewable sources are limited. eg:- coal,petroleum,natural gases.,
- 22. Definition. Fuse and earth wire.
- **23.** Neuron diagram (2 marks). Labelling (1 mark). Explanation (2 marks).
- 24. Bleaching powder is prepared by passing chlorine over dry slaked lime. Chemical equation. Any 2 uses.(2+1+2)
- **25.** To get 4 Ω resistance connect (a) 3 Ω and 6 Ω in parallel and 2 Ω in series.

(b) to get 1 Ω resistance connect all the resistors in parallel. Diagram of circuit.

- 26. b.increases
- 27. c.Oersted
- 28. a.Generator
- **29.** a.Methane
- **30.** c.Mouth
- **31.** b.Translocation
- **32.** b.Phototropism
- **33.** c.Sour taste
- **34.** c.mercury
- **35.** a.decomposition
- **36.** a.silver
- **37.** a.parallel
- **38.** a.parallel straight lines
- **39.** a.Galvanometer
- 40. d.Firewood
- 41. b.Liver