Tripura Board Class 10 Science Sample Question Paper

Sample Question Paper Science Physics and Chemistry Theory

Time: 2 Hours Full Marks: 53

Class -X

General Instructions

- 1. This Question Paper comprises 2 Groups- A (Physics) and B (Chemistry). You are to answer questions from both the Groups.
- 2. All questions are compulsory.
- 3. Use of Calculator or any other electronic device in the examination hall is strictly prohibited.

Group- APHYSICS Full Marks – 27

Answer the following questions :-

 What is the unit of Resistivity? What is the main constituent of Biogas? Choose the correct answer – The direction of rotation of Barlow's Wheel is determined by a) Fleming's Left Hand Rule b) Fleming's Right Hand Rule c) Ampere's Swimming Rule d) None of these 	1 1 1
4) What is called Electromagnetic Induction?5) Write down two advantages of alternating current over direct current.6) What is Myopia or Short-sight? Show its remedy with the help of a ray diagram.	2 2 (1+1)=2
7) During sunrise and sunset the Sun appears to be red while at noon it a to be white – why?	appears 2
 8) Write two differences between Nuclear Fission and Nuclear Fusion. 9) The equivalent resistances of two resistors connected in series and parallel are 6 ohms and 5/2 ohm respectively. Calculate the value of 	2
each resistor.	3
10)Write Joule's Laws of heating effect of current.	3
11)For a convex lens where should an object be placed so as to get its re and magnified image? Show drawing a ray diagram.	al (2+1)=3

Or

For a concave mirror where should an object be placed so as to get its virtual and magnified image? Show drawing a ray diagram. (2+1)=3

- 12)i) Calculate the focal length of a lens having power -2.5 dioptre(D) and mention the nature of the lens.
 - ii) What kind of mirror is used in the view-finder of vehicles and why?
 - iii) What do you understand by magnification in case of a lens?

 $[(1+1)+(\frac{1}{2}+1\frac{1}{2})+1]=5$

Or

- i) Refractive index of water is 1.33 what do you mean by this statement?
- ii) An object of height 2 cm. is placed at a distance of 20 cm. from a concave mirror. If a real image is formed 40 cm. away from the mirror, determine the focal length of the mirror. (2+3) =5

Group - BCHEMISTRY

Full Marks - 26

- 13. Which of the following is an acidic salt?
 - a) CaCl₂
- b) Na₂CO₃
- c) KHCO₃
- d) MgSO₄

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14. State the composition of stainless steel.

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15. Name the second member of ketone.

 $\frac{1}{2} + \frac{1}{2} = 1$

- 16. Name two greenhouse gases.
- 17. Balance the following equations:
 - i) Fe + H_2O \longrightarrow $Fe_3O_4 + H_2$
 - ii) $NH_3 + O_2 \longrightarrow N_2O_5 + H_2O$

- 1 + 1 = 2
- 18. Oxidation and reduction take place simultaneously explain with one example. 2
- 19. Zinc oxide is an amphoteric oxide. Support your answer with equations.

Or

Dilute H₂SO₄ can be kept in copper container but not in Zinc container. Explain.

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- 21. i) Write the chemical name and formula of bleaching powder.
 - ii) How is it prepared?

20. Why should we conserve forest?

iii) Write two uses of it.

1+1+1=3

- 22. i) Name a metal which can be extracted by carbon reduction process from its ore.
 - ii) Name the anode, the cathode and the electrolyte used in the electrolytic refining of impure copper. What is anode mud?

 1+2 = 3
- 23. i) State the modern periodic law.
 - ii) Why does the size of the atom decrease across a period?
 - iii) Mention the position (period and group) of Sodium in the Periodic table. 1+1+1 = 3

24. i) Complete the equation:

- ii) Show with one chemical test that ethylene is an unsaturated hydrocarbon.
- iii) Write the IUPAC name of the following:

1+2+2 = 5

Or

- i) What is the difference between soap and detergent chemically?
- ii) Describe in brief the cleansing action of soap.
- iii) Why do soaps not form lather easily in hard water?

1+2+2 = 5

CLASS - X

PHYSICS PRACTICAL EXAMINATION (Full Marks: 07)

SI	Particulars	Marks				
No.						
a) Mar	a) Marks distribution for Experiments					
1.	Principle or working formula 1/2					
2.	Circuit Diagram/Ray Tracing	1				
3.	Recording of Data with graph if there be any	11/2				
4.	Calculation	1/2				
5.	Result with accuracy	1/2				
	4					
b) Labo	2					
duly sig	duly signed					
c) Viva	1					
	7					

CHEMISTRY PRACTICAL EXAMINATION (Full Marks : 07)

- a) Marks distribution on experiment:
 - i) Procedure
 - ii) Observation

iii) Inference

2+1+1=4

b) Laboratory Note Book with at least three experiments written and duly signed:

2

c) Viva Voce:

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Blueprint for Question Paper setting for Group A (Physics and Chemistry)

	Unit	Name	MCQ/Objective	SA	LA-I	LA-II	Total
CS			(1 mark)	(2marks)	(3 marks)	(5 marks)	Marks
	I)	Electricity	1	-	3 x 2	ı	7
	II)	Magnetic	1	2 x 2	-	-	5
		effect of					
		Current					
	III)	Light –	-	-	3 x1	5 x 1	8
		Reflection,					
hysics		Refraction					
	IV)	Human	-	2 x 2	-	-	4
		Eye and					
		Colorful					
		World					
	V)	Sources	1	2 x 1	-	-	3
		of Energy			- 4		
	Total Ma		→		- 6.1		27
	No. of qu	ıestions 	3	5	3	1	12

	Unit	Name	MCQ/Objective	SA	LA-I	LA-II	Total
			(1 mark)	(2	(3 marks)	(5 marks)	Marks
				marks)		. 1	
	VI	Chemical	-	2 x 2	1 - 1	O ₃ '-	4
		Reaction					
	VII	Acids, Bases	1 x 1	0-	3 x 1	-	4
Chemistry		and Salts			200		
	VIII	Metals and	1 x 1	2 x 1	3 x1	-	6
		Non-Metals					
	IX	Carbon	1 x 1	·6-	-	5 x 1	6
		Compounds					
)	X	Periodic		-	3 x 1	-	3
Ö		classification					
		of elements					
	ΧI	Conservation	1 x 1	2 x 1	-	-	3
		of natural					
		resources					
	Total Marks					26	
	No. o	f questions	4	4	3	1	12

উপরে প্রশ্নপত্র তৈরীর একটি ছক দেওয়া হল । প্রশ্নসংখ্যা এবং প্রশ্নের মান (marks) অপরিবর্তিত রেখে এই ছক পরিবর্তন করা যেতে পারে । যেমন MCQ/Objective এবং LA-I এর প্রশ্ন Unit-I থেকে না নিয়ে অন্য Unit থেকেও নেওয়া যেতে পারে । অনুরূপভাবে LA-II এর প্রশ্নটি Unit-III থেকে না নিয়ে অন্য Unit থেকেও নেওয়া যেতে পারে ।

Note: i)There should be approximately 30% knowledge based, 40% understanding based, 25% application based and 5% skill based questions.

ii) While framing question, alternative question for internal choice should be given from the same unit.