

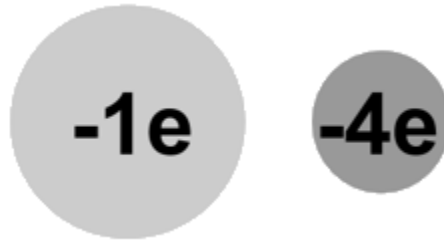
BYJU'S Home Learning Program

Topic covered:

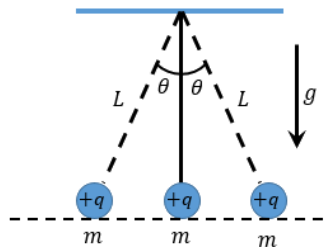
- Electrostatics (Session - 1)

Daily Practice Problems

1. When 10^{19} electrons are removed from a neutral metal plate, the electric charge on it is
 - a. -1.6 C
 - b. $+1.6 \text{ C}$
 - c. 10^{+19} C
 - d. 10^{-19} C
2. A metal disk with a charge $Q_1 = -1e$ sits near a metal sphere that has a charge $Q_2 = -4e$. If the disk touched the sphere, what is the possible final charge on each object?



- a. $Q_1 = -2e, Q_2 = -3e$
 - b. $Q_1 = -3e, Q_2 = -3e$
 - c. $Q_1 = -1e, Q_2 = +7e$
 - d. $Q_1 = -2e, Q_2 = -2e$
3. Three identical point charges, each of mass m and charge q , hang from three strings as shown in the given figure. The value of q in terms of m, L and θ is



- a. $q = \sqrt{\left(\frac{16}{5}\right) \pi \epsilon_0 m g L^2 \sin^2 \theta \tan \theta}$
- b. $q = \sqrt{\left(\frac{16}{15}\right) \pi \epsilon_0 m g L^2 \sin^2 \theta \tan \theta}$
- c. $q = \sqrt{\left(\frac{15}{16}\right) \pi \epsilon_0 m g L^2 \sin^2 \theta \tan \theta}$
- d. none of these



Answer Key

Question Number	1	2	3	4	5
Answer Key	(b)	(a)	(a)	a. 1.875×10^{12} b. 17×10^{-19}	(d)

Question Number	6	7	8	9	10
Answer Key	(d)	(a)	(a)	(b)	(a)



Solutions

1. (b)

Charge on an electron is a constant i.e., $-1.6 \times 10^{-19} \text{ C}$. So, if we remove an electron from a substance, we remove a charge of $-1.6 \times 10^{-19} \text{ C}$. Therefore, by quantization of charge, the charge gained by the substance is equal to $+1.6 \times 10^{-19} \text{ C}$ and if 10^{19} electrons are removed then the charge gain by the metal plate is

$$\begin{aligned} Q &= ne \\ Q &= 10^{19} \times 1.6 \times 10^{-19} \text{ C} \\ Q &= 1.6 \text{ C} \end{aligned}$$

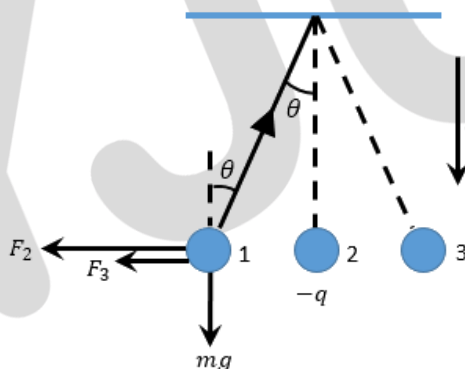
2. (a)

Total charge initially = $-1e + (-4e) = -5e$

Using the quantization of charges, the only possibility is option (a)

Total charge finally = $-2e + (-3e) = -5e$

3. (a)



$$\begin{aligned} F &= F_2 + F_3 = \frac{kq^2}{(L \sin \theta)^2} + \frac{kq^2}{(2L \sin \theta)^2} \\ &= \frac{5}{4} \frac{kq^2}{L^2 \sin^2 \theta} \dots (1) \end{aligned}$$

$$T \sin \theta = Fe \dots (2)$$

$$T \cos \theta = mg \dots (3)$$

$$\text{From (1), (2) and (3), } q = \sqrt{\left(\frac{16}{5}\right) \pi \epsilon_0 mg L^2 \sin^2 \theta \tan \theta}$$



4.

Given,

$$q = 3 \times 10^{-7} \text{ C}, e = 1.6 \times 10^{-19} \text{ C}$$

(a) To estimate the number of the electron transferred,

$$q = ne$$

$$n = \frac{3 \times 10^{-7}}{1.6 \times 10^{-19}} = 1.875 \times 10^{12} \text{ electrons}$$

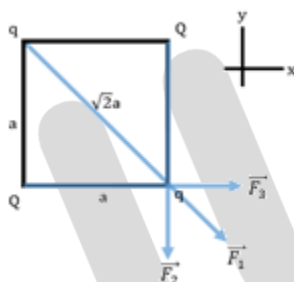
(b) Yes, mass is transferred from wool to polythene.

We know that the mass of electron, $m = 9.1 \times 10^{-31} \text{ kg}$

The mass transferred is $M = nm$

$$M = 1.875 \times 10^{12} \times 9.1 \times 10^{-31} = 17 \times 10^{-19} \text{ kg}$$

5. (d)



The net force on q at one corner is zero if $\vec{F}_1 + \vec{F}_2 + \vec{F}_3 = 0$

$$\text{Or } F_1 \cos 45^\circ \hat{i} - F_1 \sin 45^\circ \hat{j} - F_2 \hat{j} + F_3 \hat{i}$$

So, $F_1 \cos 45^\circ = -F_3 \dots (1)$ and $F_1 \sin 45^\circ = -F_2 \dots (2)$

$$\text{Using (1), } \frac{kq^2}{(\sqrt{2}a)^2} \times \frac{1}{2} = -\frac{kqQ}{a^2}$$

$$\text{Or } q = -2\sqrt{2}Q$$

6. (d)

$$\text{Gravitational force between electrons } F_g = \frac{G(m_e)^2}{r^2}$$

$$\text{Electrostatic force between electrons } F_e = \frac{k(e)^2}{r^2}$$

$$\frac{F_g}{F_e} = \frac{G(m_e)^2}{k(e)^2} = \frac{6.67 \times 10^{-11} \times (9.1 \times 10^{-31})^2}{9 \times 10^9 \times (1.6 \times 10^{-19})^2} = 2.39 \times 10^{-43}$$



7. (a)

The force between $4q$ and q : $F_1 = \frac{1}{4\pi\epsilon_0} \frac{4q \times q}{l^2}$

The force between Q and q : $F_2 = \frac{1}{4\pi\epsilon_0} \frac{Q \times q}{\left(\frac{l}{2}\right)^2}$

Given $F_1 + F_2 = 0$

$$\frac{4q^2}{l^2} = \frac{4Qq}{l^2} \Rightarrow Q = -q$$

8. (a)

$$F = \frac{kQ^2}{r^2} = 9 \times 10^9 \times 1^2 \times \frac{1}{(1000)^2} = 9 \times 10^3 \text{ N}$$

9. (b)

A hydrogen atom consists of an electron and a proton

\therefore Charge on one hydrogen atom = $q_e + q_p = -e + (e + \Delta e) = \Delta e$

Since a hydrogen atom carry a net charge Δe ,

Electrostatic force, $F_e = \frac{1}{4\pi\epsilon_0} \frac{(\Delta e)^2}{d^2}$ will act between the two hydrogen atoms.

The gravitational force between them is given as $F_g = \frac{Gm_h m_h}{d^2}$

Since the net force on the system is zero, $F_e = F_g$

$$\begin{aligned} \frac{1}{4\pi\epsilon_0} \frac{(\Delta e)^2}{d^2} &= \frac{Gm_h m_h}{d^2} \\ (\Delta e)^2 &= 4\pi\epsilon_0 Gm_h^2 \\ &= 9 \times 10^9 \times 6.67 \times 10^{-11} \times (1.67 \times 10^{-27})^2 = 10^{-37} \text{ C} \end{aligned}$$

10. (a)

From the figure, $T \cos \theta = mg$(i)

and $T \sin \theta = \frac{kq^2}{x^2}$ (ii)

From equations (i) and (ii), $\tan \theta = \frac{kq^2}{x^2 mg}$

Since θ is small, $\tan \theta \cong \sin \theta = \frac{x}{2l}$

$$\therefore \frac{x}{2l} = \frac{kq^2}{x^2 mg}$$

$$q^2 = x^3 \frac{mg}{2lk} \Rightarrow \frac{dq}{dt} \propto \frac{3}{2} \sqrt{x} \frac{dx}{dt} = \frac{3\sqrt{x}v}{2}$$

Since, $\frac{dq}{dt} = \text{constant}$

From the above equation, $v \propto \frac{1}{\sqrt{x}}$

