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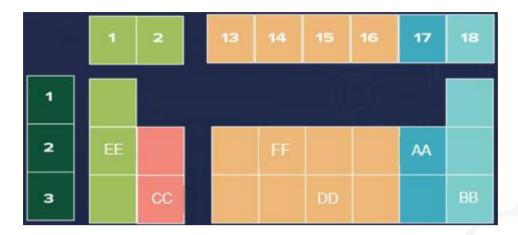
- 1. Two elements X and Y need to be arranged in Mendeleev's periodic table. X has 3 protons more than Y but their neutrons are equal in number. Which of the following is correct about X and Y?
 - A. X will be placed after Y
 - **B.** X and Y should be placed together.
 - **C.** X and Y are isotopes.
 - **D.** Y will be placed after X.
- 2. Identify the option that is not one of the Dobereiner's triad?
 - A. Chlorine-35, Bromine-80, Iodine-127
 - B. Potassium-39, Chlorine-35.5, Bromine-80
 - C. Calcium-40, Strontium-88, Barium-137
 - D. Lithium-6, Sodium-22, Potassium-39
- 3. The atomic masses of X and Z are 20 u and 40 u, respectively. If X, Y and Z form a Dobereiner's triad, what should be the atomic mass of Y?
 - **A**. 20 u
 - **B**. 40 u
 - **c**. 30 u
 - **D**. 50 u



- 4. Identify the set of elements that will form Dobereiner triad.
 - A. Lithium, sodium, potassium
 - B. Nitrogen, phosphorus, arsenic
 - C. Helium, neon, argon
 - **D.** Helium, hydrogen, potassium
- 5. Which of the following statements is correct?
 - **A.** Law of octaves worked quite well for heavier elements but failed for lighter elements.
 - **B.** Law of octaves failed for both lighter elements and heavier elements.
 - C. Law of octaves worked quite well for lighter elements but failed for heavier elements.
 - **D.** Law of octaves was first given by Mendeleev.
- 6. How many groups and periods are there in the modern periodic table?
 - A. 18 Groups and 7 Periods
 - **B.** 7 Groups and 18 Periods
 - **C.** 18 Groups and 9 Periods
 - **D.** 18 Groups and 18 Periods



7. Which of the following statement is incorrect regarding the given elements?



- **A.** AA is the most electronegative element
- **B.** BB is a noble gas
- C. CC is an alkali metal
- **D.** DD has 5 electrons in its outermost shell
- 8. Metallic character _____ down the group in the periodic table.
 - A. increases
 - B. decreases
 - C. increases and then decreases
 - **D.** decreases and then increases
- 9. Which of the following elements has twice as many electrons in its second shell as it has in its first shell?
 - A. Carbon
 - B. Nitrogen
 - C. Oxygen
 - D. Sulphur



10.	The	valency	of a	luminium	is	
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- **A**. ₁
- B. 2
- C. 3
- D. 4