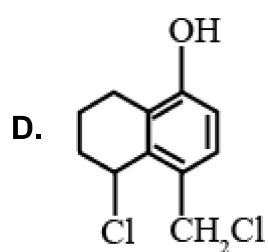
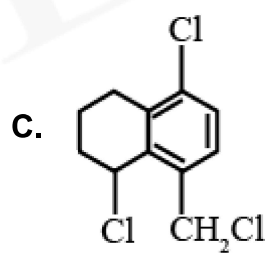
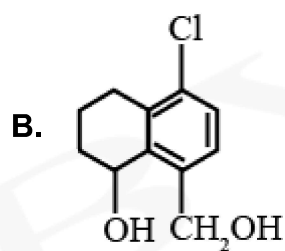
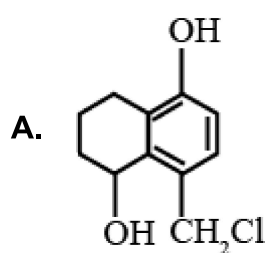
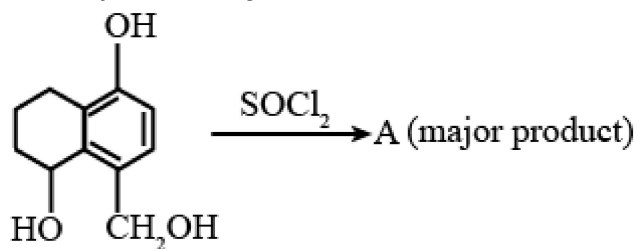


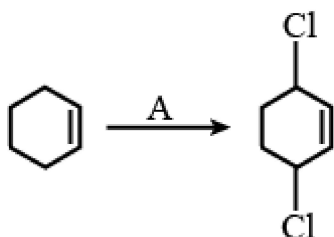
Halo compounds + Compounds containing Oxygen

1. Identify A in the given reaction,



Halo compounds + Compounds containing Oxygen

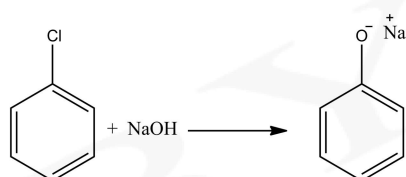
2.



Identify the reagent(s) 'A' and condition(s) for the reaction.

- A. $A = HCl$; Anhydrous $AlCl_3$
- B. $A = Cl_2$; UV light
- C. $A = Cl_2$; dark, Anhydrous $AlCl_3$
- D. $A = HCl, ZnCl_2$

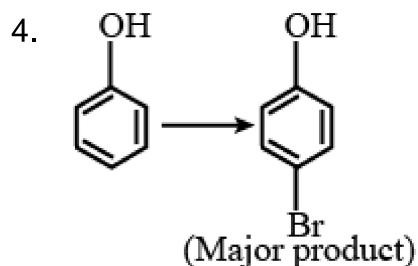
3.



The above reaction requires which of the following reaction condition?

- A. 573 K, 300 atm
- B. 623 K, Cu, 300 atm
- C. 573 K, Cu, 300 atm
- D. 623 K, 300 atm

Halo compounds + Compounds containing Oxygen

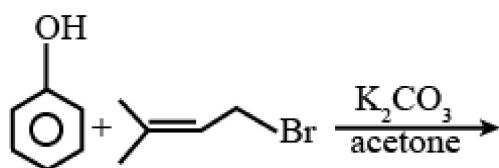


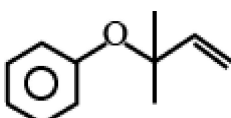
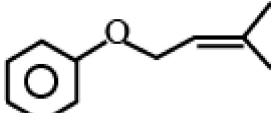
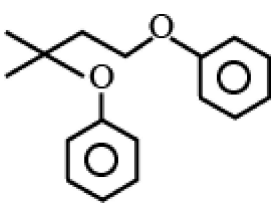
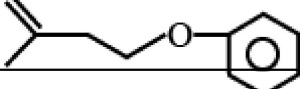
The given reaction can occur in the presence of

- (a) Bromine water
- (b) Br_2 in CS_2 , 273 K
- (c) $Br_2/FeBr_3$
- (d) Br_2 in $CHCl_3$, 273 K

Choose the correct answer from the options given below

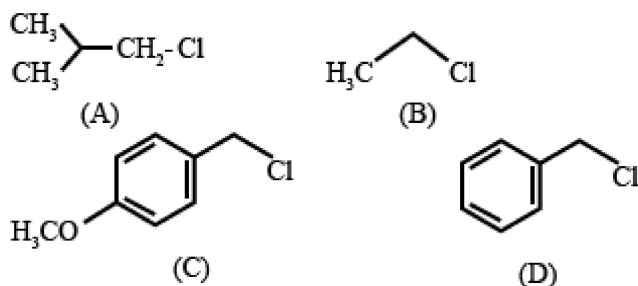
- A. (a) and (c) only
 - B. (a), (b) and (d) only
 - C. (b) and (d) only
 - D. (b), (c) and (d) only
5. The major product of the following reaction, if it occurs by S_N2 mechanism is :



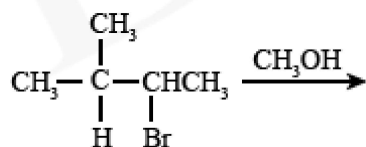
- A. 
- B. 
- C. 
- D. 

Halo compounds + Compounds containing Oxygen

6. Increasing order of reactivity of the following compounds for S_N1 substitution is:



- A. $(A) < (B) < (D) < (C)$
 B. $(B) < (C) < (D) < (A)$
 C. $(B) < (C) < (A) < (D)$
 D. $(B) < (A) < (D) < (C)$
7. The major product of the following reaction is :

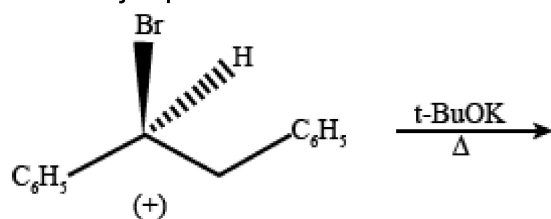


- A. $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CH}_2\text{CH}_3 \\ | \\ \text{OCH}_3 \end{array}$
 B. $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CH}=\text{CH}_3 \\ | \\ \text{H} \end{array}$
 C. $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CHCH}_3 \\ | \quad | \\ \text{H} \quad \text{OCH}_3 \end{array}$
 D. $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CHCH}_3 \end{array}$

- E. Answer not mention

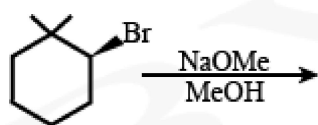
Halo compounds + Compounds containing Oxygen

8. The major product obtained in the following reaction is :



- A. $\text{C}_6\text{H}_5\text{CH}=\text{CHC}_6\text{H}_5$
- B. $(+)\text{C}_6\text{H}_5\text{CH}(\text{O}^t\text{Bu})\text{CH}_2\text{C}_6\text{H}_5$
- C. $(-)\text{C}_6\text{H}_5\text{CH}(\text{O}^t\text{Bu})\text{CH}_2\text{C}_6\text{H}_5$
- D. $(\pm)\text{C}_6\text{H}_5\text{CH}(\text{O}^t\text{Bu})\text{CH}_2\text{C}_6\text{H}_5$

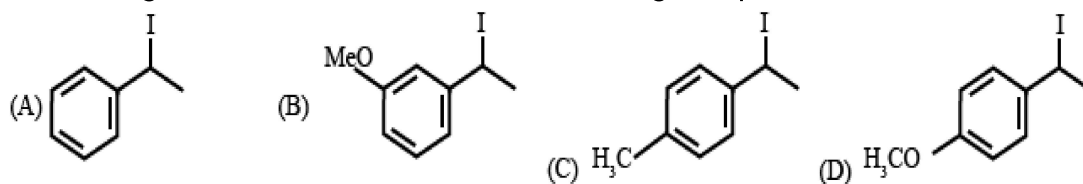
9. The major product of the following reaction is :



- A.
- B.
- C.
- D.

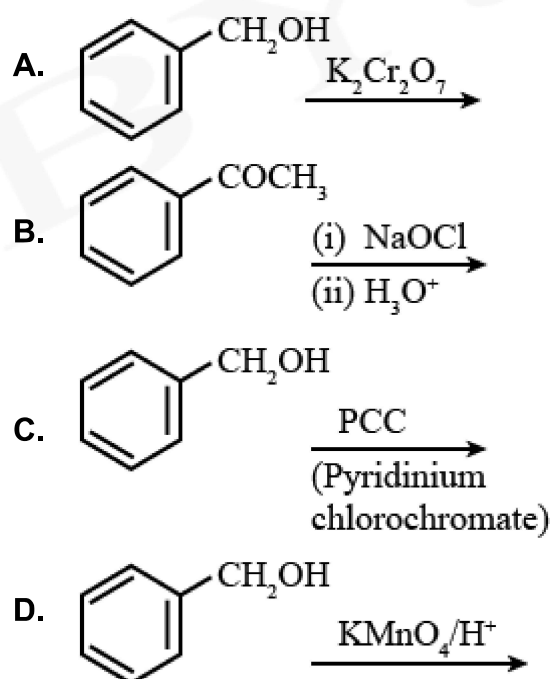
Halo compounds + Compounds containing Oxygen

10. Increasing rate of S_N1 reaction in the following compounds is :



- A. $(B) < (A) < (D) < (C)$
 B. $(A) < (B) < (C) < (D)$
 C. $(A) < (B) < (D) < (C)$
 D. $(B) < (A) < (C) < (D)$

11. The reaction that does not give benzoic acid as the major product is:

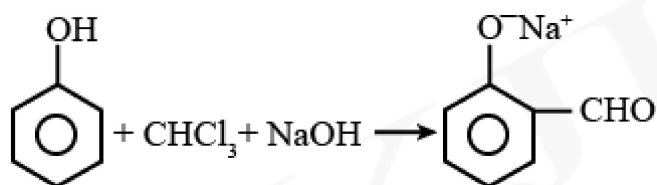


Halo compounds + Compounds containing Oxygen

12. When vapours of a secondary alcohol is passed over heated copper at 573 K , the product formed is:

- A. a carboxylic acid
- B. an aldehyde
- C. a ketone
- D. an alkene

13. In the reaction

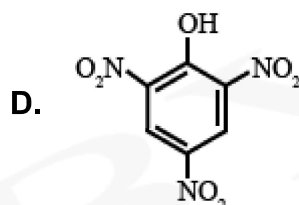
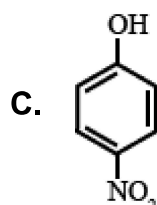
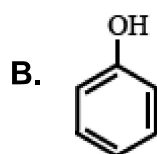
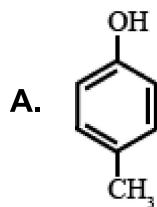


The electrophile involved is

- A. Dichloromethyl cation (CHCl_2^+)
- B. Formyl cation (CHO^+)
- C. Dichlorocarbene ($:\text{CCl}_2$)
- D. Dichloromethyl anion (CHCl_2^-)

Halo compounds + Compounds containing Oxygen

14. Which one is the most acidic compound?

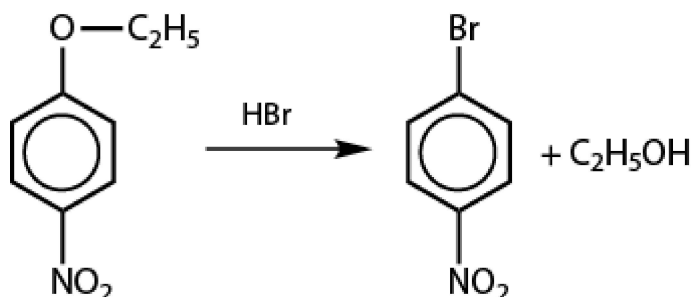


15. Which of the following reagents would distinguish cis-cyclopenta-1, 2-diol from the trans-isomer?

- A. Aluminium isopropoxide
- B. Acetone
- C. Ozone
- D. MnO_2

Halo compounds + Compounds containing Oxygen

16. Assertion:



Reason: Due to formation of highly stable carbocation.

- A. Both assertion and reason are true and reason is the correct explanation of assertion
- B. Both assertion and reason are true and reason is not the correct explanation of assertion
- C. Assertion is true but reason is false
- D. Both assertion and reason are false

17. In the following reaction,

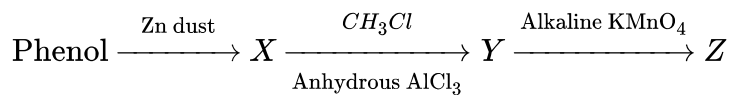


(A) and (B) respectively, are

- A. Both are
- B. Both are
- C. and
- D. and

Halo compounds + Compounds containing Oxygen

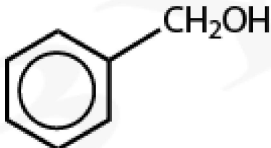
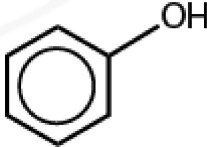
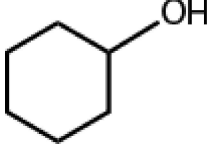
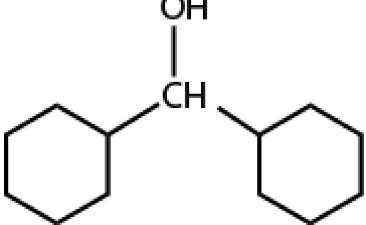
18. Consider the following reaction:



The product Z is

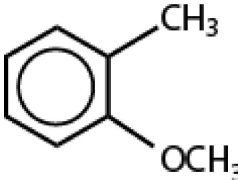
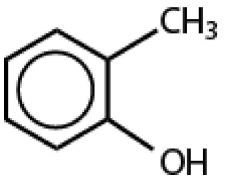
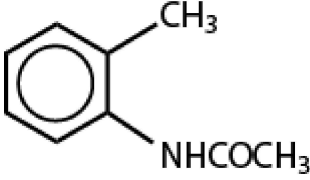
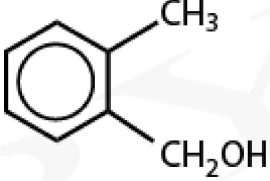
- A. Benzene
- B. Toluene
- C. Benzaldehyde
- D. Benzoic acid

19. Which one of the following compound has the most acidic nature?

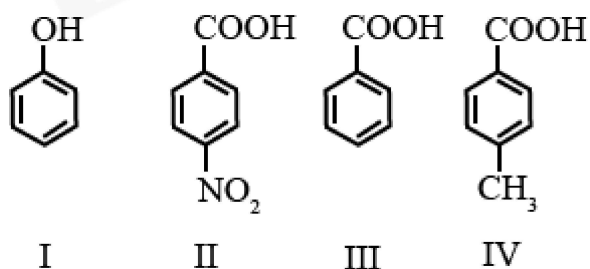
- A. 
- B. 
- C. 
- D. 

Halo compounds + Compounds containing Oxygen

20. Which one of the following is most reactive towards electrophilic reagent?

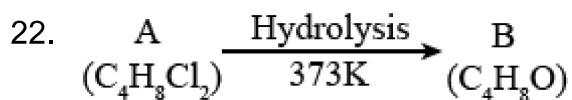
- A. 
- B. 
- C. 
- D. 

21. The correct order of acid character of the following compounds is :



- A. $IV > III > II > I$
- B. $II > III > IV > I$
- C. $I > II > III > IV$
- D. $III > II > I > IV$

Halo compounds + Compounds containing Oxygen

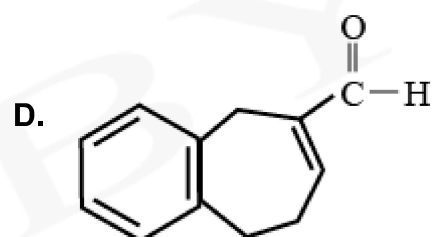
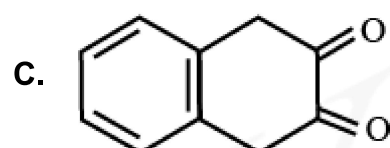
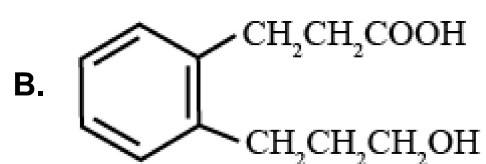
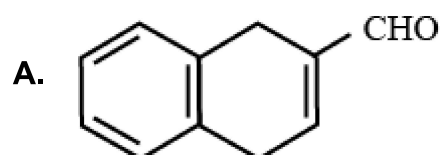
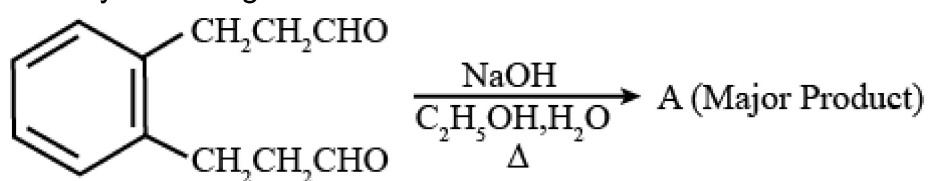


B reacts with Hydroxyl amine but does not give Tollen's test. Identify A and B.

- A. 2, 2-Dichlorobutane and Butanal
 - B. 1, 1-Dichlorobutane and Butanal
 - C. 1, 1-Dichlorobutane and Butan-2-one
 - D. 2, 2-Dichlorobutane and Butan-2-one
23. 2, 4-DNP test can be used to identify
- A. Aldehyde
 - B. Amine
 - C. Ether
 - D. Halogens

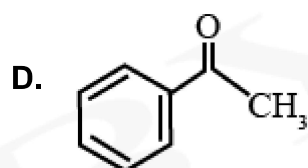
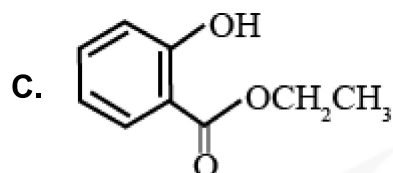
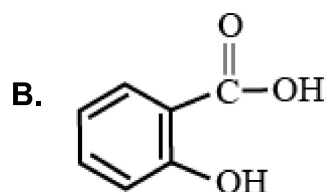
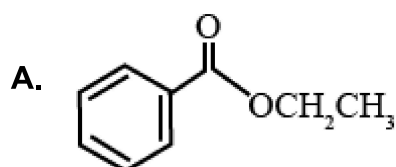
Halo compounds + Compounds containing Oxygen

24. Identify A in the given chemical reaction.

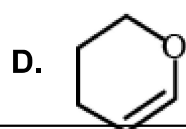
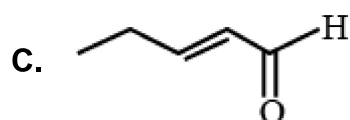
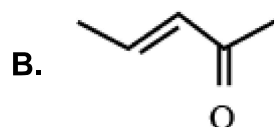
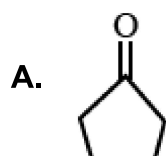
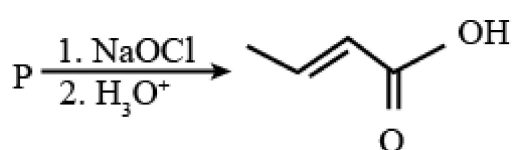


Halo compounds + Compounds containing Oxygen

25. Which one of the following compounds will give orange precipitate when treated with 2, 4-dinitrophenyl hydrazine?

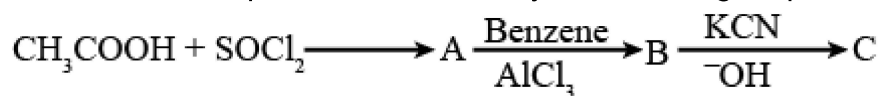


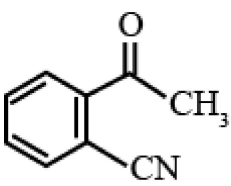
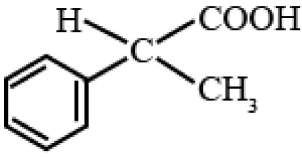
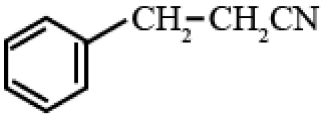
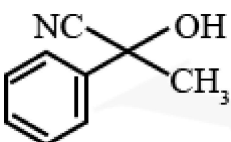
26. The structure of the starting compound P used in the reaction given below is:



Halo compounds + Compounds containing Oxygen

27. The structure of product C, formed by the following sequence of reaction is :



- A. 
- B. 
- C. 
- D. 

28. $\text{R} - \text{CN} \xrightarrow[\text{(ii) H}_2\text{O}]{\text{(i) DIBAL-H}} \text{R} - \text{Y}$ Consider the above reaction and identify "Y"

- A. $-\text{CH}_2\text{NH}_2$
- B. $-\text{CHO}$
- C. COOH
- D. $-\text{CONH}_2$

Halo compounds + Compounds containing Oxygen

29. Match List-I with List-II

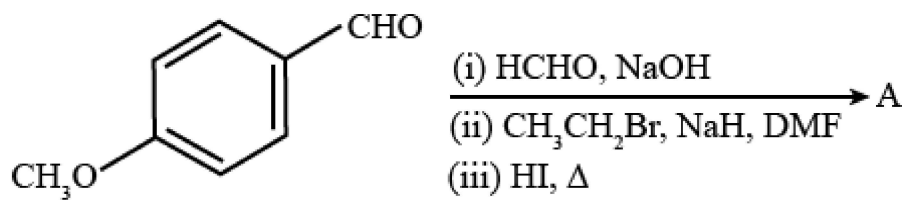
List-I (Chemical reaction)	List-II (Reagent used)
(a) $CH_3COOCH_2CH_3 \rightarrow CH_3CH_2OH$	(i) CH_3MgBr/H_3O^+ (1 equivalent)
(b) $CH_3COOHCH_3 \rightarrow CH_3CHO$	(ii) H_2SO_4/H_2O
(c) $CH_3C \equiv N \rightarrow CH_3CHO$	(iii) <i>DIBAL</i> – H/H_2O
(d) $CH_3C \equiv N \rightarrow CH_3COCH_3$	(iv) $SnCl_2, HCl/H_2O$

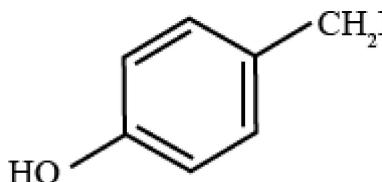
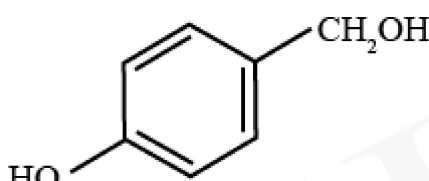
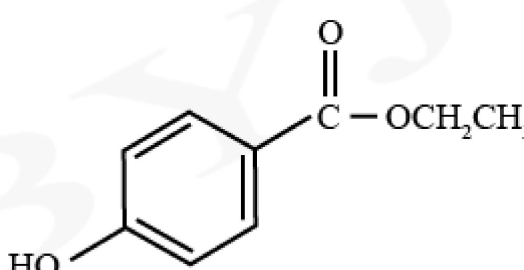
Choose the most appropriate match.

- A.** (a) – (ii), (b) – (iii), (c) – (iv), (d) – (i)
- B.** (a) – (iii), (b) – (ii), (c) – (i), (d) – (iv)
- C.** (a) – (ii), (b) – (iv), (c) – (iii), (d) – (i)
- D.** (a) – (iv), (b) – (ii), (c) – (iii), (d) – (i)

Halo compounds + Compounds containing Oxygen

30. Identify A in the following chemical reaction.



- A. 
- B. 
- C. 
- D. 