

CBSE Class 12 Chemistry Chapter 13 Amines Worksheet – Set 2

Q1. How many isomeric amines with the formula C_7H_9N contain a benzene ring?

- (a) Five
- (b) Four
- (c) Three
- (d) None of the above

Q2. The oxidation of aniline with $K_2Cr_2O_7 / H_2SO_4$ produces

- (a) Benzoic acid
- (b) p- Benzo quinone
- (c) p- Nitro phenol
- (d) None of the above

Q3. The electrolytic reduction of nitro benzene in a strongly acidic medium produces

- (a) Aniline
- (b) Phenyl hydroxy amine
- (c) p- amino phenol
- (d) None of the above

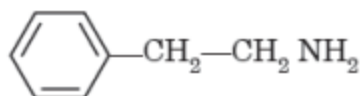
Q4. In the nitration of benzene with concentrated nitric acid and concentrated sulphuric acid, the electrophile is

- (a) NO_2^+
- (b) NO_2^-
- (c) NO_2
- (d) None of the above

Q5. An organic amino compound reacts with aqueous nitrous acid at low temperatures to produce an oily nitrosamine. The compound is

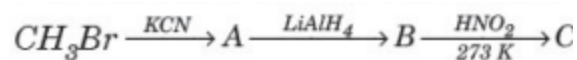
- (a) $(C_2H_5)_3N$
- (b) $(C_2H_5)_2NH$
- (c) $C_2H_5NH_2$
- (d) None of the above

Q6. Write the IUPAC and common name of the following compound.



Q7. Draw the structure of m- toluidine.

Q8. Identify A, B and C in the reaction mentioned below.



Q9. Convert nitrobenzene to aniline.

Q10. Why does methyl amine have a lower boiling point than methanol?

- Q11.** Why is methyl amine a more substantial base than ammonia?
- Q12.** Why does aniline dissolve in an aqueous hydrochloric acid solution?
- Q13.** Why is it difficult to prepare pure amines by the ammonolysis of alkyl halides?
- Q14.** Convert toluene to p-iodotoluene.
- Q15.** What is Hinsberg reagent?
- Q16.** Draw the structure, IUPAC names, and indicate primary, secondary and tertiary to the five isomeric amines with the formula C_7H_9N containing a benzene ring.
- Q17.** An aromatic compound A on treatment with aqueous ammonia and heating forms compound B, which on heating with Br and KOH forms a compound C of molecular formula C_6H_7N . Write the structures and IUPAC names of compounds A, B and C.
- Q18.** Write the chemical equation for the following reactions.
- (a) The reaction of ethanolic NH_3 with C_2H_5Cl .
- (b) Ammonolysis of benzyl chloride and amine reaction formed with two moles of CH_3Cl .
- Q19.** Complete the following reactions.
- (a)
$$C_6H_5N_2Cl + H_3PO_2 + H_2O \longrightarrow$$
- (b)
$$CH_3CH_2NH_2 + CHCl_3 + alc. KOH \longrightarrow$$
- (c)
$$C_6H_5N_2^+Cl^- \xrightarrow[room\ temp.]{H_2O}$$
- (d)
$$C_6H_5NH_2 + HCl(aq) \longrightarrow$$
- Q20.** How will you convert?
- (a) Propionamide to Ethyl amine?
- (b) Aniline to Phenol
- (c) Aniline to Acetanilide.