

## CBSE Class 12 Chemistry Chapter 10 Haloalkanes and Haloarenes Worksheet – Set 3

- Q1. n-propyl bromide reacts with ethanolic KOH to form
- (a) Propane
- (b) Propene
- (c) Propyl alcohol
- (d) None of the above
- Q2. The formation of alkane by the action of zinc on alkyl halide is known as
- (a) Wurtz reaction
- (b) Kolbe's reaction
- (c) Frankland's reaction
- (d) None of the above
- Q3. Chloroform, when treated with benzene in the presence of anhydrous AlCl<sub>3</sub>, the product formed is
- (a) Chloro benzene
- (b) A mixture of ortho and para chloro toluene
- (c) Tri phenyl methane
- (d) None of the above
- Q4. Phosgene is a common name for
- (a) Carbon dioxide and phosphane
- (b) Phosphoryl chloride
- (c) Carbonyl chloride
- (d) None of the above
- Q5. The reactivity of alkyl halides depends upon
- (a) The nature of the alkyl group only
- (b) The nature of the halogen atom only
- (c) The nature of the alkyl group and halogen atoms
- (d) None of the above
- **Q6.** Write the IUPAC names of the following compound.



- Q7. How will you convert benzene to 3-bromo nitro benzene?
- **Q8.** What will happen when bromine attacks  $CH_2 = CH CH_2 C = CH$ ?
- **Q9.** How will you distinguish between C<sub>2</sub>H<sub>5</sub>Br and C<sub>6</sub>H<sub>5</sub>Br? Also, write the chemical equation involved.
- **Q10.** How will you distinguish between carbon tetrachloride and chloroform? Also, write the chemical equation involved.
- **Q11.** Justify the optical activity of 2 Chloro butane.
- Q12. What is an ambident nucleophile? Explain with an example.
- **Q13.** What is beta elimination?
- Q14. What happens when chlorobenzene is treated with Cl<sub>2</sub> / FeCl<sub>3</sub>?
- **Q15.** Why is Grignard reagent prepared under anhydrous conditions?
- **Q16.** What happens when n-butyl chloride is treated with alcoholic KOH?
- **Q17.** What happens when methyl bromide is treated with sodium in the presence of dry ether?
- Q18. How would you differentiate between SN<sub>1</sub> and SN<sub>2</sub> mechanisms of substitution reactions?
- **Q19.** How will you convert propene to 1-iodopropane?
- Q20. How will you differentiate between an alkyl halide and an aryl halide?