

Chemistry Worksheet Class 11 on Chapter 13 Hydrocarbons - Set 4

Q-1: Which of the following reactions is used to determine the double bond position in unsaturated compounds?

- a) Pyrolysis
- b) Aromatisation
- c) Hydrogenation
- d) Ozonolysis
- Q-2: Conformation arises due to rotation around
- a) C-C double bond
- b) C-C single bond
- c) C-C triple bond
- d) All of the above

Q-3: Which of the following happens in the Kharasch effect?

- a) Free radical substitution
- b) Electrophilic addition
- c) Free radical addition
- d) Nucleophilic addition

Q-4: $C_6H_6 + X + AICI_3 \rightarrow Toluene, X is$

- a) Acetic acid
- b) Chloromethane
- c) Acetone
- d) Acetic anhydride

Q-5: Describe the procedure for making dihydrogen gas for industrial applications.

Q-6: Why are electrophilic reagents able to attack alkenes so quickly?

Q-7: How are alkynes distinguished from alkenes and alkanes? Describe using the reactions as support.

Q-8: Why are naphthalene balls used in toilets and for clothing preservation?

Q-9: a) Write the general formula for alkynes.

b) Write down the IUPAC name and structures of all alkynes having the molecular formula C_5H_8 .

Q-10: What is the end result of passing propyne through a hot iron tube at 873 K?

Q-11: Which of the following is aromatic in nature?

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- **Q-12:** Write the products of the following reaction.
- a) Hydrogenation of benzene at high temperatures and pressures in the presence of a Ni catalyst.
- b) Chlorination of benzene in the presence of ultraviolet light.
- c) Chlorination of benzene in the dark.
- **Q-13:** Give any two methods for preparing ethyne.
- Q-14: How do structural isomers and stereoisomers differ from each other?
- Q-15: Why does cis isomer have a lower melting point than trans isomer?

Q-16: How can we discuss geometrical isomerism in the compounds where distinct atoms or groups of atoms are bound to the double-bonded carbon atoms?

- **Q-17:** State the Saytzeff rule.
- Q-18: What is Wurtz reaction? Give an example of the reaction.
- Q-19: Define aromaticity using a few examples.

Q-20: a) Which polymer can be used as electrodes in batteries? And why?

b) What happens when acetylene is passed through dil. H₂SO₄ in the presence of HgSO₄?