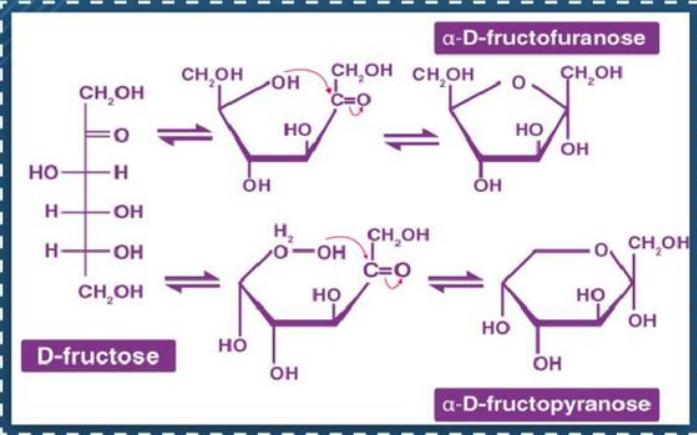


BIOMOLECULES-L2



ZOOLOGY | CLASS 11

PUSHPENDU SIR



MISSION MBBS

MONDAY TO FRIDAY
4 PM - 8 PM



PUSHPENDU SIR
ZOOLOGY

SACHIN SIR
ZOOLOGY

VIVEK SIR
CHEMISTRY

PANKHURI MA'AM
BOTANY

ANUSHRI MA'AM
PHYSICS

Take the Aakash BYJU'S All India NEET Mock Test 2022



- ° India's largest NEET mock test



- ° Self-Assessment from additional sample papers and detailed solutions

- ° Based on NEET 2022 Pattern



- ° All India Rank (AIR) Projection

- ° Designed by

Aakash BYJU'S experts

Registration Date

8th June
Onwards

Exam Dates and Time

3rd & 10th July

2:00 to 5:20 PM

Register Now

(Link in Description)

FREE FOR 14 DAYS!

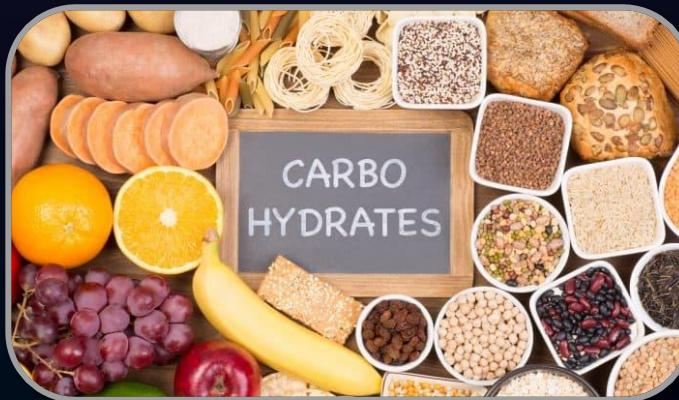


Aakash
+  **BYJU'S**

Recall! Carbohydrates

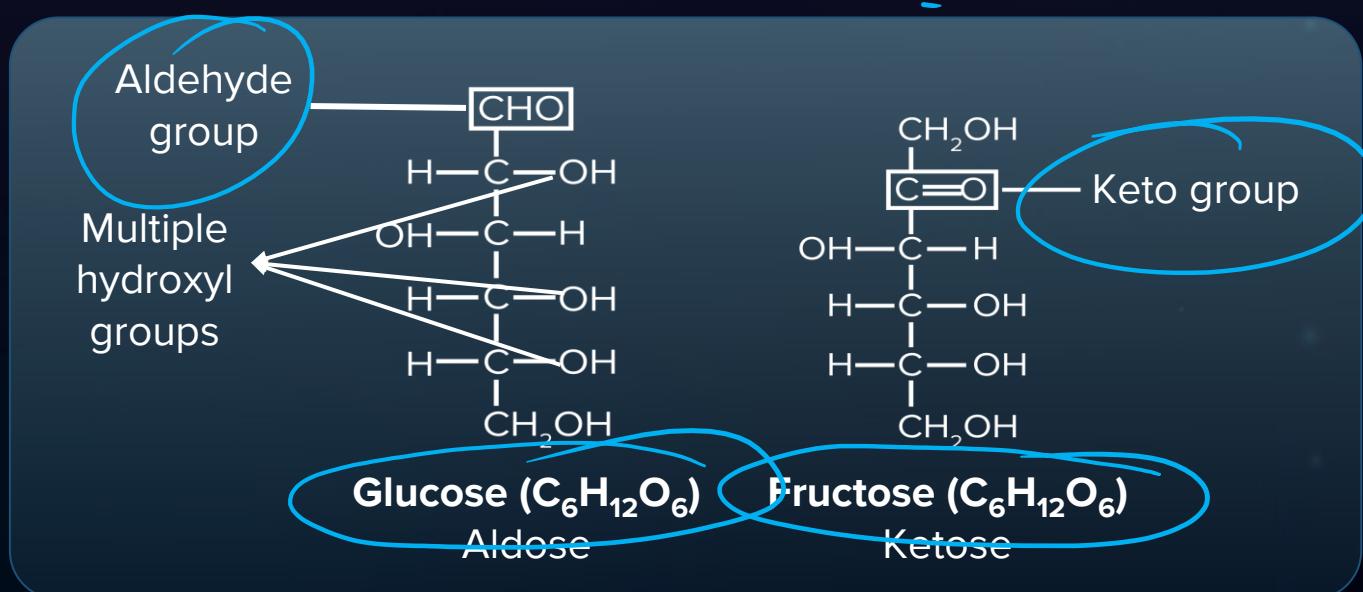
Aldehyde group

- 'Carbohydrates' = Hydrates of carbon
- Compounds having carbon, hydrogen and oxygen in the ratio 1:2:1



Carbohydrates: General Structure

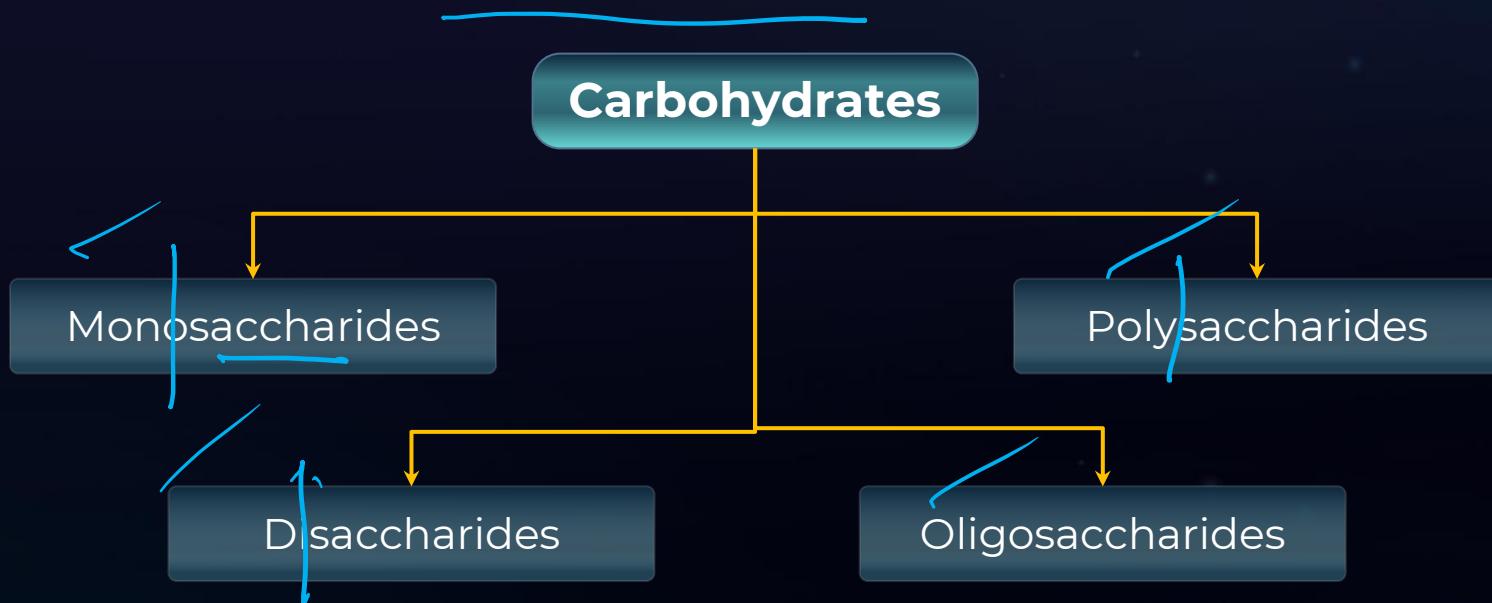
- Atleast **3 carbon** atoms
- Multiple hydroxyl (-OH) groups
- Either an aldehyde (-CHO) group or a ketone (C=O) group



Classification of Carbohydrates

Classification of Carbohydrates

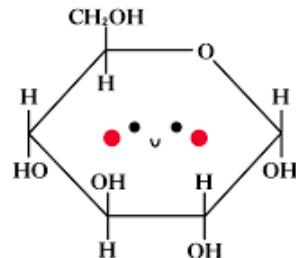
Based on the number of monomeric units



Monosaccharides

Monosaccharides

IM SO SWEET!

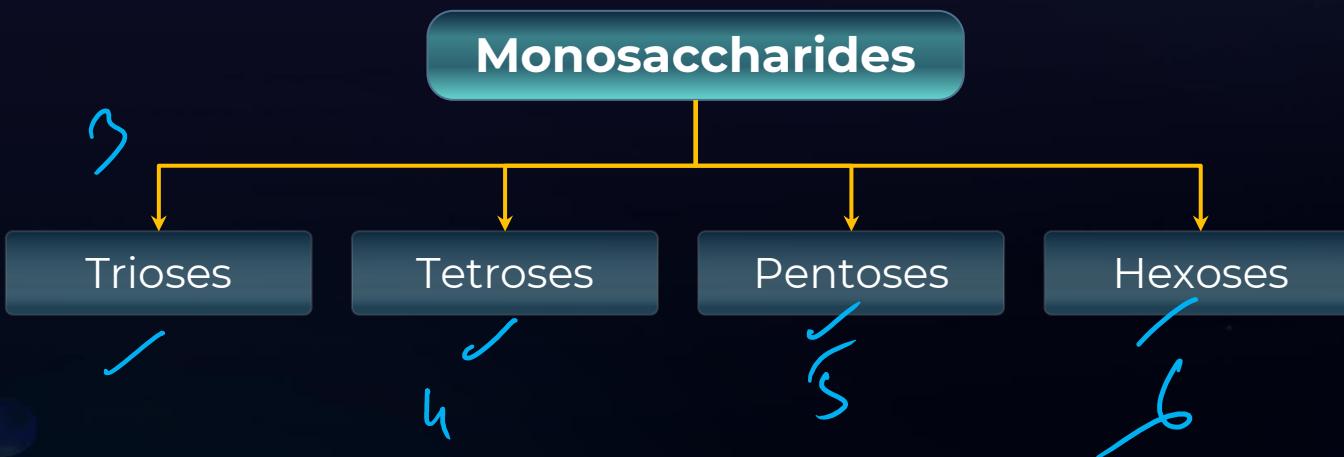


- Mono = Single; Saccharide = Sugar
- **Simplest** carbohydrates
- **Building blocks** of larger carbohydrates

cannot be
hydrolysed.

Monosaccharides

Based on the **number of carbon atoms**



Monosaccharides

Monosaccharides

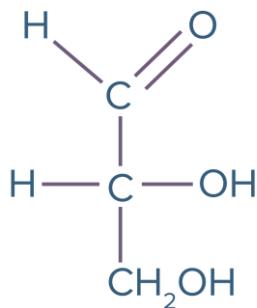
Trioses

Tetroses

Pentoses

Hexoses

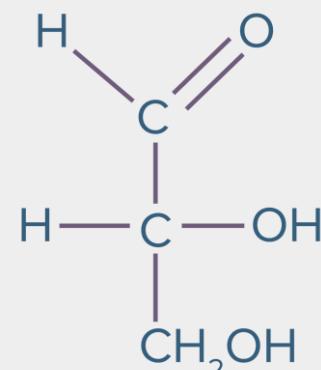
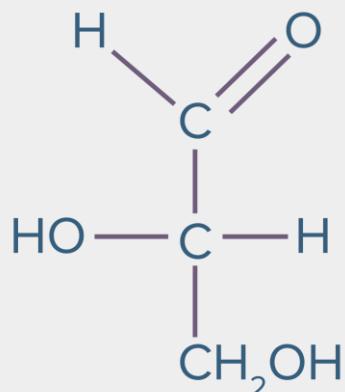
Have 3 carbon atoms



→ Glyceraldehyde

Optical Activity of Monosaccharides

Two Forms - Enantiomers



L-Glyceraldehyde

D-Glyceraldehyde

Why have we written the letters 'L' and 'D' in front of the name of the sugars?

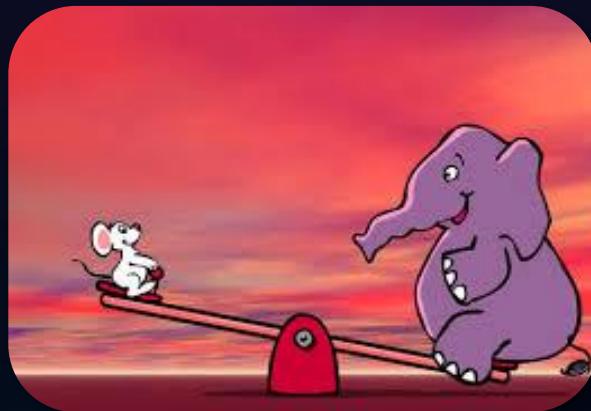




**That is a special property of
these sugars! Let us see
what it is!**

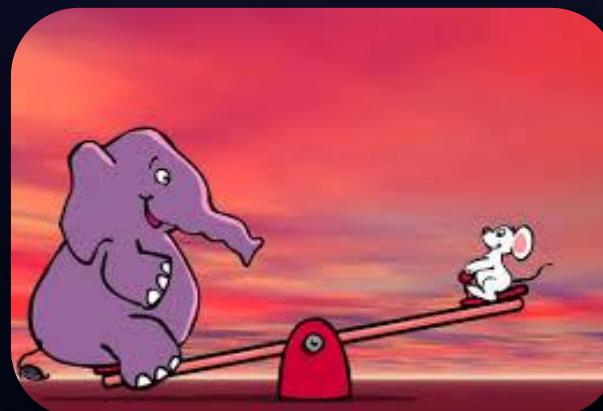
Monosaccharides - Glyceraldehyde

- Two **optically active** forms
- **Non-superimposable** mirror images



Dextrorotatory (d)

Turns the plane polarised light
to the **right**



Laevorotatory (l)

Turns the plane polarised light
to the **left**



Monosaccharides

Monosaccharides

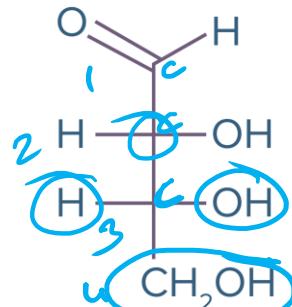
Trioses

Tetroses

Pentoses

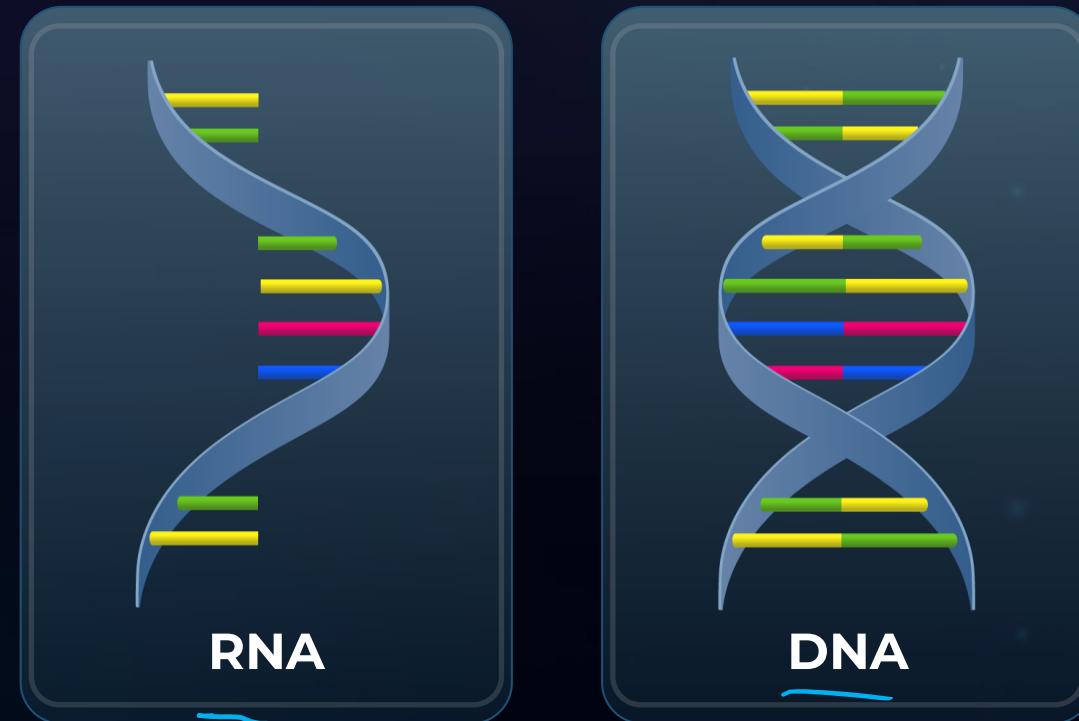
Hexoses

Have 4 carbon atoms



D. Erythrose

Do you know the type of sugar
present in RNA and DNA?



Monosaccharides

Monosaccharides

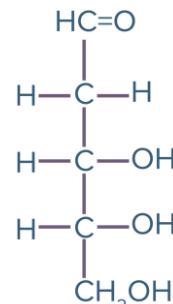
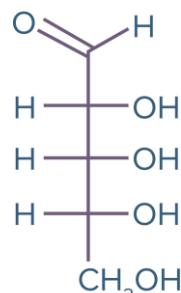
Trioses

Tetroses

Pentoses

Hexoses

Have 5 carbon atoms



D- Ribose D- Deoxyribose

Have you drunk this?



Monosaccharides

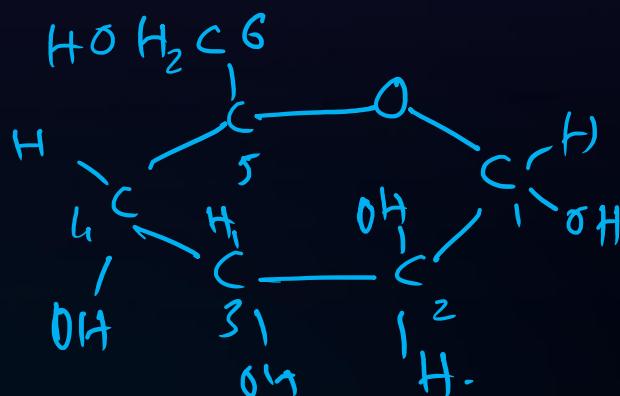
Monosaccharides

Trioses

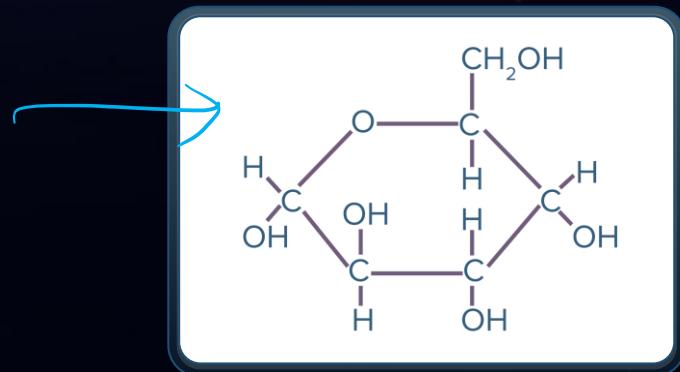
Tetroses

Pentoses

Hexoses

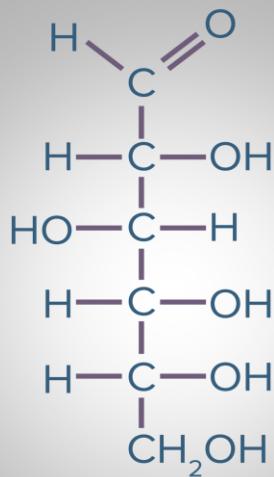


Have 6 carbon atoms

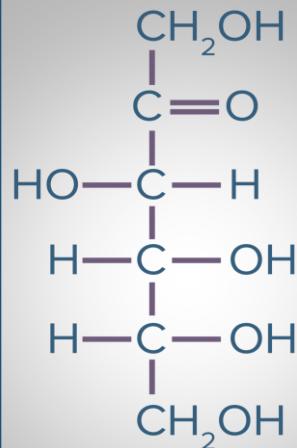


L-Glucose

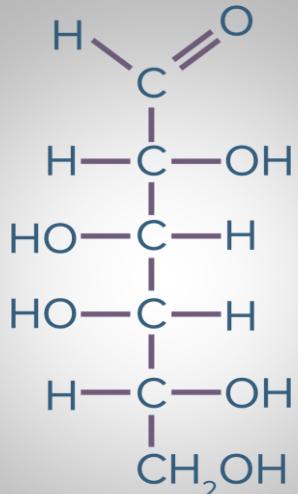
Monosaccharides - Hexoses



Glucose



D-Fructose



Galactose

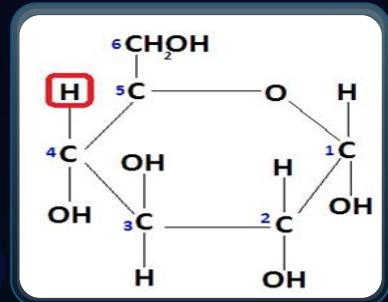
Monosaccharides also show
reducing property



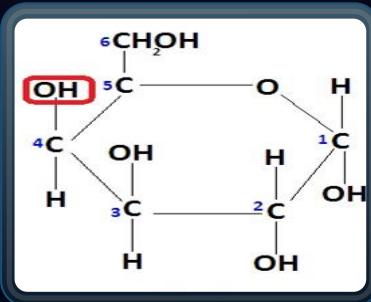
Monosaccharides

- Reducing sugars: **Free aldehyde/ketone group present**
- Non-reducing sugars: **No free aldehyde/ketone group**
- All monosaccharides are reducing sugars**

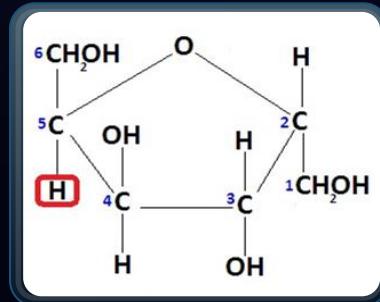
Hemiacetal
OR
R-O-C



Glucose



Galactose



Fructose



Let's test the sugars out!

Benedict's Test

Benedict's Test

Objective: To identify
whether a sugar is reducing
or non-reducing

Benedict's Test

Experiment



About 2 cm^3 of
test solution
(glucose)

Add an equal
amount of
Benedict's
solution



Heat in water
bath

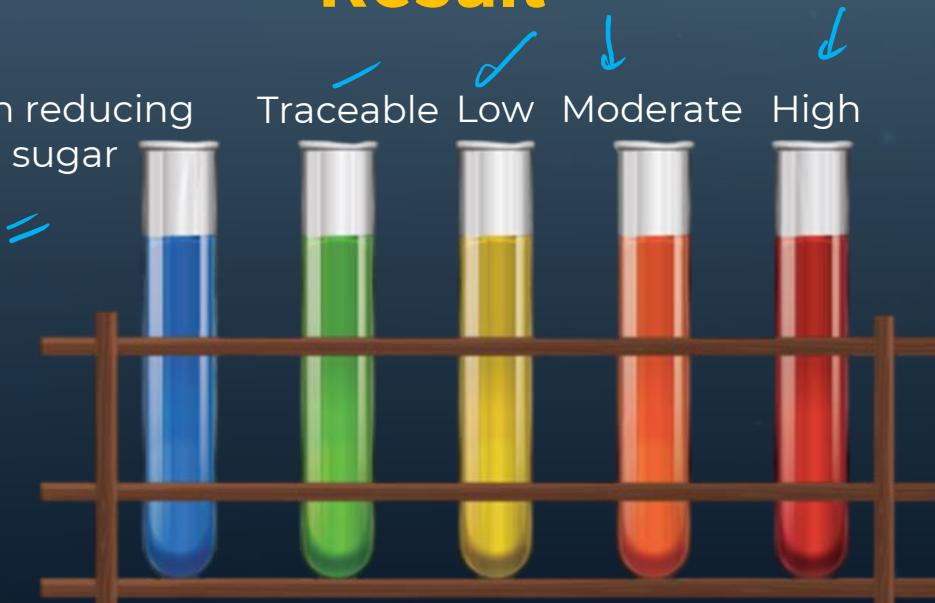


Brick red
precipitate

Benedict's Test

Result

Non reducing sugar Traceable Low Moderate High





QUESTION
TIME



Reducing sugars must have:



A Free aldehyde group only

B Free ketone group only

C Either free aldehyde or ketone group

D Both free aldehyde and ketone groups

Reducing sugars must have:

A

Free aldehyde group only

B

Free ketone group only

C

Either free aldehyde or ketone group

D

Both free aldehyde and ketone groups

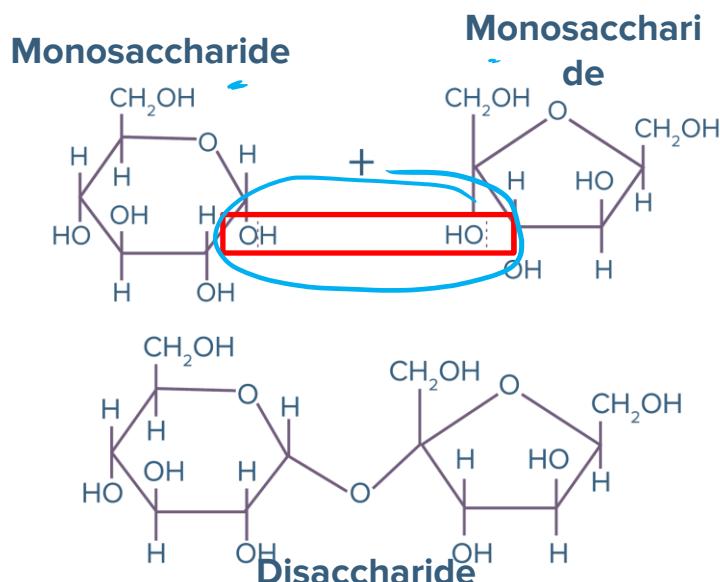


Let us see what happens
when two
monosaccharides unite.

Disaccharides

Disaccharides

- Di = Two; Saccharide = Sugar unit
- **Two monosaccharide** units join with a **glycosidic bond**



Disaccharides

Sucrose



Table sugar

Maltose



Maltose syrup

Lactose



Milk products

Disaccharides

| Disaccharide | Monomers | Reducing nature |
|----------------|----------------------------|--------------------|
| Sucrose | Glucose + <u>Fructose</u> | Non-reducing sugar |
| Maltose | Glucose + <u>Glucose</u> | Reducing sugar |
| Lactose | Glucose + <u>Galactose</u> | Reducing sugar |

**What happens when more
than two monosaccharides
join?**



Oligosaccharides

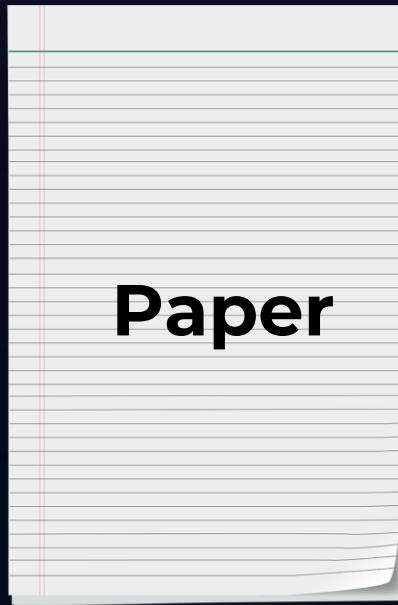
Oligosaccharides

- 'Oligo' = Few; 'Saccharide' = Sugar
- 3-9 monosaccharide units join by glycosidic linkage

| | |
|------------------------|--|
| Trisaccharide | 3 monosaccharide units Raffinose = Glucose + Galactose + Fructose |
| Tetrasaccharide | 4 monosaccharide units Stachyose = Glucose + Galactose + Galactose + Fructose |



Do you know the common factor
in both wood and paper?



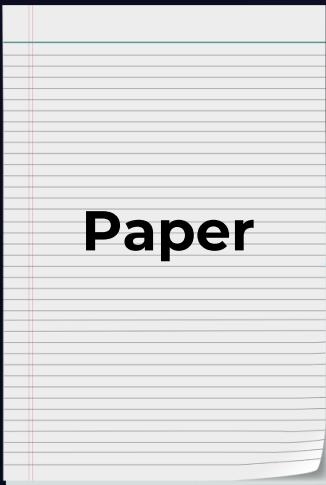
Paper



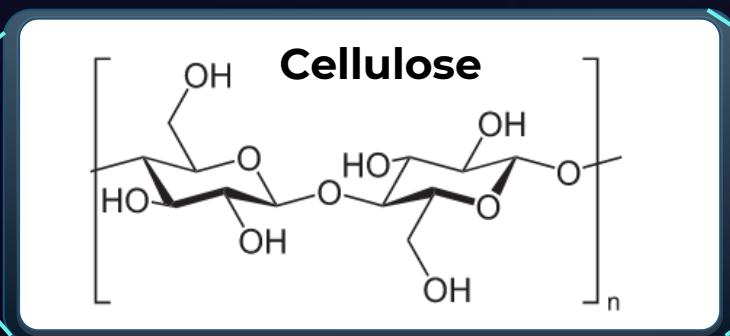
Wood



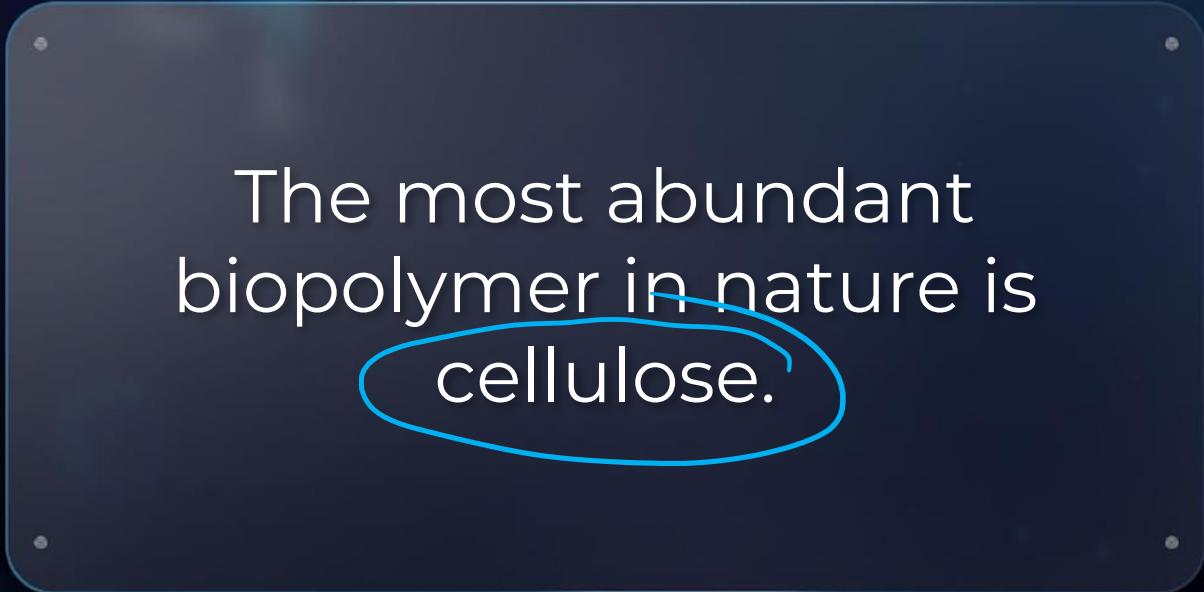
Both are derived from trees and are rich in cellulose



Paper



Wood



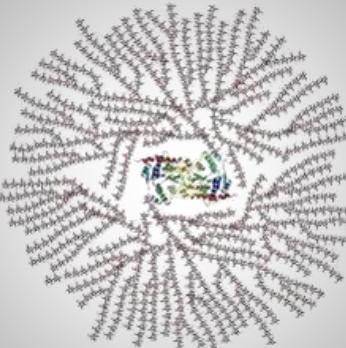
The most abundant
biopolymer in nature is
cellulose.

Polysaccharides



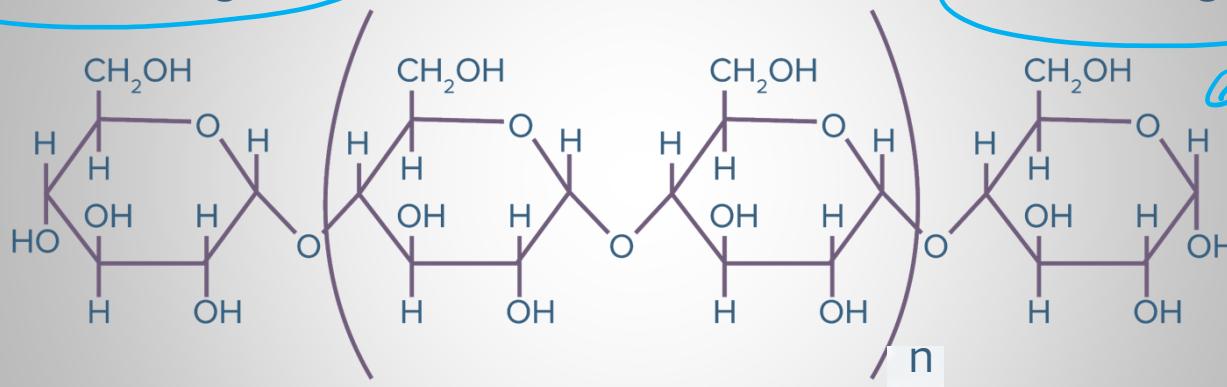
Polysaccharides

- Polymers of repeating units of monosaccharides
- Also called 'glycans'
- High molecular weight
- Macromolecules



Polysaccharides

Non-reducing end



Reducing end

Polysaccharides

Polysaccharide

Homopolysaccharide

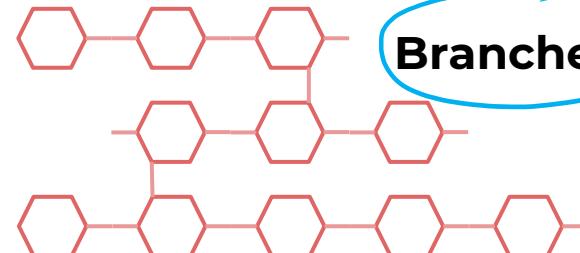
Heteropolysaccharide



Homopolysaccharide

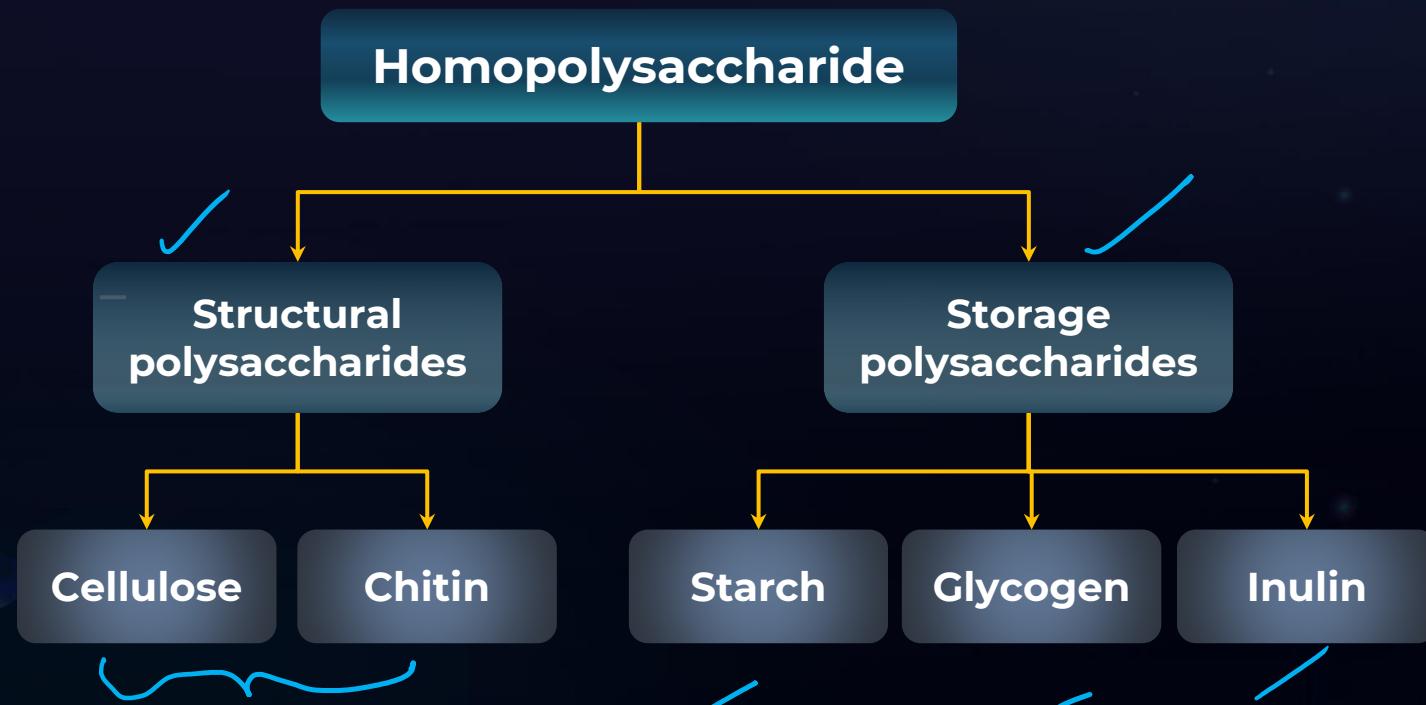


Unbranched

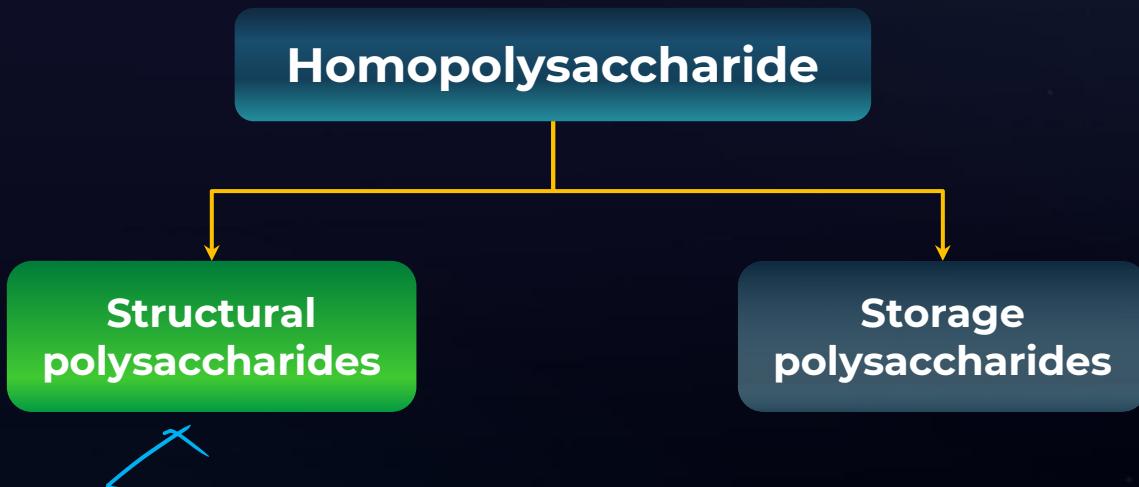


Branched

Homopolysaccharide



Homopolysaccharide



Structural polysaccharides - Cellulose



Cellulose in
plants

- Most **abundant biomolecule** on the earth
- Forms the **structural part** of the living organisms
- Polymer of **glucose units**
- Straight chain, Unbranched

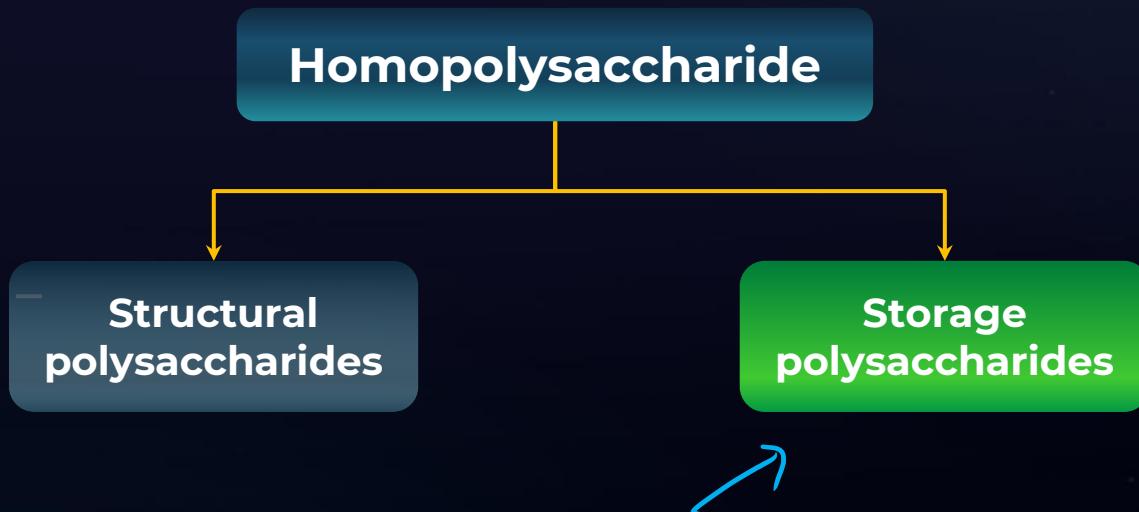
Structural polysaccharides - Chitin



Chitin in exoskeleton
of insects

- **Second most** abundant organic substance
- Polymer of **N-acetyl glucosamine**

Homopolysaccharide



Storage Polysaccharide: Starch



- **Polymer of glucose**
- Food reserve in plants
- Consists of:
 - **Amylose - 20% - 30%**
 - **Amylopectin- 70% - 80%**

Storage Polysaccharide: Starch

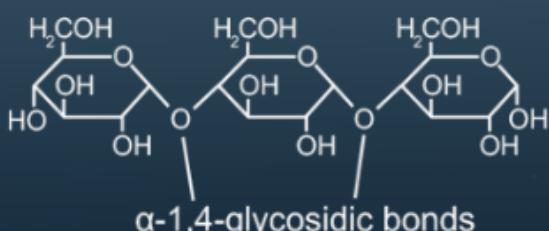
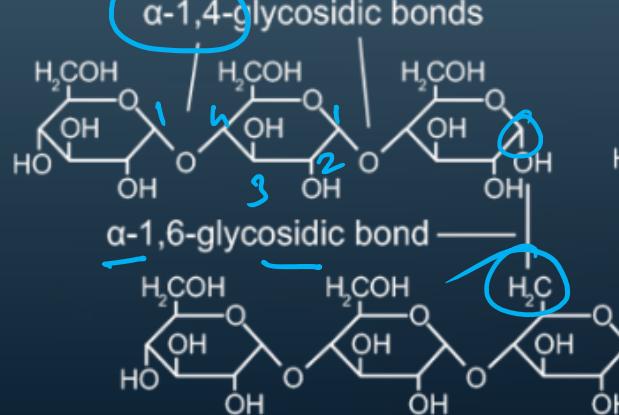
amylopectin



amylose



O = single glucose unit



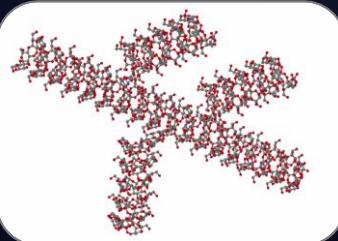


Did you know?

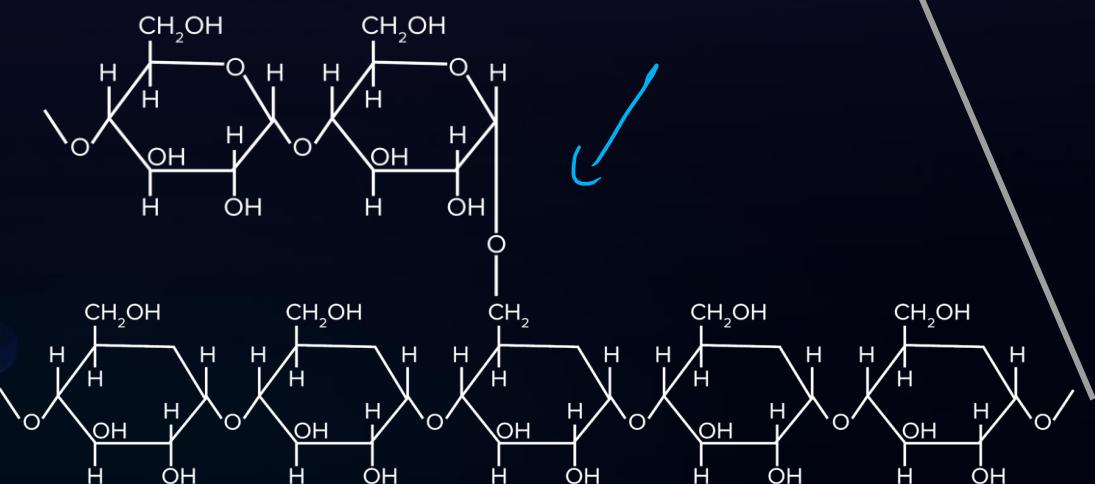


The starch helices in potato can hold Iodine and it turns blue-black.

Storage Polysaccharide: Glycogen



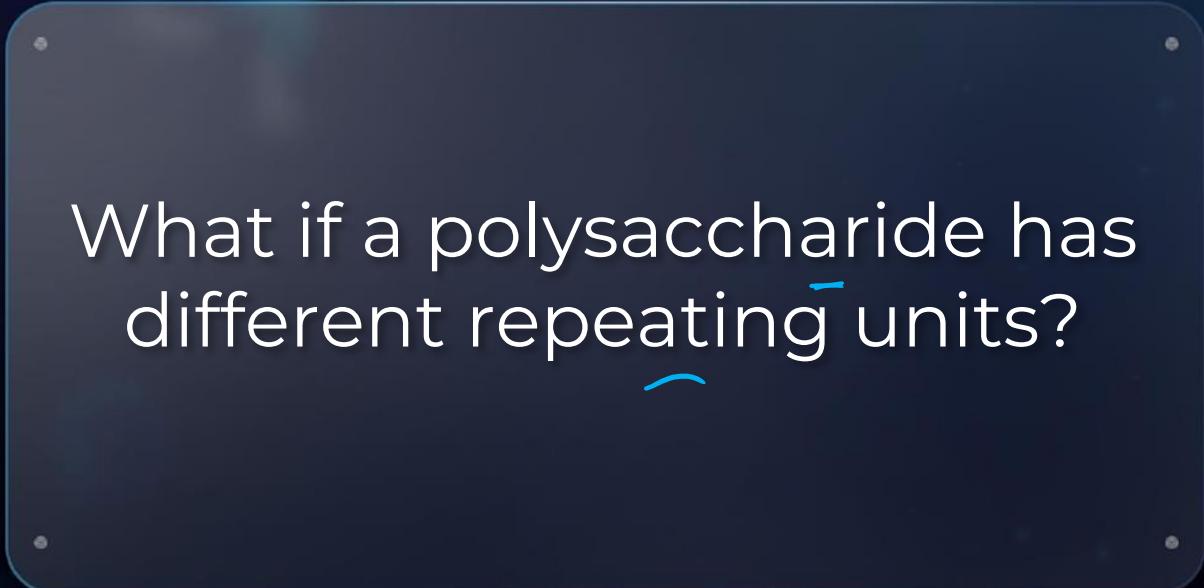
- Polymer of **glucose**
- Food reserve in animals
- Stored in **liver and muscles**
- **Highly branched**



Storage Polysaccharide: Inulin



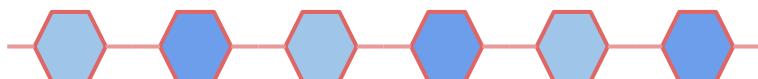
- Polymer of **fructose**
- Stored in roots of *Dahlia*, Dandelion and Artichoke
- **Not metabolised** in human body



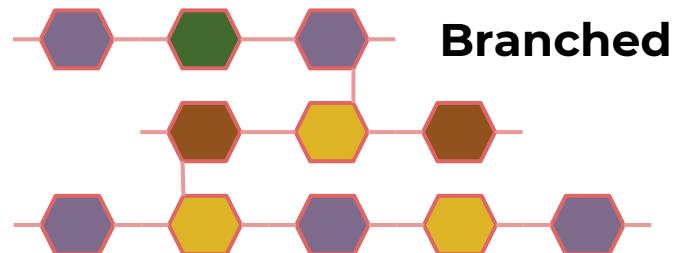
What if a polysaccharide has
different repeating units?

Heteropolysaccharide

Heteropolysaccharide



Unbranched



Branched

Heteropolysaccharide

Heteropolysaccharide

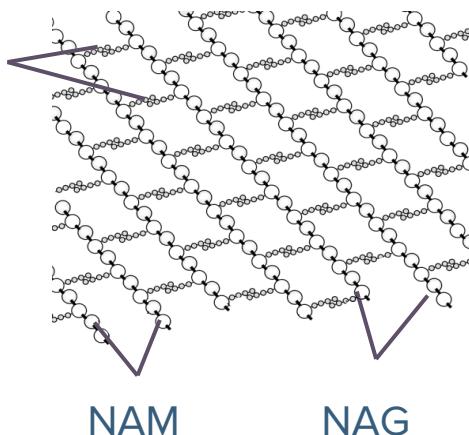
Peptidoglycan

Agar



Heteropolysaccharide: Peptidoglycan

Polypeptide



- Component of bacterial cell wall
- Made of two different kind of repeating units
 - **N- Acetyl glucosamine (NAG)**
 - **N-Acetyl muramic acid (NAM)**

Heteropolysaccharide: Agar



- Obtained from the red algae: *Gelidium* and *Gracilaria*
- Made of two different units:
 - Galactose
 - 3,6-anhydro-L-galactopyranose



Past Year Questions



Which among the following is not a reducing sugar?(NEET-2014)

A Maltose

B Lactose

C Ribose

D Sucrose



Which among the following is not a reducing sugar?(AIPMT-2014)

A

Maltose

B

Lactose

C

Ribose

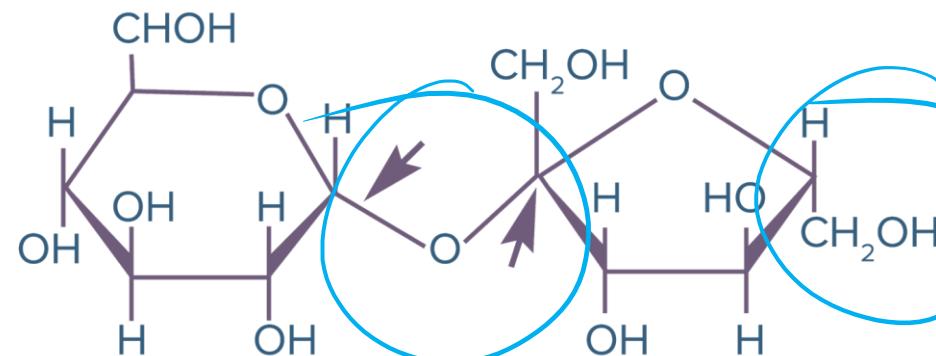
D

Sucrose



Discussion

- Glucose + Fructose = Sucrose
- Non-Reducing sugar as there is **no free aldehyde or ketone group**





Macromolecule in chitin is (NEET-2013)

A

nitrogen containing
polysaccharide

B

phosphorus containing
polysaccharide

C

Sulphur containing
polysaccharide

D

Simple polysaccharide



Macromolecule in chitin is (NEET-2013)

A

**nitrogen containing
polysaccharide**

B

phosphorus containing
polysaccharide

C

Sulphur containing
polysaccharide

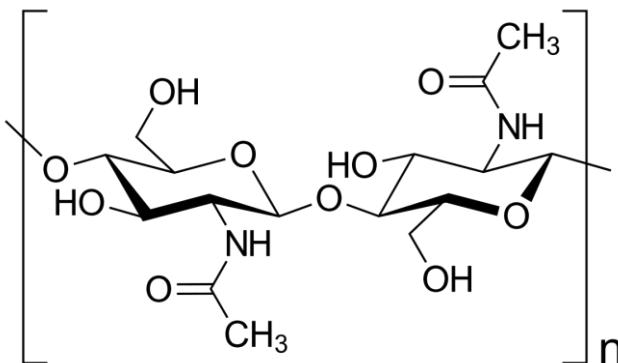
D

Simple polysaccharide



Discussion

Chitin

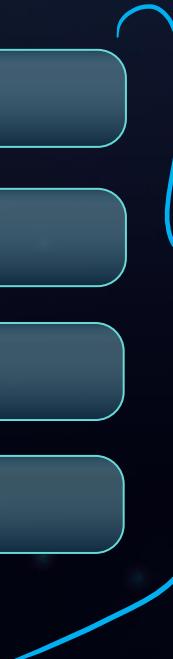


- Long-chain polymer of a N-acetylglucosamine
- Nitrogen containing polysaccharide
and a derivative of glucose

Summary

What we have learnt so far

- 1 Monosaccharides
- 2 Disaccharides
- 3 Oligosaccharides
- 4 Polysaccharides



A dark blue background featuring a laboratory setup on the left with glassware containing blue liquid and a molecular model. A central white speech bubble contains the text "Keep Learning!"

Keep
Learning!