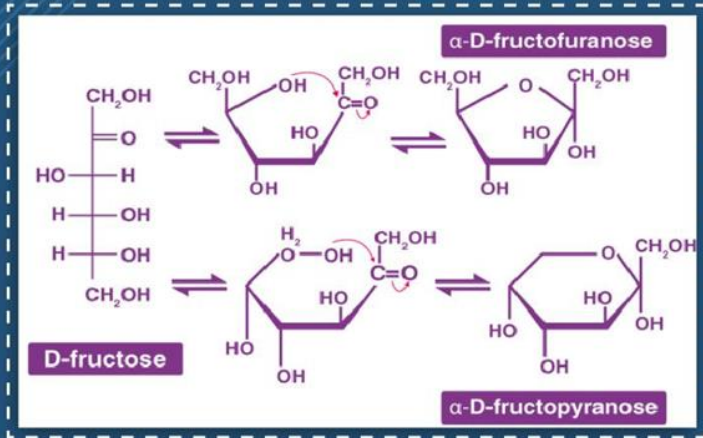


CARBOHYDRATES

MISSION MBBS | 2024



BIOMOLECULES-L2



ZOOLOGY | CLASS 11

PUSHPENDU SIR



MONDAY TO FRIDAY
4 PM - 8 PM



PUSHPENDU SIR
ZOOLOGY



SACHIN SIR
ZOOLOGY



VIVEK SIR
CHEMISTRY



PANKHURI MA'AM
BOTANY



ANUSHRI MA'AM
PHYSICS

Take the Aakash BYJU'S All India NEET Mock Test 2022



° **India's largest**
NEET mock test

° **Self-Assessment**
from additional sample papers
and detailed solutions

° **Based on**
NEET 2022 Pattern

° **All India Rank (AIR)**
Projection

° **Designed by**
Aakash BYJU'S experts

Register Now

(Link in Description)

Registration Date

8th June
Onwards

Exam Dates and Time

3rd & 10th July

2:00 to 5:20 PM

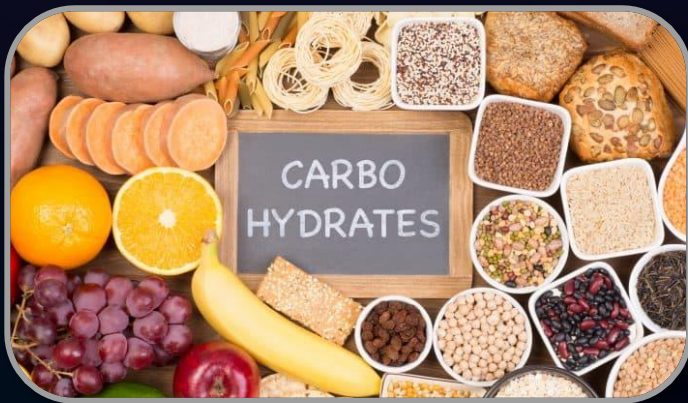
FREE FOR 14 DAYS!



Recall! Carbohydrates

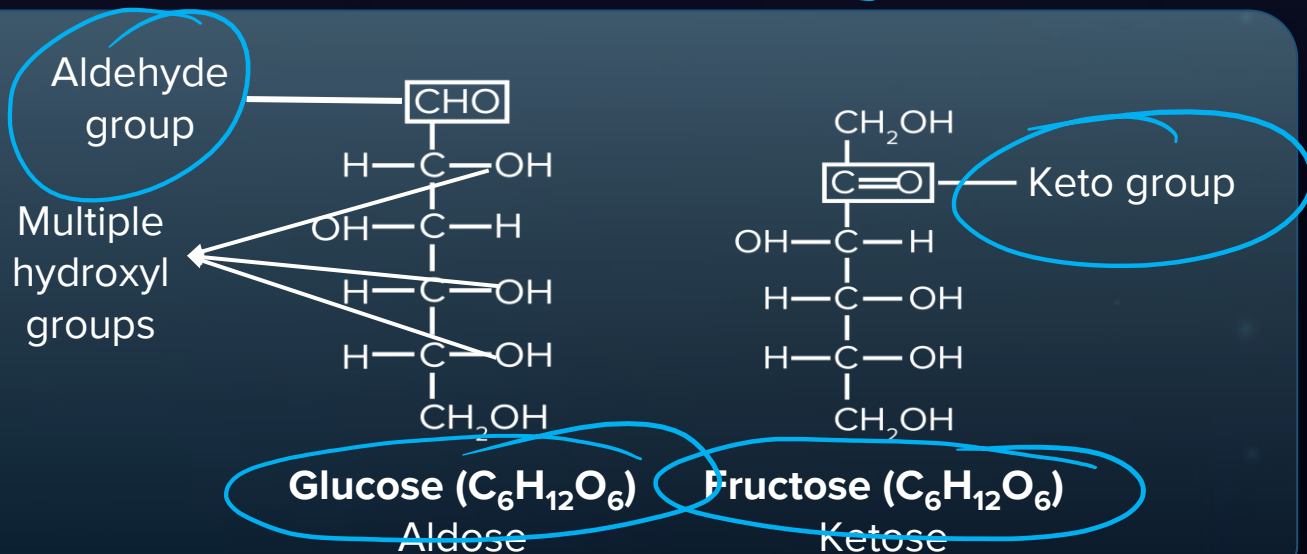
Aldehyde
group

- 'Carbohydrates' = Hydrates of carbon
- Compounds having carbon, hydrogen and oxygen in the ratio **1:2:1**



Carbohydrates: General Structure

- Atleast **3 carbon** atoms
- Multiple hydroxyl (-OH) groups
- Either an aldehyde (-CHO) group or a ketone (C=O) group

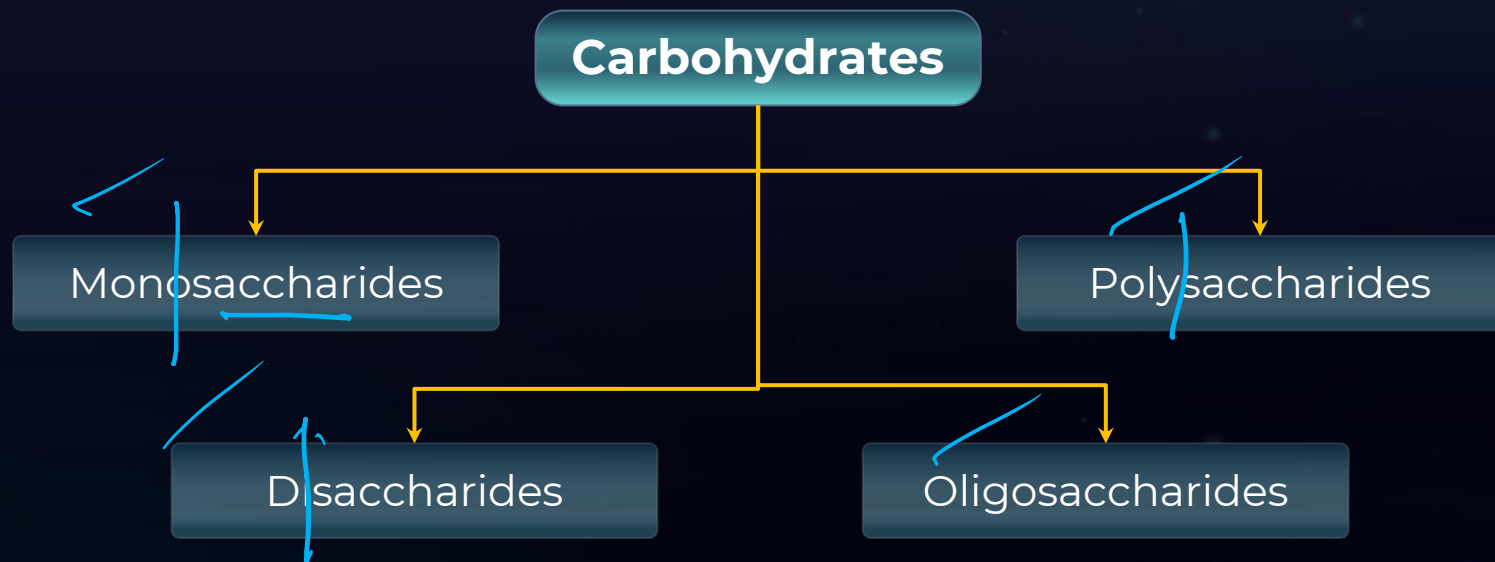




Classification of Carbohydrates

Classification of Carbohydrates

Based on the number of monomeric units

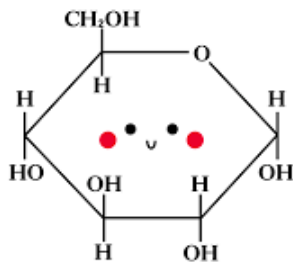




Monosaccharides

Monosaccharides

I'M SO SWEET!



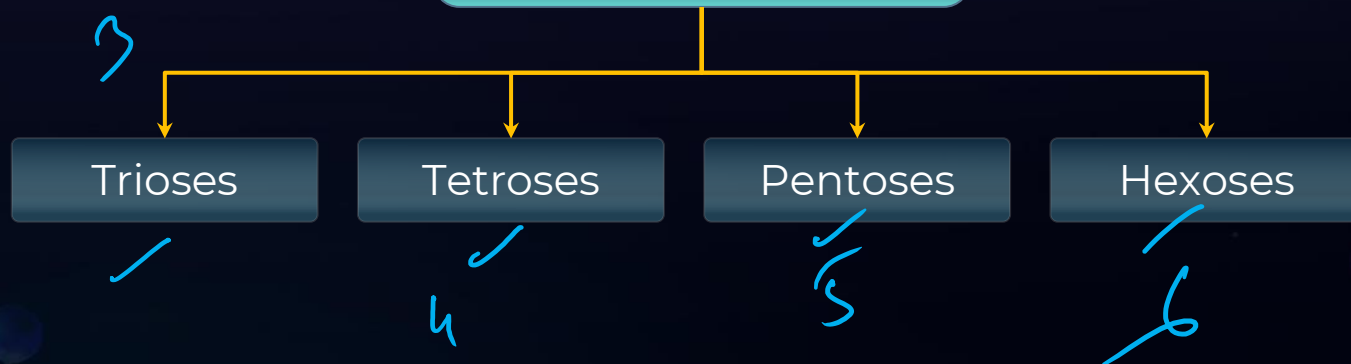
- Mono = Single; Saccharide = Sugar
- **Simplest** carbohydrates
- **Building blocks** of larger carbohydrates

*Cannot be
hydrolysed.*

Monosaccharides

Based on the **number of carbon atoms**

Monosaccharides



Monosaccharides

Monosaccharides

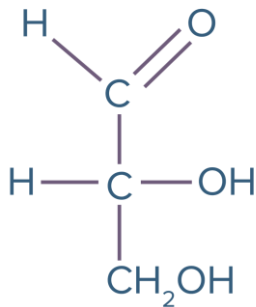
Trioses

Tetroses

Pentoses

Hexoses

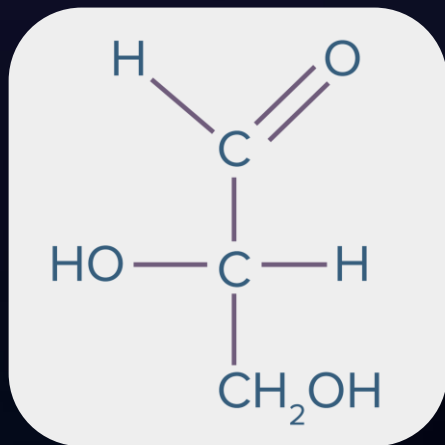
Have 3 carbon atoms



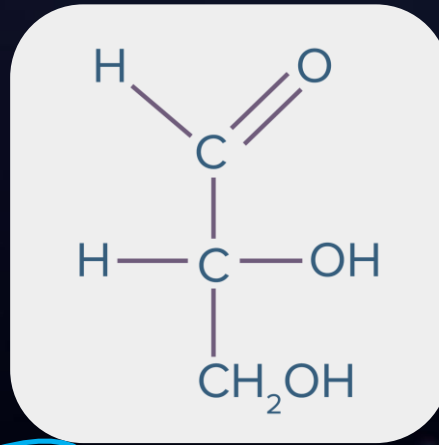
Glyceraldehyde

Optical Activity of Monosaccharides

Two Forms - Enantiomers



L- Glyceraldehyde



D- Glyceraldehyde

Why have we written the
letters 'L' and 'D' in front of
the name of the sugars?

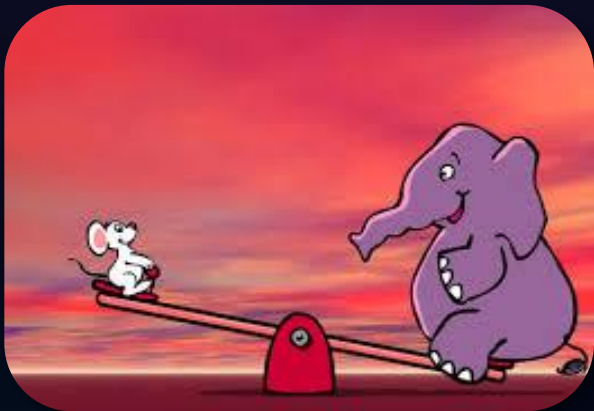




That is a special property of
these sugars! Let us see
what it is!

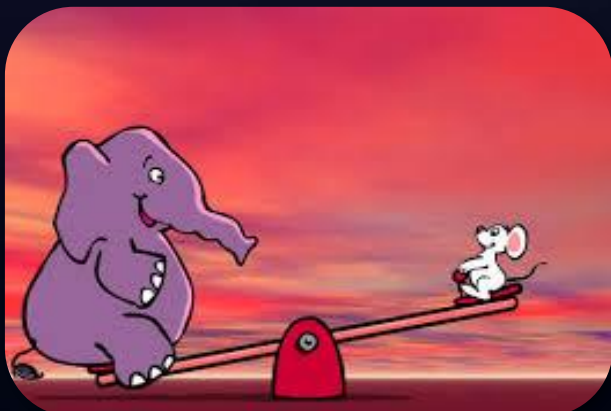
Monosaccharides - Glyceraldehyde

- Two **optically active** forms
- **Non-superimposable** mirror images



Dextrorotatory (d)

Turns the plane polarised light
to the **right**



Laevorotatory (l)

Turns the plane polarised light
to the **left**



Monosaccharides

Monosaccharides

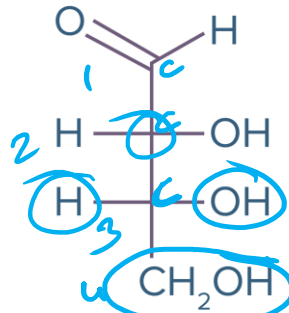
Trioses

Tetroses

Pentoses

Hexoses

Have 4 carbon atoms



D- Erythrose

**Do you know the type of sugar
present in RNA and DNA?**



RNA



DNA

Monosaccharides

Monosaccharides

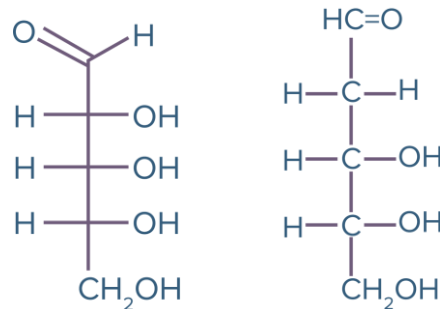
Trioses

Tetroses

Pentoses

Hexoses

Have 5 carbon atoms



D- Ribose **D- Deoxyribose**

Have you drunk this?



Monosaccharides

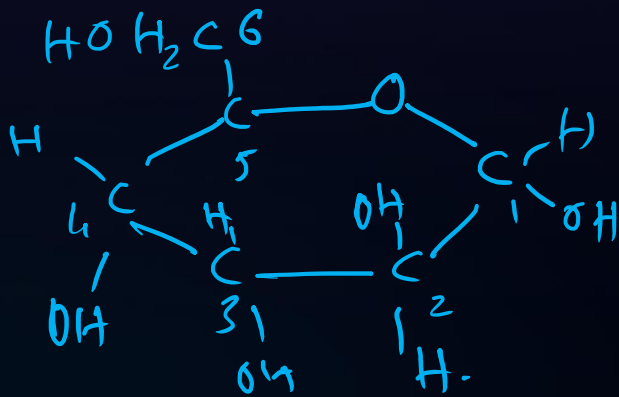
Monosaccharides

Trioses

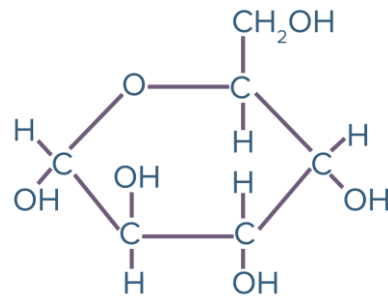
Tetroses

Pentoses

Hexoses

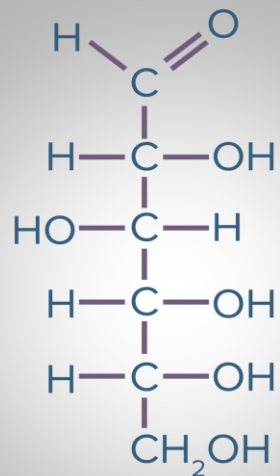


Have 6 carbon atoms

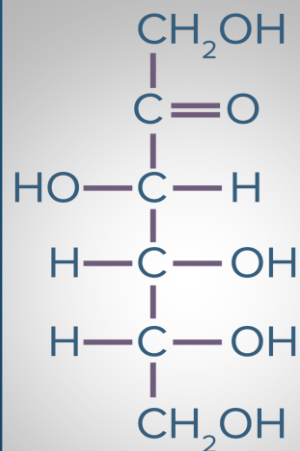


L-Glucose

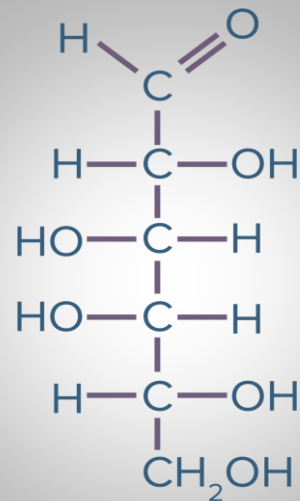
Monosaccharides - Hexoses



Glucose



D-Fructose



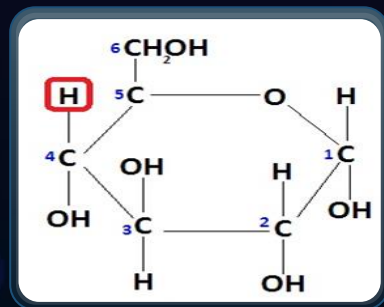
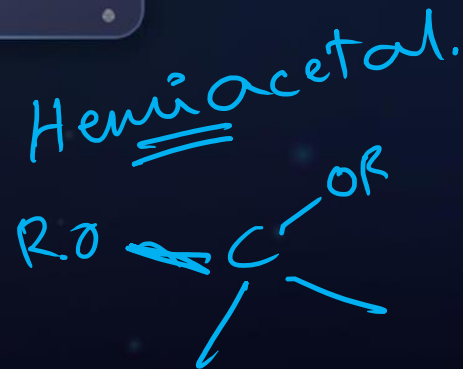
Galactose

Monosaccharides also show
reducing property

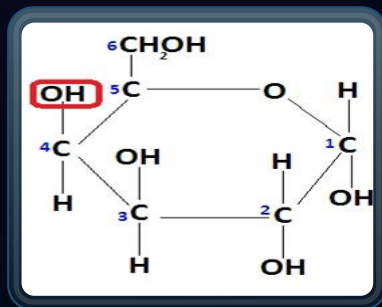


Monosaccharides

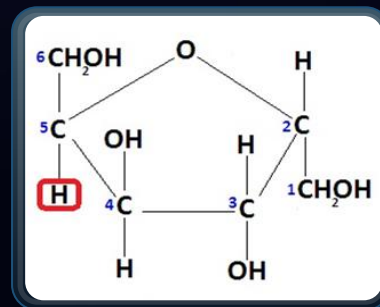
- Reducing sugars: **Free aldehyde/ketone group present**
- Non-reducing sugars: **No free aldehyde/ketone** group
- All monosaccharides are reducing sugars**



Glucose



Galactose



Fructose



Let's test the sugars out!



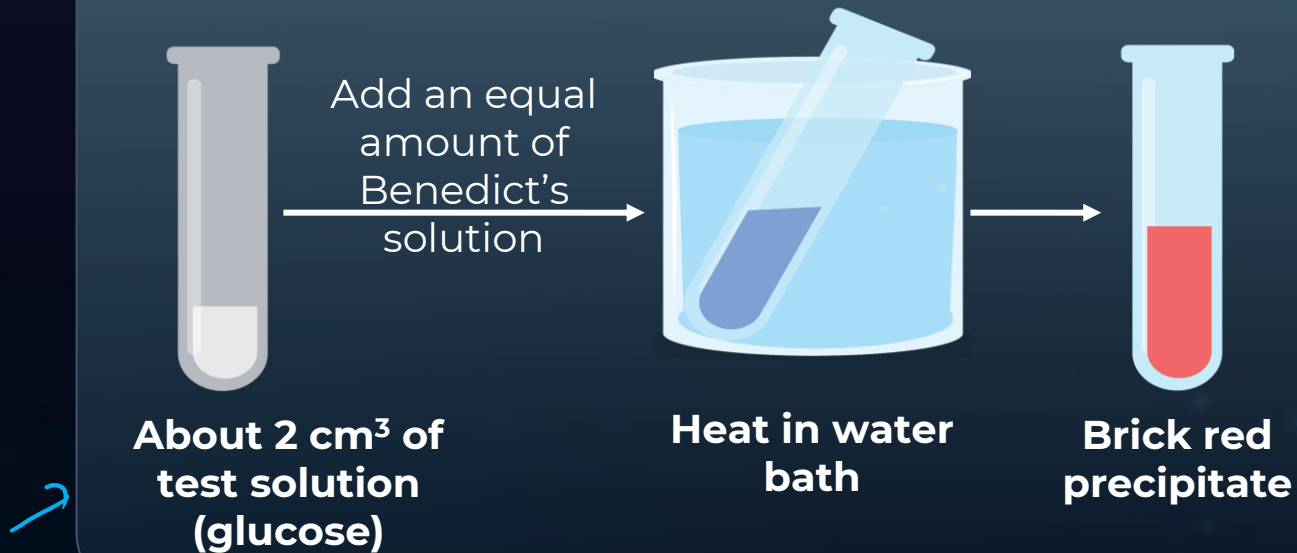
Benedict's Test

Benedict's Test

Objective: To identify whether a sugar is reducing or non-reducing

Benedict's Test

Experiment



Benedict's Test

Result

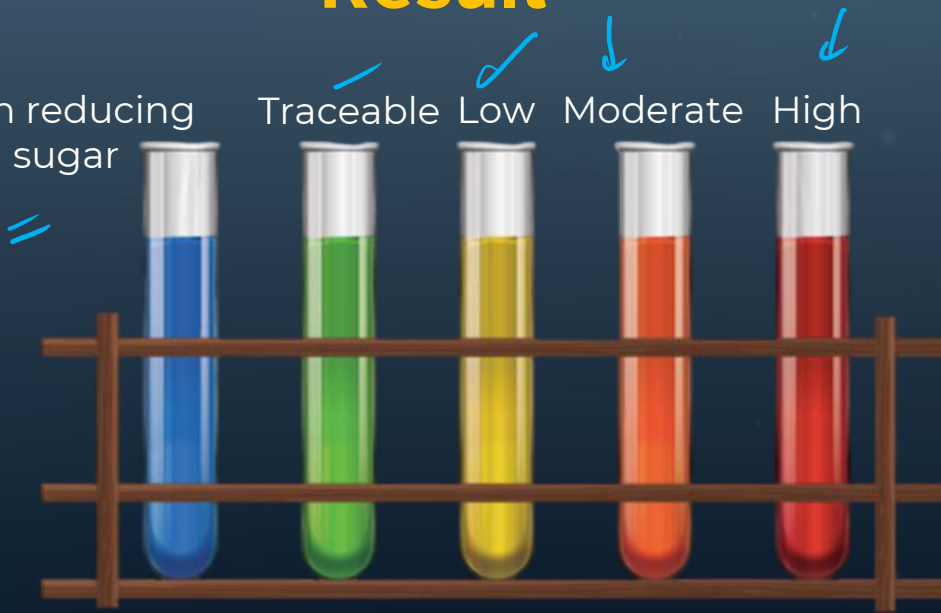
Non reducing
sugar

Traceable

Low

Moderate

High





**QUESTION
TIME**



Reducing sugars must have:



A

Free aldehyde group only

B

Free ketone group only

C

Either free aldehyde or ketone group

D

Both free aldehyde and ketone groups



Reducing sugars must have:

A

Free aldehyde group only

B

Free ketone group only

C

Either free aldehyde or ketone group

D

Both free aldehyde and ketone groups



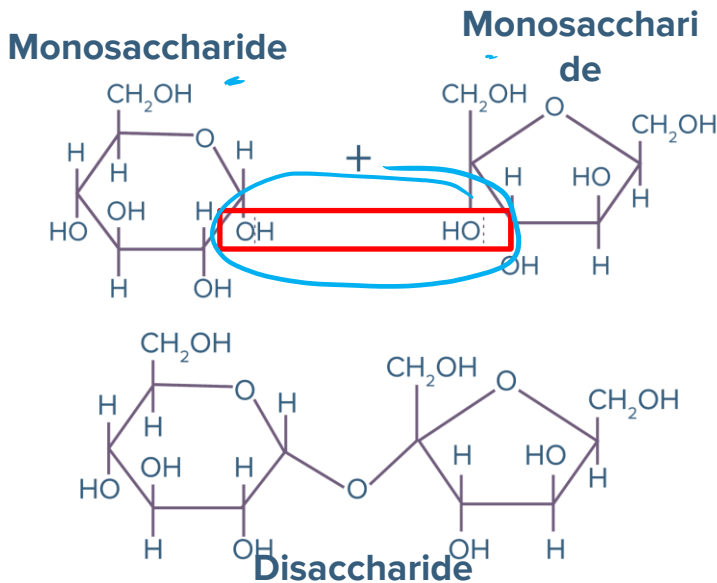
Let us see what happens
when two
monosaccharides unite.



Disaccharides

Disaccharides

- Di = Two; Saccharide = Sugar unit
- **Two monosaccharide** units join with a **glycosidic bond**



Disaccharides

Sucrose



Table sugar



Maltose



Maltose syrup



Lactose



Milk products



Disaccharides

Disaccharide	Monomers	Reducing nature
Sucrose	Glucose + <u>Fructose</u>	Non-reducing sugar
<u>Maltose</u>	Glucose + <u>Glucose</u>	Reducing <u>sugar</u>
<u>Lactose</u>	Glucose + <u>Galactose</u>	Reducing <u>sugar</u>

What happens when more
than two monosaccharides
join?



A molecular model in the top-left corner showing a central blue sphere (likely carbon) bonded to three white spheres (likely hydrogen) and one other blue sphere, all on a dark blue background with faint molecular structures.

Oligosaccharides

A molecular model in the bottom-right corner showing a blue sphere (likely carbon) bonded to two white spheres (likely hydrogen) and another blue sphere, all on a dark blue background with faint molecular structures.

Oligosaccharides

- 'Oligo' = Few; 'Saccharide' = Sugar
- 3-9 monosaccharide units join by glycosidic linkage

<u>Trisaccharide</u>	3 <u>monosaccharide</u> units Raffinose = <u>Glucose</u> + <u>Galactose</u> + <u>Fructose</u>
<u>Tetrasaccharide</u>	4 <u>monosaccharide</u> units Stachyose = <u>Glucose</u> + <u>Galactose</u> + <u>Galactose</u> + <u>Fructose</u>



Do you know the common factor
in both wood and paper?



Paper



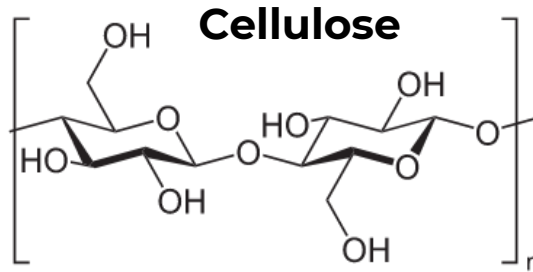
Wood




Both are derived from trees and are rich in cellulose




Paper



Wood



The most abundant
biopolymer in nature is
cellulose.

A dark blue background with faint molecular structures. In the top-left corner, a water molecule is shown with a large blue sphere (oxygen) and two smaller white spheres (hydrogen). In the bottom-right corner, another water molecule is partially visible, also with a blue oxygen sphere and white hydrogen spheres. The text is centered in a dark blue rounded rectangle.



Polysaccharides



Polysaccharides

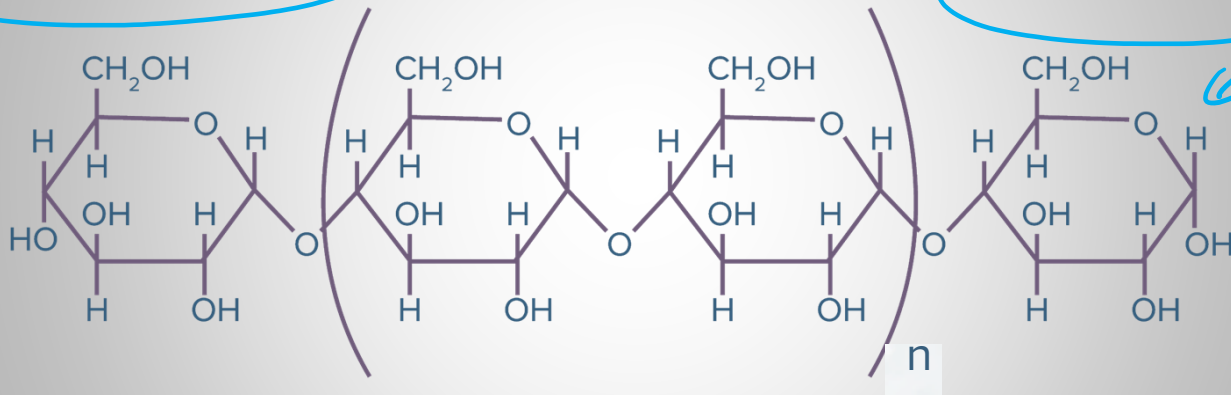
- Polymers of repeating units of monosaccharides
- Also called 'glycans'
- High molecular weight
- Macromolecules



Polysaccharides

Non-reducing end

Reducing end



Polysaccharides

Polysaccharide

Homopolysaccharide

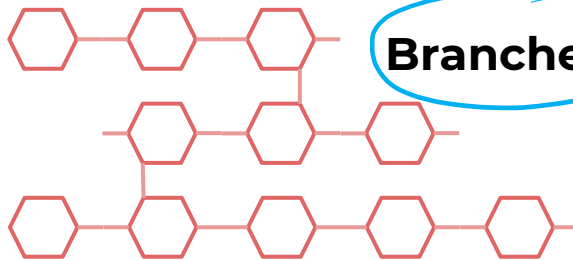
Heteropolysaccharide



Homopolysaccharide



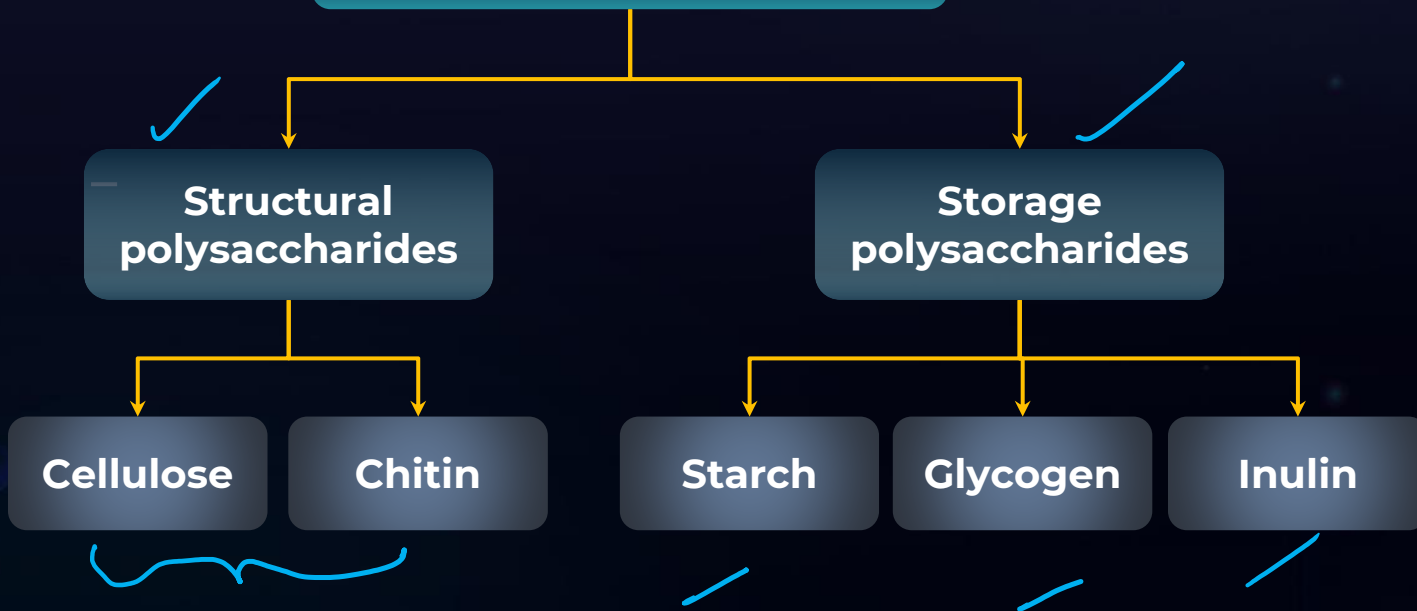
Unbranched



Branched

Homopolysaccharide

Homopolysaccharide

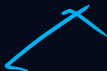


Homopolysaccharide

Homopolysaccharide

**Structural
polysaccharides**

**Storage
polysaccharides**



Structural polysaccharides - Cellulose



Cellulose in
plants

- Most **abundant biomolecule** on the earth
- Forms the **structural part** of the living organisms
- Polymer of **glucose units**
- Straight chain, Unbranched

Structural polysaccharides - Chitin



**Chitin in exoskeleton
of insects**

- **Second most** abundant organic substance
- Polymer of **N-acetyl glucosamine**

Homopolysaccharide

Homopolysaccharide

**Structural
polysaccharides**

**Storage
polysaccharides**



Storage Polysaccharide: Starch



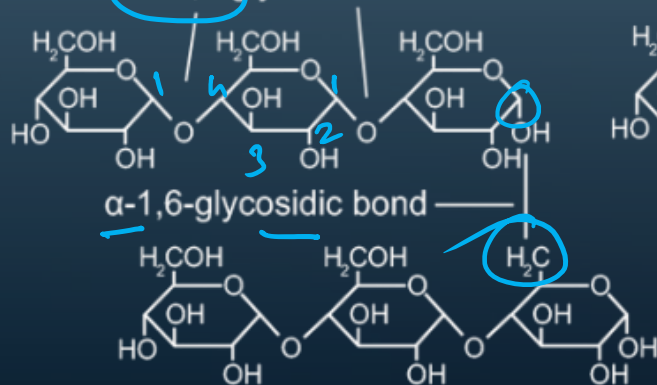
- **Polymer of glucose**
- Food reserve in plants
- Consists of:
 - ✓ **Amylose - 20% - 30%**
 - ✓ **Amylopectin- 70% - 80%**

Storage Polysaccharide: Starch

amylopectin



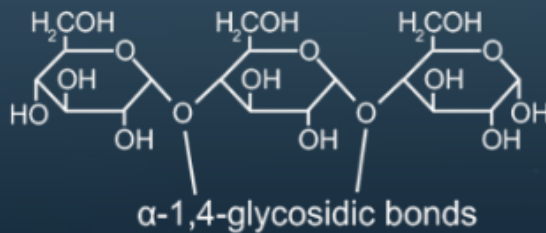
α -1,4-glycosidic bonds



amylose



O = single glucose unit





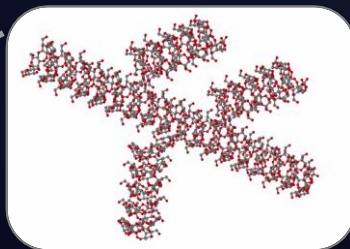
Did you know?



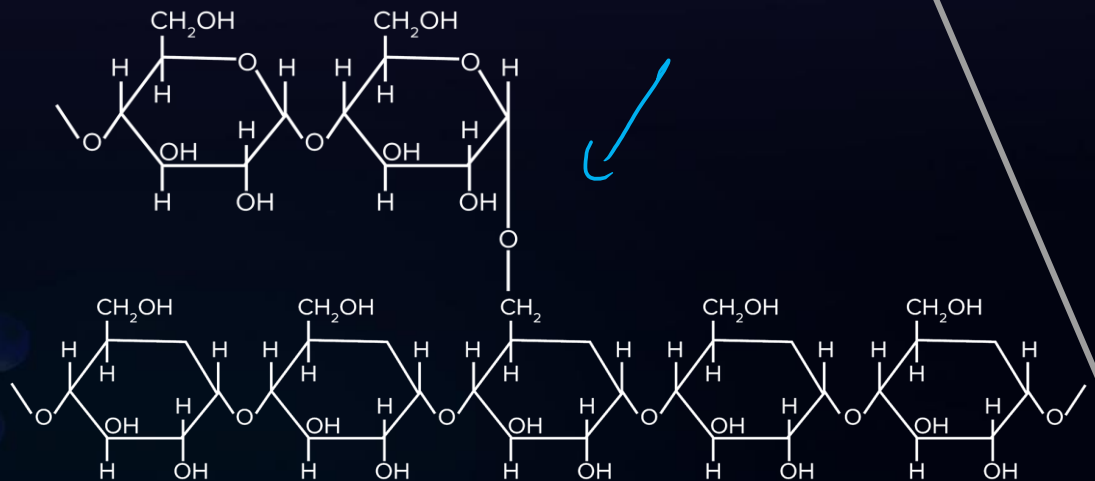
The starch helices in potato can hold iodine and it turns blue-black.



Storage Polysaccharide: Glycogen



- Polymer of **glucose**
- Food reserve in animals
- Stored in **liver and muscles**
- **Highly branched**



Storage Polysaccharide: Inulin



- Polymer of **fructose**
- Stored in roots of *Dahlia*, Dandelion and Artichoke
- **Not metabolised** in human body



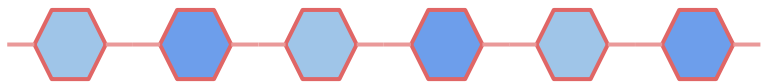
What if a polysaccharide has
different repeating units?



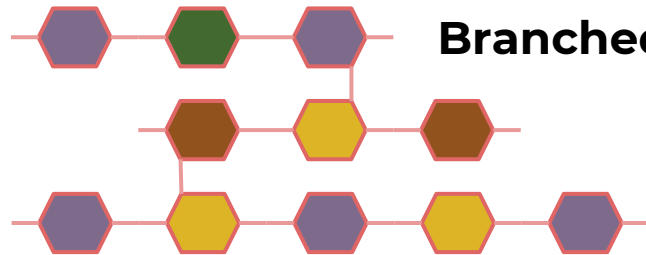


Heteropolysaccharide

Heteropolysaccharide



Unbranched



Heteropolysaccharide

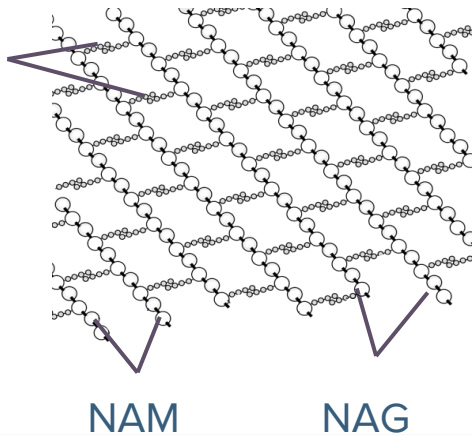
Heteropolysaccharide

Peptidoglycan

Agar

Heteropolysaccharide: Peptidoglycan

Polypeptide



- Component of bacterial cell wall
- Made of two different kind of repeating units

✓ **N- Acetyl glucosamine (NAG)**

✓ **N-Acetyl muramic acid (NAM)**

Heteropolysaccharide: Agar



- Obtained from the red algae: ***Gelidium*** and ***Gracilaria***
- Made of two different units:
 - **Galactose**
 - **3,6-anhydro-L-galactopyranose**



Past Year Questions



Which among the following is not a reducing sugar?(NEET-2014)

A

Maltose

B

Lactose

C

Ribose

D

Sucrose



Which among the following is not a reducing sugar?(AIPMT-2014)

A

Maltose

B

Lactose

C

Ribose

D

Sucrose

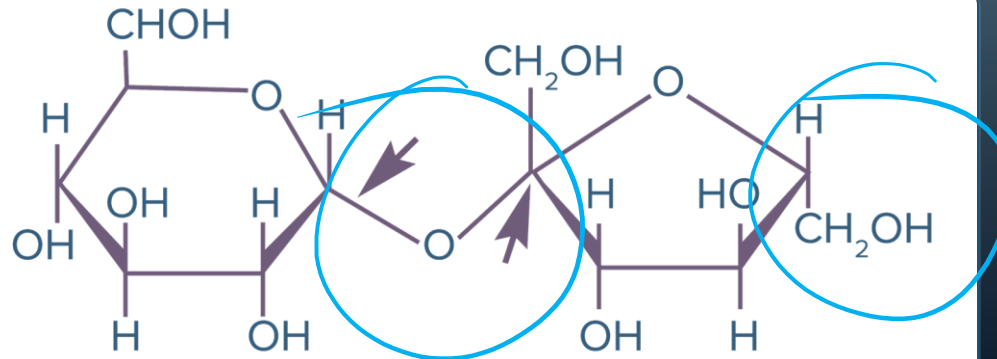




Discussion



- Glucose + Fructose = Sucrose
- Non-Reducing sugar as there is **no free aldehyde or ketone group**





Macromolecule in chitin is (NEET-2013)

A

nitrogen containing
polysaccharide

B

phosphorus containing
polysaccharide

C

Sulphur containing
polysaccharide

D

Simple polysaccharide



Macromolecule in chitin is (NEET-2013)

A

**nitrogen containing
polysaccharide**

B

phosphorus containing
polysaccharide

C

Sulphur containing
polysaccharide

D

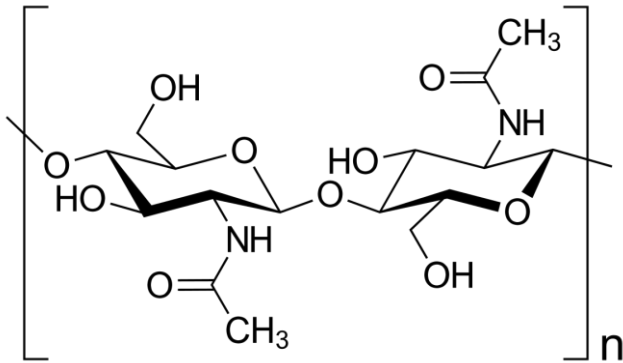
Simple polysaccharide



Discussion



Chitin



- Long-chain polymer of a **N-acetylglucosamine**
- **Nitrogen containing polysaccharide** and a derivative of glucose

A molecular structure graphic in the top-left corner, featuring a central blue sphere with three smaller grey spheres attached to it by white rods.

Summary

A molecular structure graphic in the bottom-right corner, featuring a central blue sphere with two smaller grey spheres attached to it by white rods.

What we have learnt so far

1

Monosaccharides

2

Disaccharides

3

Oligosaccharides

4

Polysaccharides



**Keep
Learning!**