

## CBSE Class 11 Chemistry Chapter 14 Environmental Chemistry Worksheet – Set 4

Q1. Which of the following gases is responsible for ozone layer depletion?  (a) Chloro fluoro carbon (CFCs)  (b) Sulphur dioxide  (c) Nitrogen oxide  (d) All of the above
Q2. Which of the following acids is present in acid rain?  (a) Carbonic acid  (b) Sulphuric acid  (c) Nitric acid  (d) All of the above
Q3. Which of the following conditions show the polluted environment?  (a) The pH of rainwater is 5.6  (b) Biochemical oxygen demand is 10 ppm  (c) Amount of carbon dioxide in the atmosphere is 0.03%  (d) All of the above
Q4. Which of the following gases causes the greenhouse effect?  (a) Methane (b) Sulphur dioxide (c) Both (a) and (b) (d) None of the above
Q5. Water samples with biological oxygen demand values of 4 ppm and 18 ppm, respectively are and  (a) Clean and highly polluted (b) Highly polluted and clean (c) Highly polluted and highly polluted (d) Clean and clean
Q6. Which of the following molecules has a minimum role in the formation of photochemical smog?
Q7. Which of the following is present in the maximum amount in acid rain?
Q8. Green fuel is obtained from
Q9. What is environmental chemistry?



- **Q10.** Compare the toxicity of carbon monoxide and carbon dioxide.
- **Q11.** What are the causes of pneumoconiosis?
- Q12. What is BOD in water systems?
- **Q13.** Why do fish not grow well in warm water?
- **Q14.** What do you understand by green chemistry?
- **Q15.** What is the pH of rainwater?
- **Q16.** How is ozone formed?
- **Q17.** What is sludge?
- **Q18.** What is the importance of the ozone layer in the atmosphere?
- **Q19.** What are the harmful effects of ozone layer depletion?
- Q20. Differentiate between classical smog and photochemical smog.