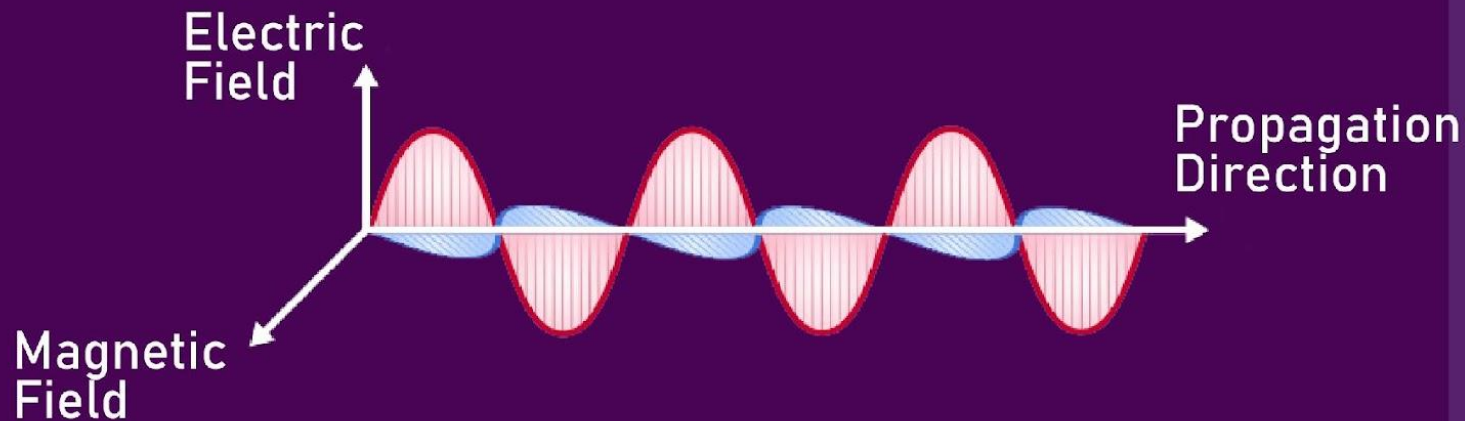




# ***ELECTROMAGNETIC WAVES - L3***



***PHYSICS***

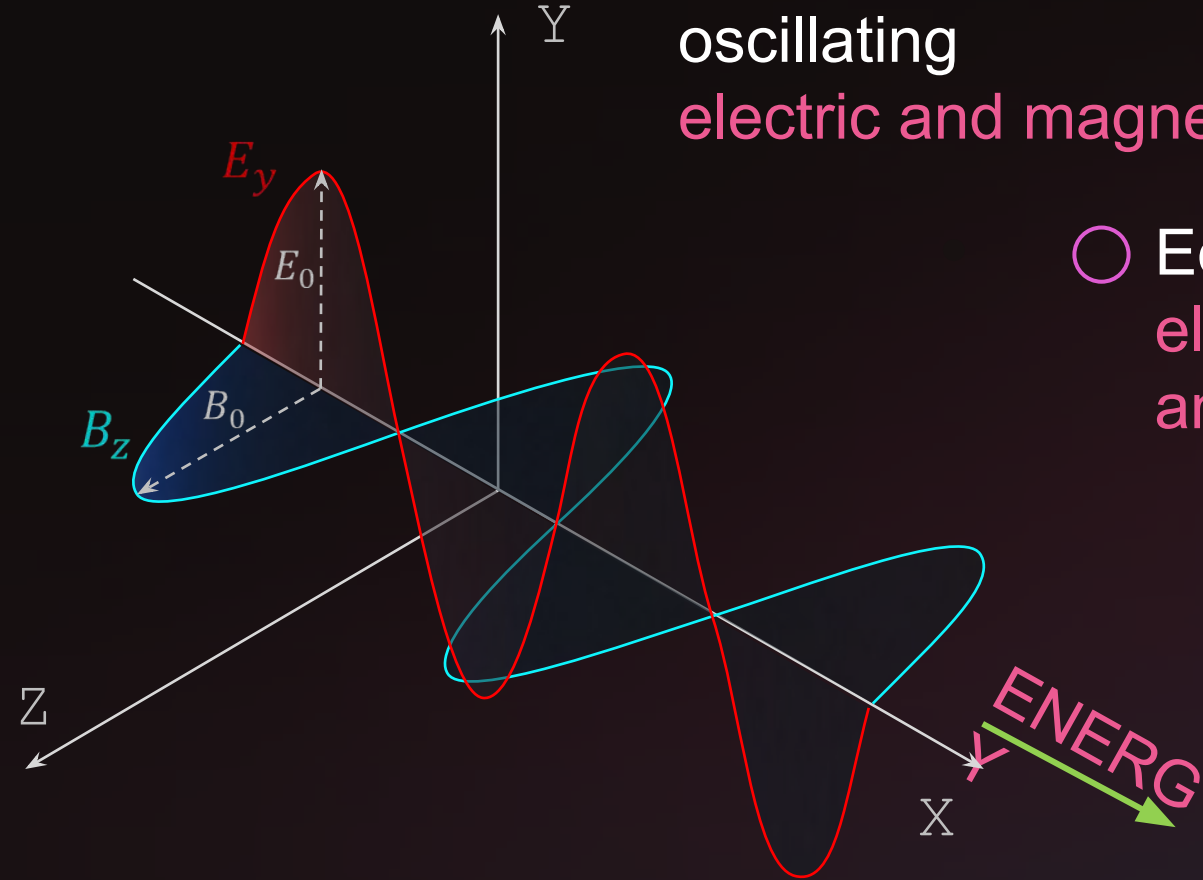


<https://t.me/neetaakashdigital>



- Electromagnetic waves carry energy in oscillating electric and magnetic field.

- Equal amount of energy is contributed by electric and magnetic field.





$$U = \frac{1}{2} C V^2$$

$$V = \vec{E} \cdot \vec{d} = E d \cos 0^\circ = E d$$

$$U = \frac{1}{2} \frac{\epsilon_0 A}{d} (E d)^2$$

$$U = \frac{1}{2} \epsilon_0 E^2 A d$$

$$U = \frac{1}{2} \epsilon_0 E^2 (Vol.)$$

$$\frac{U}{Vol.} = \frac{1}{2} \epsilon_0 E^2$$

$$\mu_E = \frac{1}{2} \epsilon_0 E^2$$

# DENSITY OF ELECTRIC FIELD

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$$\mu_E = \frac{1}{2} \epsilon_0 E^2$$

$$E = E_0 \sin(Kx - \omega t)$$

# DENSITY OF ELECTRIC FIELD



$$\mu_E = \frac{1}{2} \epsilon_0 E^2 \qquad E = E_0 \sin(Kx - \omega t)$$

$$\mu_E = \frac{1}{2} \epsilon_0 E_0^2 \sin^2(Kx - \omega t)$$

$$\overline{\mu_E} = \frac{1}{2} \epsilon_0 E_0^2 \overline{\sin^2(Kx - \omega t)}$$

$$\overline{\mu_E} = \frac{1}{2} \epsilon_0 E_0^2 \times \frac{1}{2}$$

$$\sim (\mu_E)_{avg} = \frac{1}{4} \epsilon_0 E_0^2 \sim$$





$$U = \frac{1}{2} L i^2$$

$$L = \mu_0 n^2 A l \quad \text{and} \quad i = \frac{B}{\mu_0 n}$$

$n$  = no. of turns per unit length

$$B = \mu_0 n i$$

$$U = \frac{1}{2} \mu_0 n^2 A l \times \frac{B^2}{\mu_0^2 n^2}$$

$$U = \frac{1}{2} \frac{B^2}{\mu_0} A l$$

$$\frac{U}{Vol.} = \frac{1}{2} \frac{B^2}{\mu_0}$$

$$\mu_B = \frac{1}{2} \frac{B^2}{\mu_0}$$

# DENSITY OF MAGNETIC FIELD

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$$\mu_B = \frac{1}{2} \frac{B^2}{\mu_0}$$

$$B = B_0 \sin(Kx - \omega t)$$

# DENSITY OF MAGNETIC FIELD



$$\mu_B = \frac{1}{2} \frac{B^2}{\mu_0}$$

$$B = B_0 \sin(Kx - \omega t)$$

$$\mu_B = \frac{1}{2} \frac{B_0^2}{\mu_0} \sin^2(Kx - \omega t)$$

$$\overline{\mu_B} = \frac{1}{2} \frac{B_0^2}{\mu_0} \overline{\sin^2(Kx - \omega t)}$$

$$\overline{\mu_B} = \frac{1}{2} \frac{B_0^2}{\mu_0} \times \frac{1}{2}$$

$$\sim (\mu_B)_{avg} = \frac{1}{4} \frac{B_0^2}{\mu_0} \sim$$



## Question



Which of the following is true

**a**

$$(\mu_E)_{Avg} = (\mu_B)_{Avg}$$

**b**

$$(\mu_E)_{Avg} = 2(\mu_B)_{Avg}$$

**c**

$$2(\mu_E)_{Avg} = (\mu_B)_{Avg}$$

**d**

$$(\mu_E)_{Avg} = 4(\mu_B)_{Avg}$$



# DISCUSSION



$$(\mu_E)_{Avg} = \frac{1}{4} \epsilon_0 E_0^2$$

$$(\mu_B)_{Avg} = \frac{1}{4\mu_0} B_0^2$$



Y



$$(\mu_E)_{Avg} = \frac{1}{4} \epsilon_0 E_0^2$$

$$(\mu_B)_{Avg} = \frac{1}{4\mu_0} B_0^2$$

$$(\mu_E)_{Avg} = \frac{1}{4} \epsilon_0 (B_0 c)^2$$

$$\frac{E_0}{B_0} = c \quad c = \frac{1}{\sqrt{\mu_0 \epsilon_0}}$$

$$(\mu_E)_{Avg} = \frac{1}{4} \epsilon_0 B_0^2 \frac{1}{\mu_0 \epsilon_0}$$

$$(\mu_E)_{Avg} = \frac{1}{4} B_0^2 \frac{1}{\mu_0}$$

$$(\mu_E)_{Avg} = \frac{1}{4\mu_0} B_0^2 = (\mu_B)_{Avg}$$

$$(\mu_E)_{Avg} = (\mu_B)_{Avg}$$



# Answer



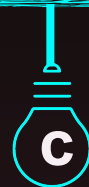
Which of the following is true



$$(\mu_E)_{Avg} = (\mu_B)_{Avg}$$



$$(\mu_E)_{Avg} = 2(\mu_B)_{Avg}$$



$$2(\mu_E)_{Avg} = (\mu_B)_{Avg}$$



$$(\mu_E)_{Avg} = 4(\mu_B)_{Avg}$$



$$\mu = \mu_E + \mu_B$$

$$\mu = \mu_E + \mu_E = 2\mu_E$$

$\Downarrow$

$$\mu = 2 \frac{1}{4} \epsilon_0 E_0^2$$

$\Downarrow$

$$\mu = \frac{1}{2} \epsilon_0 E_0^2$$

$$\mu = \mu_B + \mu_B = 2\mu_B$$

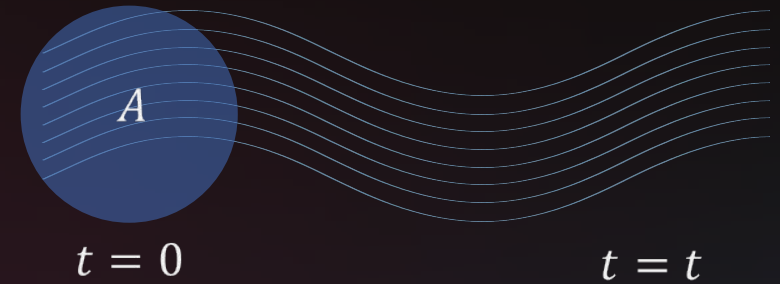
$\Downarrow$

$$\mu = 2 \frac{1}{4\mu_0} B_0^2$$

$\Downarrow$

$$\mu = \frac{1}{2\mu_0} B_0^2$$

- The energy crossing per unit time in a direction perpendicular to the direction of propagation is called intensity of the wave ( $I$ ).



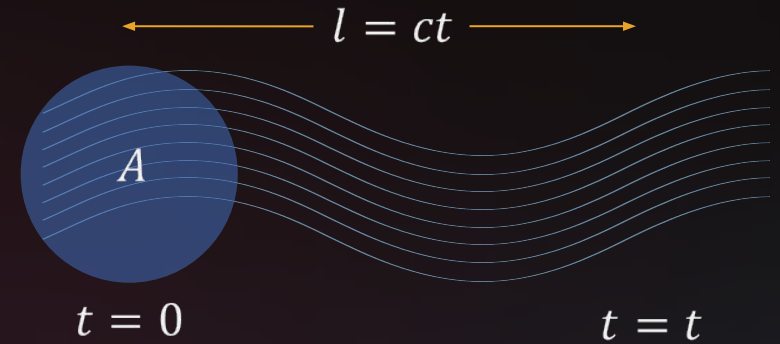
# waves



- The energy crossing per unit time in a direction perpendicular to the direction of propagation is called intensity of the wave ( $I$ ).

$$I = \frac{U}{At}$$

Energy ( $U$ ) in unit volume of cylinder  $\mu = \frac{1}{2} \epsilon_0 E_0^2$



$$U = \mu \times \text{volume}$$

$$U = \frac{1}{2} \epsilon_0 E_0^2 \times A \times ct$$

$$I = \frac{\frac{1}{2} \epsilon_0 E_0^2 \times A \times ct}{At}$$

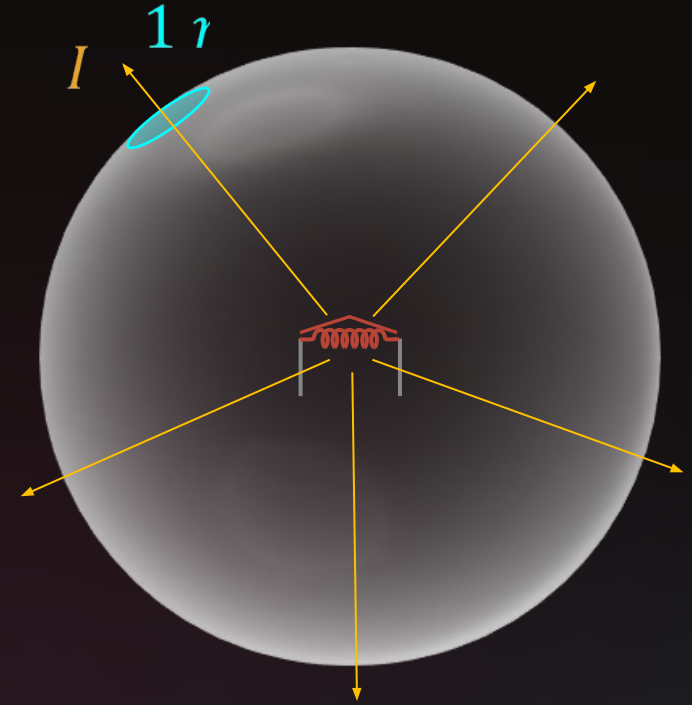
$$I = \frac{1}{2} \epsilon_0 E_0^2 c$$

# Relation between power and intensity

$$I = \frac{U}{At}$$

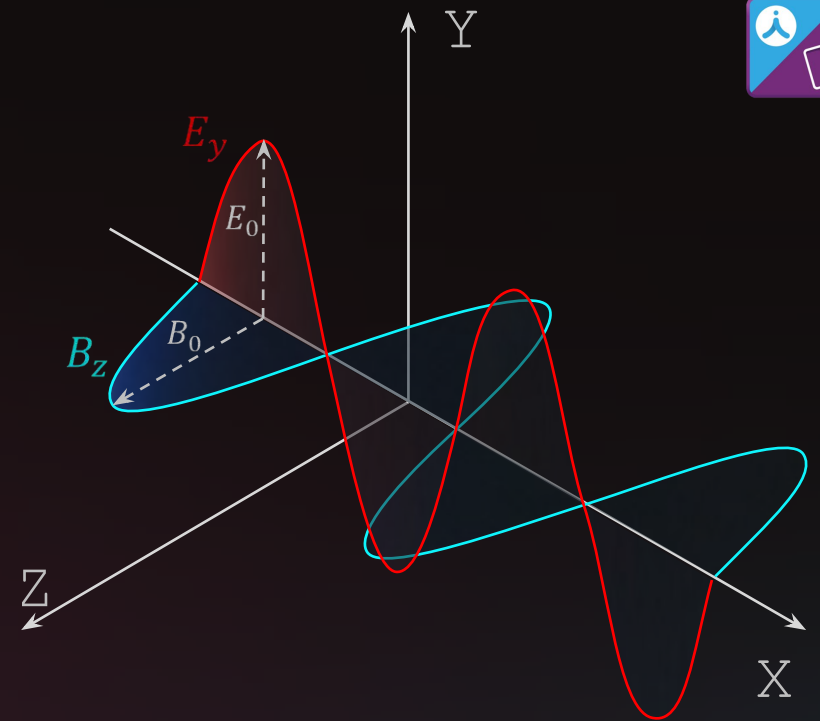
$$\frac{U}{t} = \text{Power } (P)$$

$$I = \frac{P}{A}$$



# WAVE

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# WAVE

○ De-Broglie hypothesis  $\lambda = \frac{h}{\bar{p}}$

$$P = \frac{h}{\lambda}$$

Moment  
um

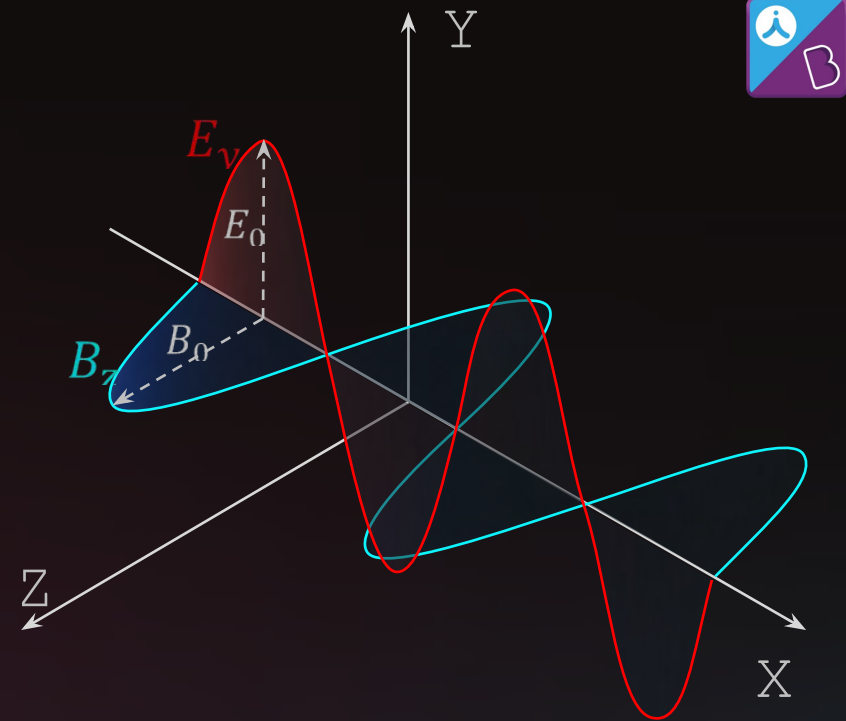
$$U = hf$$

$$c = f\lambda$$

$$U = \frac{hc}{\lambda}$$

$$U = Pc$$

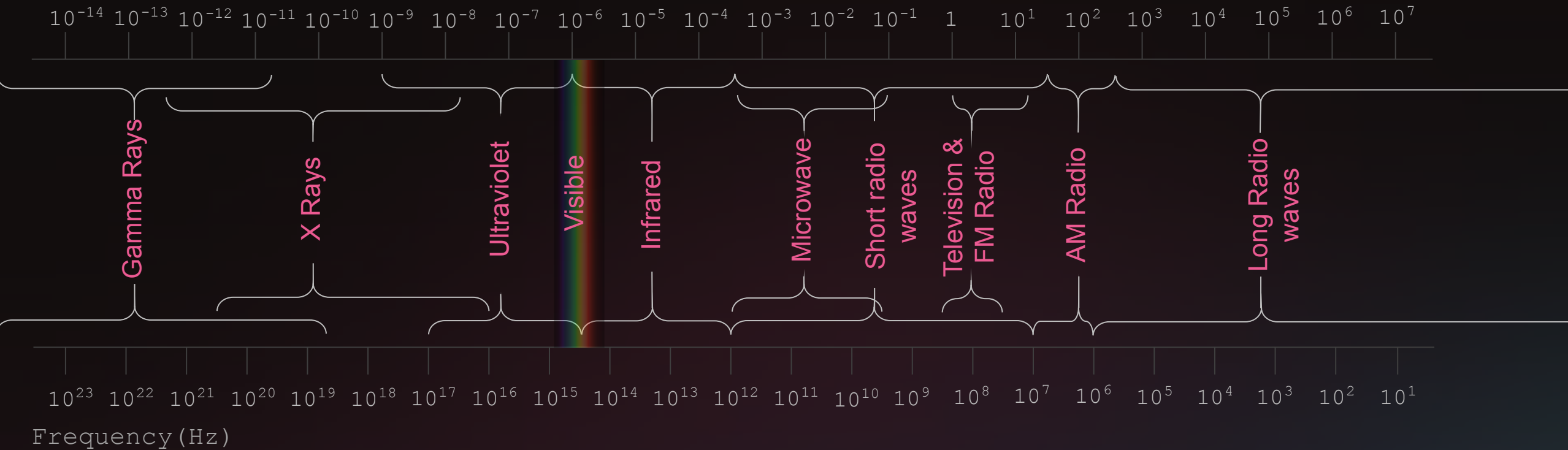
$$P = \frac{U}{c}$$



# ELECTROMAGNETIC SPECTRUM



Wavelength(m)



- The classification of EM waves according to its frequency/wavelength is called **Electromagnetic Spectrum**.

# ELECTROMAGNETIC SPECTRUM

Frequency:  $10^{19} \text{ Hz} - 10^{24} \text{ Hz}$

Wavelength:  $< 10^{-11} \text{ m}$



## GAMMA RAYS

They are produced by the disintegration of radioactive atomic nuclei.



They are used in medicine to destroy cancer cells.



# ELECTROMAGNETIC SPECTRUM

Frequency:  $10^{16} \text{ Hz} - 10^{20} \text{ Hz}$

Wavelength:  $< 10^{-8} - 10^{-12} \text{ m}$



## X-RAYS

X-rays are produced most commonly when fast moving electrons decelerates inside a metal target.



————• They are harmful to living tissues.

They are widely used in medical

# ELECTROMAGNETIC SPECTRUM

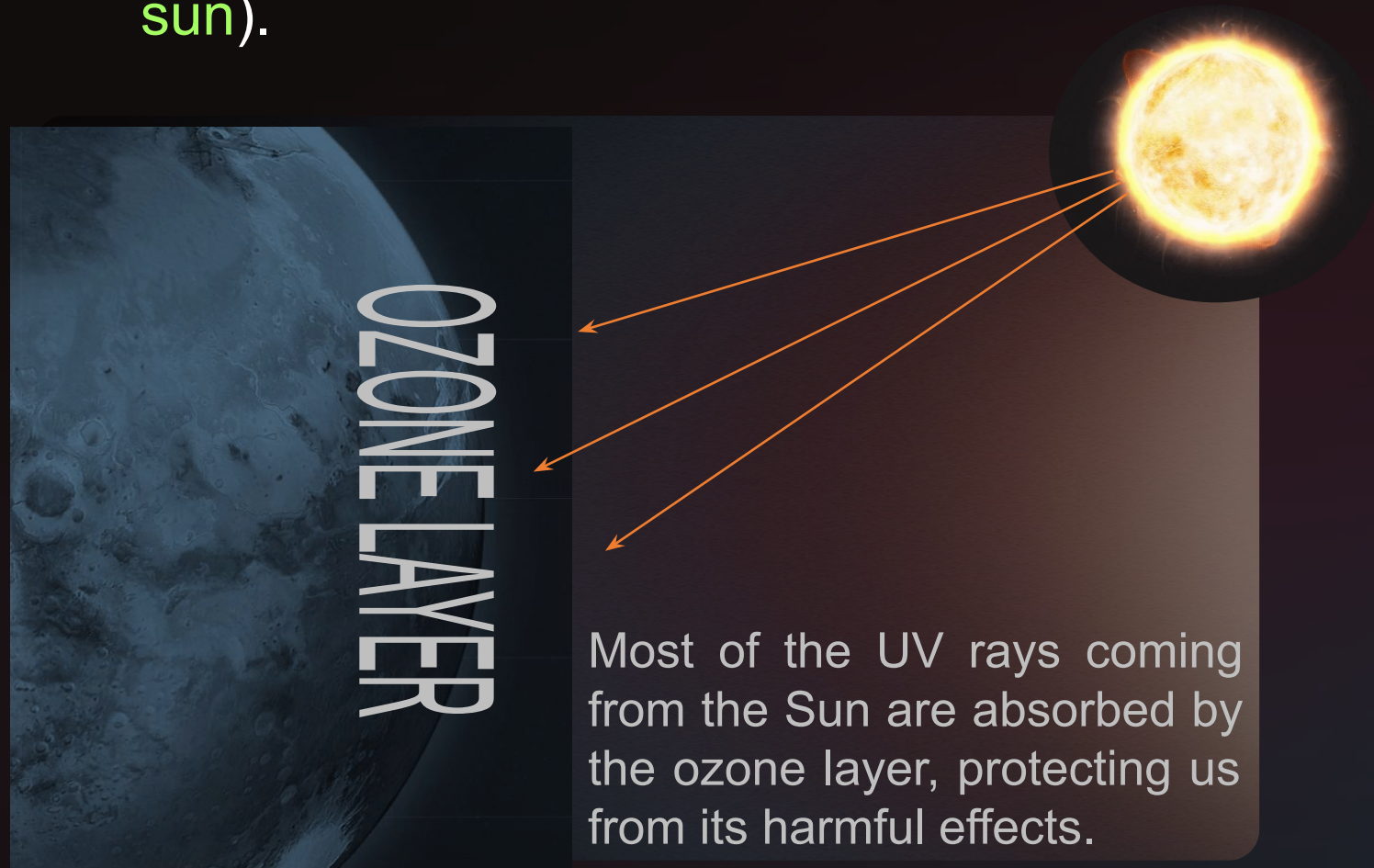
Frequency:  $10^{15} \text{ Hz} - 10^{17} \text{ Hz}$

Wavelength:  $< 400 \text{ nm} - 1 \text{ nm}$



## ULTRAVIOLET RAYS

UV-rays are produced by special lamps and very hot bodies (eg sun).



# ELECTROMAGNETIC SPECTRUM



## ULTRAVIOLET RAYS

UV radiation is absorbed by ordinary glass.



Welders wear special glass shields to protect their eyes from large amount of UV produced by welding arcs.

They are used in eye surgery and for water purification.

# ELECTROMAGNETIC SPECTRUM

Frequency:  $4 \times 10^{14} \text{ Hz} - 7 \times 10^{14} \text{ Hz}$

Wavelength:  $< 700 \text{ nm} - 400 \text{ nm}$



## VISIBLE RAYS

It is the part of the spectrum that is detected by the human eye.

V	I	B	G	Y	O	R
VIOLET	INDIGO	BLUE	GREEN	YELLOW	ORANGE	RED
400 nm	440 nm	470 nm	515 nm	560 nm	620 nm	700 nm



# ELECTROMAGNETIC SPECTRUM

Frequency:  $10^{12} \text{ Hz} - 10^{14} \text{ Hz}$

Wavelength:  $< 780 \text{ nm} - 1000 \text{ }\mu\text{m}$



## INFRARED RAYS

They are produced by hot bodies and molecules. They are also known as **HEAT WAVES**.



They are used in the remote switches of household electronic system.

THERMAL IMAGING using Infrared radiation.

# ELECTROMAGNETIC SPECTRUM

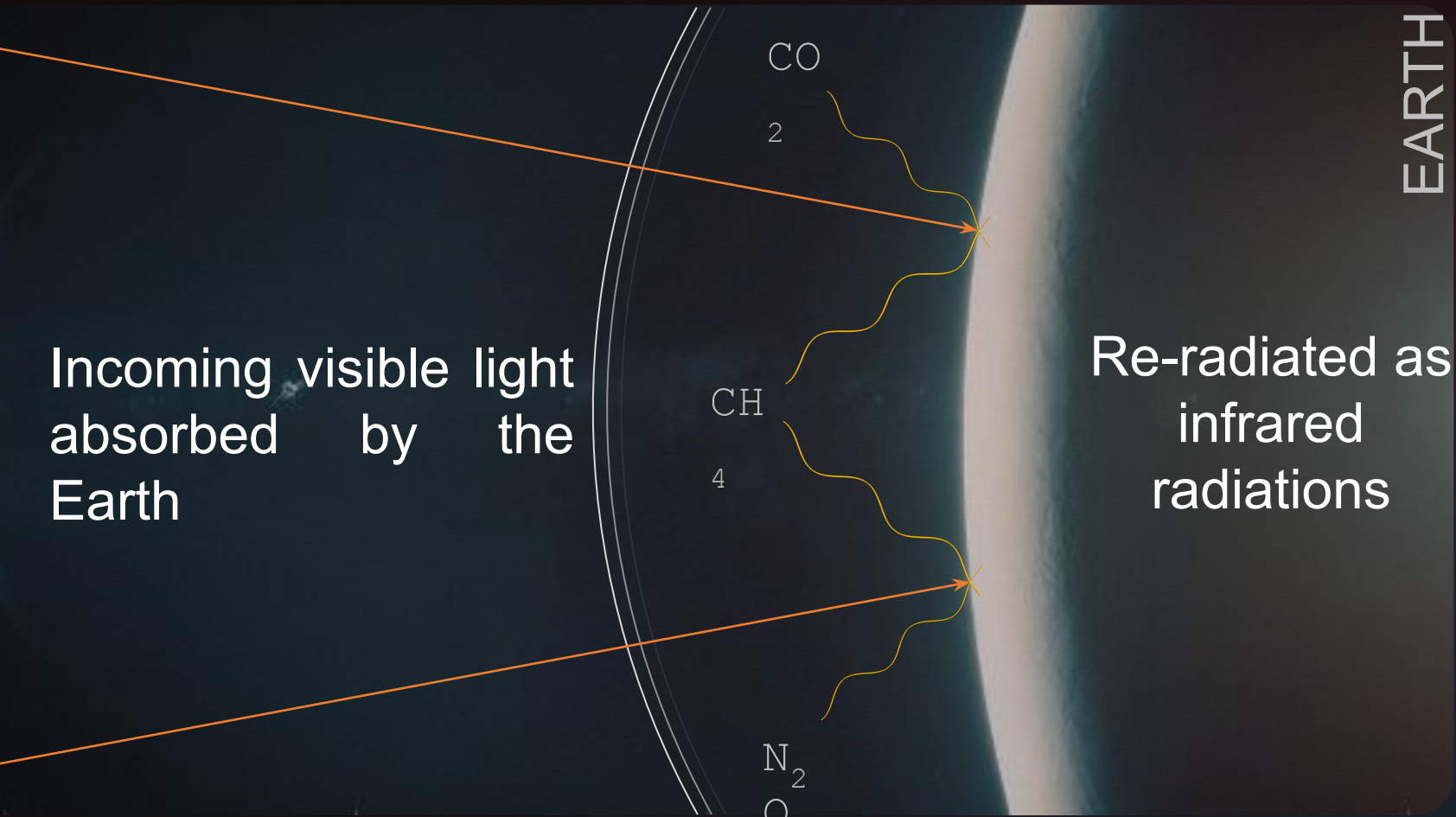
Frequency:  $10^{12} \text{ Hz} - 10^{14} \text{ Hz}$

Wavelength:  $< 780 \text{ nm} - 1000 \text{ }\mu\text{m}$



## INFRARED RAYS

It plays an important role in maintaining the earth's warmth through the **GREENHOUSE EFFECT**.



# ELECTROMAGNETIC SPECTRUM

Frequency:  $3 \times 10^{10} \text{ Hz} - 10^{12} \text{ Hz}$

Wavelength:  $< 1 \text{ mm} - 25 \text{ mm}$



## Microwaves

They are produced by special vacuum tubes.

Due to their shorter wavelengths, they are suitable for the radar system used in aircraft navigation.



Microwave oven is an interesting domestic application of these waves.

# ELECTROMAGNETIC SPECTRUM

Frequency:  $< 3 \times 10^{11} \text{ Hz}$   
Wavelength:  $> 1 \text{ mm}$



## RADIO WAVE



- They produced by the accelerated motion of charges are in conducting wires
- They are used in radio and television communication system.

Amplitude  
Modulated (AM)  
radio wave  
(530 kHz-1710  
kHz)

Television radio  
wave  
(54 MHz-890  
MHz)

Frequency  
Modulated (FM)  
radio wave  
(88 MHz-108  
MHz)

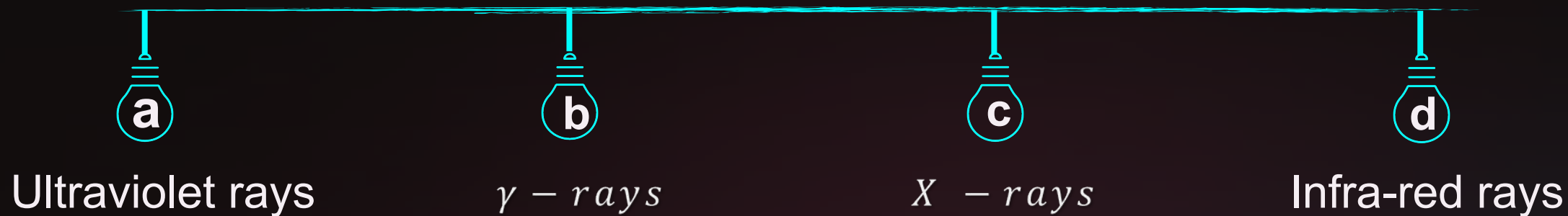




## Question



The energy of the E.M.Waves is of the order of  $15 \text{ keV}$ . To which part of the spectrum does it belong?





# DISCUSSION



*Given:  $E = 15 \text{ keV}$*

$$hc = 1240 \text{ eV nm}$$



*Given:  $E = 15 \text{ keV}$*

$$hc = 1240 \text{ eV nm}$$

$$\lambda = \frac{hc}{E}$$

$$\lambda = \frac{1240}{15 \times 10^3}$$

$$\lambda = 0.083 \text{ nm}$$

*This wavelength belongs to X-rays*

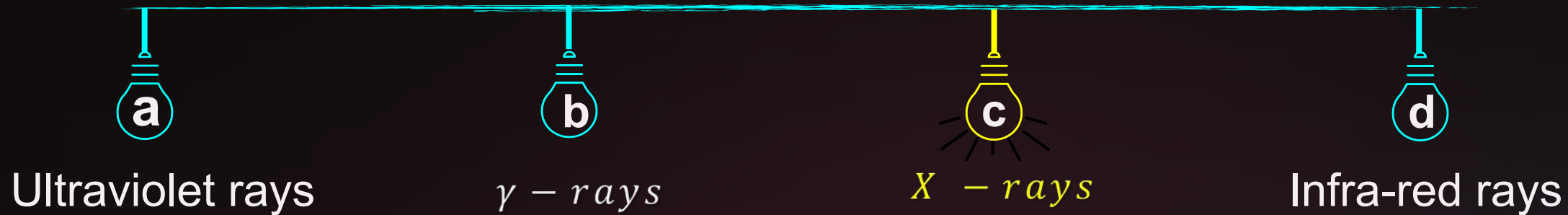


# Answer



The energy of the E.M.Waves is of the order of  $15 \text{ keV}$ . To which part of the spectrum does it belong?

NEET 2015





## Question



If  $\lambda_v$ ,  $\lambda_x$  and  $\lambda_m$  represent the wavelengths of visible light,  $X$  - rays and microwaves respectively, then



$$\lambda_m > \lambda_x > \lambda_v$$



$$\lambda_m > \lambda_v > \lambda_x$$



$$\lambda_v > \lambda_x > \lambda_m$$



$$\lambda_v > \lambda_m > \lambda_x$$



# ON



Wavelength(m)

$10^{-14}$   $10^{-13}$   $10^{-12}$   $10^{-11}$   $10^{-10}$   $10^{-9}$   $10^{-8}$   $10^{-7}$   $10^{-6}$   $10^{-5}$   $10^{-4}$   $10^{-3}$   $10^{-2}$   $10^{-1}$  1  $10^1$   $10^2$   $10^3$   $10^4$   $10^5$   $10^6$   $10^7$

Gamma Rays

X Rays

Ultraviolet

Visible

Infrared

Microwave

Short radio  
waves

Television &  
FM Radio

AM Radio

Long Radio  
waves

$10^{23}$   $10^{22}$   $10^{21}$   $10^{20}$   $10^{19}$   $10^{18}$   $10^{17}$   $10^{16}$   $10^{15}$   $10^{14}$   $10^{13}$   $10^{12}$   $10^{11}$   $10^{10}$   $10^9$   $10^8$   $10^7$   $10^6$   $10^5$   $10^4$   $10^3$   $10^2$   $10^1$

Frequency (Hz)



# Answer



If  $\lambda_v$ ,  $\lambda_x$  and  $\lambda_m$  represent the wavelengths of visible light,  $X$  - rays and microwaves respectively, then

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$$\lambda_m > \lambda_x > \lambda_v$$



$$\lambda_m > \lambda_v > \lambda_x$$



$$\lambda_v > \lambda_x > \lambda_m$$



$$\lambda_v > \lambda_m > \lambda_x$$



# Question



Ozone layer blocks the radiation of wavelength

*NEET 2005*



More than  
 $3 \times 10^{-7} \text{ m}$



Equal to  
 $3 \times 10^{-7} \text{ m}$



Less than  
 $3 \times 10^{-7} \text{ m}$



All of these





# Answer



Ozone layer blocks the radiation of wavelength

NEET 2005



More than  
 $3 \times 10^{-7} \text{ m}$



Equal to  
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Less than  
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All of these



# Question



Ozone layer blocks the radiation of wavelength

*NEET 1999*



More than  
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All of these



# Answer



Ozone layer blocks the radiation of wavelength

*NEET 1999*



More than  
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Equal to  
 $3 \times 10^{-7} \text{ m}$



Less than  
 $3 \times 10^{-7} \text{ m}$



**All of these**